

Sources and Properties of Amazonian Aerosol Particles

Scot T. Martin,^{1,*} Meinrat O. Andreae,² Paulo Artaxo,³ Darrel Baumgardner,⁴ Qi Chen,¹ Allen H. Goldstein,⁵ Alex Guenther,⁶ Colette L. Heald,⁷ Olga L. Mayol-Bracero,⁸ Peter H. McMurry,⁹ Theotonio Pauliquevis,¹⁰ Ulrich Pöschl,² Kimberly A. Prather,¹¹ Gregory C. Roberts,¹² Scott R. Saleska,¹³ M.A. Silva Dias,¹⁴ Dominick V. Spracklen,¹⁵ Erik Swietlicki,¹⁶ and Ivonne Trebs²

1. School of Engineering and Applied Sciences & Department of Earth and Planetary Sciences, Harvard University, USA
2. Biogeochemistry Department, Max Planck Institute for Chemistry, Mainz, Germany
3. Institute of Physics, University of São Paulo, Brazil
4. Centro de Ciencias de la Atmosfera, Universidad Nacional Autonoma de Mexico, Ciudad Universitaria, Mexico
5. Department of Environmental Science, Policy, & Management, University of California at Berkeley, USA
6. Earth and Sun Systems Laboratory, National Center for Atmospheric Research, USA
7. Department of Atmospheric Science, Colorado State University, USA
8. Institute for Tropical Ecosystem Studies, University of Puerto Rico, USA
9. Department of Mechanical Engineering, University of Minnesota, USA.
10. Climate and Environmental Modelling Group, Instituto Nacional de Pesquisas da Amazônia, Brazil
11. Department of Chemistry and Biochemistry, Scripps Institution of Oceanography, University of California at San Diego, USA
12. Center for Atmospheric Sciences, Scripps Institution of Oceanography, USA
13. Department of Ecology & Evolutionary Biology, University of Arizona, USA
14. Center for Weather Forecasting and Climate Studies, National Institute for Space Research, and University of São Paulo, Brazil
15. Institute for Climate and Atmospheric Science, School of Earth and Environment, University of Leeds, United Kingdom
16. Department of Physics, Lund University, Lund, Sweden

*To Whom Correspondence Should be Addressed (scot_martin@harvard.edu)

Submitted December 2008 to Reviews of Geophysics

Revised August 2009

Table of Contents

1. Introduction	2
2. Sources	8
2.1. Primary Particles	9
2.1.1. <i>Primary Biological Particles</i>	9
2.1.2. <i>Biomass Burning</i>	11
2.1.3. <i>African Mineral Dust</i>	12
2.1.4. <i>Marine Emissions</i>	13
2.2. Secondary Gas-to-Particle Conversion	14
2.2.1. <i>Biogenic Volatile Organic Compounds</i>	15
2.2.2. <i>New Particle Formation</i>	21
3. Properties	24
3.1. Mass Concentration	24
3.2. Number-Diameter Distribution	27
3.3. Chemical Composition	31
3.3.1. <i>Organic Component</i>	31
3.3.2. <i>Inorganic Component</i>	35
3.4. Hygroscopicity	38
3.5. Cloud Condensation Nuclei	41
4. Outlook and Future Priorities	44
4.1. Priorities for Improved Models	46
4.2. Priorities for Improved Measurements	47

1 **1. Introduction**

2 Aerosol particles in the Amazon Basin have been the focus of numerous field campaigns
3 over the past 20 years (Table 1 and Figure 1). These studies were motivated by a wide range of
4 objectives, the most prominent of which are as follows. (1) The Basin was used as a laboratory to
5 gain baseline knowledge concerning pristine continental aerosol particles, against which the
6 effects of human activities globally could be judged [Andreae, 2007]. (2) An understanding was
7 sought of the effects of biomass-burning aerosol particles on human health, such as increased
8 incidences of morbidity, mortality, and asthma [Ignotti et al., 2007]. (3) The effects of aerosol
9 particles on regional climate were investigated, such as changes in rainfall patterns as a
10 consequence of the redistribution of energy and cloud condensation nuclei [Andreae et al.,
11 2004]. (4) The Basin was studied as an integrated ecosystem to understand the feedback and
12 regulation of plant emissions on rainfall and, in turn, of rainfall on plant growth and emissions
13 [Barth et al., 2005; Keller et al., 2009]. These topics have in common a need to know the sources
14 and properties of Amazonian aerosol particles, yet an integrated summary of results from
15 previous field campaigns (Table 1) and associated statements of future research priorities have
16 not been prepared previously in a comprehensive review article. This gap in the literature is the
17 motivation for this review and defines its scope: the review's goals are to focus the ongoing
18 activities of researchers already investigating the sources and properties of Amazonian aerosol
19 particles and, by organizing and presenting material of what is already known and what remains
20 to be learned, to invite new researchers to join in critical ways. Complementary general reviews
21 of atmospheric particles, especially with regard to the organic component that is dominant in the
22 Amazon Basin, can be found in Andreae and Crutzen [1997], Jacobson et al. [2000], Kanakidou
23 et al. [2005], and Fuzzi et al. [2006].

24 Aerosol sources located within the Amazon Basin are dominated, with the exception of
25 some urbanized areas and transportation corridors, by natural and anthropogenic emissions from
26 the biosphere. Sources include both high but intermittent biomass-burning emissions (both
27 natural and anthropogenic) and low but more consistent production of primary and secondary

28 biological aerosol particles and components (Figure 2). Primary particles are produced both
29 deliberately by flora (e.g., the release of pollen and fungal spores) and incidentally (e.g., as leaf
30 and soil debris or as suspended microbes). Substantial production of secondary aerosol occurs by
31 the atmospheric oxidation of trace gases to low-volatility compounds. These products can
32 deposit on pre-existing particles or possibly nucleate new particles.

33 Once in the atmosphere, particles undergo continuous transformations (Figure 3).
34 Processes include (photo)chemical reactions that occur between compounds within the particles
35 as well as interactions that occur between compounds within the particles and those in the gas
36 phase, such as the condensation of low-volatility compounds or reactions with highly reactive
37 gaseous species like the OH radical. Clouds are present at varying abundances almost
38 everywhere and all the time over the Amazon Basin, and most particles are also undergo several
39 cycles of cloud processing during their residence in the Basin. The timescale of cloud cycling of
40 boundary layer air (of the order of hours) is considerably shorter than the residence time of air
41 over the Basin or the deposition lifetime of aerosol particles (of the order of days). Cloud
42 processing can modify particle properties both by chemical reactions in the liquid phase and by
43 interactions between droplets (e.g., collision and coagulation). Particles leave the Amazonian
44 atmosphere by dry deposition to the vegetation surface, by cloud scavenging and precipitation,
45 and by advection out of the region.

46 In addition to particle sources within the Basin, there are also important and at times
47 dominant long-range natural and anthropogenic sources (Figure 4). The influence of long-range
48 transport is particularly important when in-Basin sources are weak, such as in the wet season,
49 and under these conditions the particle population can be dominated at times both in mass and
50 number by outflow from other areas. The Atlantic Ocean, upwind of the Basin, is a strong source
51 of marine particles that are generated both directly by seaspray as well as indirectly by the
52 conversion of gases, such as the oxidation of dimethyl sulfide (DMS) to form sulfate. Across the
53 Atlantic and further upwind of the Basin, the Saharan desert is the world's largest source of
54 mineral dust. Sub-Saharan Africa is one of the most important sources of smoke from vegetation

55 fires. Although the inflowing air masses that arrive with the trade winds from the Atlantic shed
56 much of their particle burden in transit from Africa and Eurasia, the impact of trans-Atlantic
57 transport on particles measured in the Basin can, nevertheless, be substantial and at times
58 dominant [*Prospero et al.* 1981; *Andreae et al.*, 1990a; *Artaxo et al.*, 1990; *Swap et al.*, 1992;
59 *Formenti et al.*, 2001; *Chen et al.*, 2009]. Furthermore, pollution-derived particles from urban
60 and industrialized areas in southern and eastern Brazil and other South American countries can
61 also be transported into the Basin, especially in the dry season.

62 During the wet season (Dec-Mar, i.e., summer of the Southern Hemisphere), atmospheric
63 particles are removed relatively fast by wet deposition, and anthropogenic sources such as
64 biomass burning are weak throughout the Basin. This combination of circumstances results in
65 natural processes (including contributions from marine and African sources) as the dominant
66 contributors to the ambient particle populations over large expanses of the Basin and during a
67 significant part of the year. The particle concentrations measured during these conditions are
68 among the lowest found on any continent and are similar to those over the remote oceans
69 [*Andreae*, 2009]. The Basin has been dubbed the “green ocean” because of the similarities in
70 particle concentrations and cloud microphysics between it and remote oceanic regions [*Williams*
71 *et al.*, 2002]. The Amazon Basin may be the only region on the tropical continents where there
72 remains the possibility to find at times populations of nearly pristine aerosol particles free of
73 direct anthropogenic influences.

74 Compared to these green-ocean conditions of the wet season, there is stark contrast in the
75 dry season for large regions of the Basin (Jun-Sep, i.e., winter of the Southern Hemisphere). Vast
76 numbers of deforestation fires burn during the dry season, especially along the peripheries of the
77 forest, and large parts of the Basin become among the most polluted places on Earth [*Artaxo et*
78 *al.*, 2002; *Cardoso et al.*, 2003]. As one effect, the regional energy balance is changed because
79 the high particle concentrations affect the amount and location of solar radiation absorbed by the
80 planet. Simulations using regional climate models show that the changes in energy delivery
81 significantly influence regional patterns of atmospheric circulation and meteorology [*Zhang et*

82 *al.*, 2009]. The high particle concentrations change cloud microphysics and rainfall, with a
83 significant influence on the overall water cycle [*Andreae et al.*, 2004; *IPCC*, 2007; *Rosenfeld et*
84 *al.*, 2008]. They also influence air quality by degrading visibility and affecting human health
85 [*Reinhardt et al.*, 2001; *Schwartz et al.*, 2002; *Watson*, 2002; *Pope and Dockery*, 2006;
86 *Barregard et al.*, 2006].

87 The meteorology of South America guides the transport of particle-free or particle-laden
88 air masses into the Amazon Basin. It also affects the rates of wet and dry particle deposition. A
89 brief introduction to the meteorology of South America is therefore provided herein, in the next
90 few paragraphs. A more in-depth presentation can be found in *Satyamurty et al.* [1998], and an
91 introduction to the regional climate of the Amazon Basin (including a history of field campaigns
92 focused on meteorology) can be found in *Nobre et al.* [2004].

93 The convective activity and the atmospheric circulation of tropical South America are
94 part of a monsoon system [*Zhou et al.*, 1998]. The seasonal cycle of circulation and convection is
95 revealed both by changes in the low-level wind field (cf. Figures 5a and 5b) and in the outgoing
96 longwave radiation (cf. Figures 5c and 5d) for December through February (DJF) compared to
97 June through August (JJA) [*Kalnay et al.*, 1996]. In both the wet and dry seasons, there is
98 considerable inflow coming from the east into the Basin. These air masses originate in the
99 Atlantic, and many of them pass through the semi-arid region of northeastern Brazil before
100 entering central and southern regions of the Basin [*Satyamurty et al.*, 1998].

101 Several aspects of Figure 5 are important for understanding the variability of particle
102 concentrations and compositions observed in the Amazon Basin. For northern and central
103 Amazonia, the Intertropical Convergence Zone (ITCZ), which is the confluence between the
104 northeastern and southeastern trade winds extending from West Africa to South America
105 (Figures 5a and 5b), has an important influence. The ITCZ reaches the northern coast of South
106 America just south of the equator in DJF but just north of it in JJA. When the ITCZ is located to
107 the south of the equator in DJF (i.e., a large part of the Basin is under the influence of air from
108 the Northern Hemisphere), low-level winds from the global northeast reach the northern coast of

109 South America and open up the possibility of advection of African dust and biomass-burning
110 particles (e.g., *Artaxo and Hansson* [1995] and *Formenti et al.* [2001]). During the wet season,
111 Figure 5c shows that the outgoing longwave radiation has a minimum over central Amazonia,
112 implying cold cloud tops and hence deep clouds with strong convection and high rates of wet
113 deposition. There is also strong vertical transport and redistribution of particles [*Freitas et al.*,
114 2000; *Andreae et al.*, 2001]. In comparison, during the dry season the convection weakens and
115 shifts to the northwestern edge of geographical Amazonia (Figure 5d).

116 In the southern part of the Amazon Basin, northwesterlies prevail during the wet season
117 and are associated with the South Atlantic Convergence Zone (SACZ), a region of enhanced
118 convective activity extending from central Amazonia to the southeast [*Kodama*, 1992]. The
119 SACZ forms a quasi-stationary front with significant intraseasonal variability. As the remains of
120 mid-latitude fronts reach the tropical region [*Garreaud and Wallace*, 1998], the SACZ is
121 reinforced. In the absence of perturbations coming from the south, the SACZ weakens and may
122 disappear. This variability leads to a change in low-level winds from easterlies in the break
123 periods to westerlies in the periods of well-defined SACZ. The active and break periods are
124 related to intraseasonal oscillations [*Nogues-Paegle et al.*, 1997] that affect the whole region
125 from the northern coast [*Wang et al.*, 2002] to the more continental area [*Jones et al.*, 2002].
126 During the break periods, deep convective systems are commonly isolated, and (given the
127 absence of sinks) particle concentrations are relatively high. By comparison, during SACZ
128 events convection and rainfall are more widespread, the atmosphere is clean, and particle
129 concentrations are relatively low [*Silva Dias et al.*, 2002; *Williams et al.*, 2002].

130 With the coming of the dry season, the low-level winds in the southern part of the Basin
131 reverse, changing from northwesterlies to southeasterlies (Figures 5a and 5b). These patterns
132 favor the flow of pollution from urban and industrial Brazil into this region of Amazonia, thereby
133 increasing particle concentrations. Alternatively, cold-front southerlies (i.e., winter in the
134 Southern Hemisphere) can bring clean air into the southern part of Amazonia, in some cases
135 extending as far as the equator [*Marengo et al.*, 1997]. At these times, particle concentrations in

136 southern Amazonia can drop considerably. More typically, however, biomass burning in the
137 Basin leads to sustained high particle concentrations regardless of southerlies or southeasterlies
138 [*Artaxo et al.*, 2002].

139 The diel evolution of the planetary boundary layer (cf. Figure 6) also affects particle
140 concentrations measured at the surface. During the afternoon, the boundary layer is well mixed
141 by strong turbulence that is driven by sensible heat flux from the surface. The depth of this
142 convective boundary layer (CBL) is variable depending on land cover and on the meteorology.
143 By late afternoon CBL depths greater than 1000 m are typical in the Amazon Basin, although
144 variability is high. *Fisch et al.* [2004] observed differences between the wet and dry seasons
145 (e.g., lower than 1500 m in the former and up to 2000 m in the latter) and between forest and
146 pasture landscapes. At sunset, radiative cooling at the surface generates a nocturnal stable layer.
147 The nocturnal boundary layer usually has a depth of a few hundred meters or less. The residual
148 layer above the nocturnal stable layer is typically without turbulence or mixing, although there
149 are exceptions at times when higher-altitude shearing jets are present that induce turnover and
150 hence the cleansing of the residual layer with cleaner higher altitude air. In the absence of
151 cleansing, the nighttime residual layer conserves the properties (such as particle and gas
152 concentrations) of the previous afternoon until the following morning.

153 These basic characteristics of the daytime and nighttime boundary layers affect particle
154 concentrations measured at the surface. During the afternoon, emissions from the surface get
155 mixed though the whole volume of the convective boundary layer, diluting their concentrations.
156 Moreover, at the top of the boundary layer, turbulence in fair weather conditions maintains an
157 entrainment of cleaner air from higher altitudes while convective clouds associated with rainy
158 weather pump the aerosol particles to higher levels. These effects favor a dip of particle
159 concentrations in the early afternoon. In comparison, the stable nocturnal boundary layer traps
160 emissions in a volume near the surface, thereby favoring higher particle concentrations at night.
161 This effect is amplified during the dry season because biomass burning usually begins at mid-day
162 and continues into the evening hours. The residual layer above the nocturnal boundary layer

163 influences the surface concentrations the following morning because the development of the
164 convective boundary layer mixes the nocturnal layer into the residual layer. As a result, mid-to-
165 late morning surface concentrations (in the absence of nighttime cleansing of the residual layer)
166 can be similar to nighttime concentrations measured the afternoon before in the convective
167 boundary layer [*Rissler et al.*, 2006].

168 Given the seasonal and regional variability of the contributions from different aerosol
169 sources, the changing transport paths of air masses, and the different removal rates in the dry and
170 wet seasons, the inference (which is supported by the observations) is that Amazonian aerosol
171 particles have considerable variations in space and time, and consequently there is a considerable
172 body of literature to review. Our approach in this review is to follow the life cycle of Amazonian
173 aerosol particles, as outlined in the previous paragraphs. The review begins with a discussion of
174 the primary and secondary sources relevant to the Amazonian particle burden, followed by a
175 presentation of the particle properties that characterize the mixed populations present over the
176 Amazon Basin at different times and places. These properties include number and mass
177 concentrations and distributions, chemical composition, hygroscopicity, and cloud nucleation
178 ability. The review presents Amazonian aerosol particles in the context of natural compared to
179 anthropogenic sources as well as variability with season and meteorology. The review concludes
180 with an outlook and priorities for further research.

181 **2. Sources**

182 Amazonian aerosol particles have a wide range of natural and anthropogenic sources, and
183 the integrated effects of emission and processing in the atmosphere lead to complex internal and
184 external mixtures of particles, even within an apparently homogeneous air mass. The complexity
185 can be usefully dissected by conceptualizing a single particle as composed of components.
186 Depending on their origin, components are usually classified as primary or secondary. Primary
187 components are directly emitted from a source into the atmosphere; secondary components are
188 formed in the atmosphere [*Fuzzi et al.*, 2006]. A single particle composed mainly of primary

189 components can be called a primary aerosol particle, and a single particle composed mainly of
190 secondary components can be called a secondary aerosol particle. After some air mass aging,
191 many, if not most, individual particles can be composed of significant quantities of both types of
192 components. In the Amazon Basin, organic components typically constitute approximately 70 to
193 90% of the particle mass concentration in both the fine and coarse size fractions [*Graham et al.*,
194 2003a; *Fuzzi et al.*, 2007].

195 Examples of primary biological aerosol (PBA) particles emitted in the Amazon Basin
196 include pollen, bacteria, fungal and fern spores, viruses, and fragments of plants and animals
197 [*Elbert et al.*, 2007]. Anthropogenic biomass burning is also an important and at times dominant
198 source at some locations, especially during the dry season. In addition to sources within the
199 Amazon Basin, primary particles are also brought in by long-range transport, such as marine
200 particles from the Atlantic Ocean and desert dust or biomass-burning particles from Africa
201 [*Andreae et al.*, 1990a; *Artaxo et al.*, 1990; *Swap et al.*, 1992; *Formenti et al.*, 2001; *Chen et al.*,
202 2009]. Regional urban and industrial activities, including traffic and industry in Manaus and
203 other cities and settlements in northeastern and southern Brazil, also have outflow plumes
204 containing combustion-derived particles and dust, and these plumes are significant when a
205 sampling location lies within them.

206 Examples of components of secondary organic aerosol (SOA) are the low-volatility
207 molecules that result from the reactions of O₃ and OH with biogenic volatile organic compounds
208 (BVOCs), such as isoprene and terpenes. BVOCs are emitted in large quantities to the gas phase
209 by plants. Low-volatility BVOC oxidation products can condense from the gas phase onto pre-
210 existing particles or alternatively can contribute to new particle formation. Liquid-phase
211 reactions inside cloud droplets can also yield low-volatility BVOC oxidation products, serving as
212 another source of organic components in particles for cloud droplets that evaporate [*Blando and*
213 *Turpin*, 2000; *Lim et al.*, 2005; *Carlton et al.*, 2006].

214 2.1. Primary Particles

215 2.1.1. Primary Biological Particles

216 Emissions of primary biological particles are often wind-driven, such as suspension of
217 pollen, plant debris, or soil dust [*Jaenicke, 2005; Pöschl, 2005*]. In addition to wind-driven
218 release, certain biological organisms also actively eject materials into the air for reproductive
219 purposes, such as wet-discharged fungal spores [*Elbert et al., 2007*]. Coarse-mode PBA particles
220 in the Amazon Basin have sizes ranging from several to tens of micrometers and include
221 fragments of plants and insects, pollen grains, algae, fern spores, and fungal spores [*Graham et*
222 *al., 2003a*]. Microscopic analysis of collected particles show that, in the absence of African dust
223 and Atlantic marine emissions, morphologically identifiable biological particles dominate both
224 the number-diameter and volume-diameter distributions of the coarse fraction for natural
225 conditions (Figure 7). PBA components like carbohydrates, proteins, and lipids, as well as
226 elemental tracers, have also been detected in the fine fraction [*Artaxo and Hansson, 1995;*
227 *Andreae and Crutzen, 1997; Graham et al., 2003a*]. AMAZE-08 observations by *Sinha et al.*
228 [2009] and *Chen et al.* [2009], however, suggest that PBA components contribute in a minor way
229 to the size class below 1 μm . Nevertheless, the actual number, mass, and size of PBA particles
230 emitted to the fine fraction remain to be fully quantified for Amazonia, and the abundance and
231 composition of PBA particles is highly variable and still poorly characterized, partially because
232 the distinction between biological and other carbonaceous components requires advanced
233 analytical techniques and intensive investigation [*Pöschl, 2005; Fuzzi et al., 2006; Despres et al.,*
234 *2007*].

235 Fungi are an especially important source of coarse-mode PBA particles in the Basin.
236 They actively discharge their spores with liquid jets and droplets into the air, and these processes
237 are most active under humid conditions, such as those in the rain forest [*Gilbert, 2005*]. For
238 particles of 1 to 10 μm , fungi are estimated to contribute 25% of the particles during the day and
239 45% at night. In corroboration, measurements using an ultraviolet aerodynamic particle sizer
240 deployed during AMAZE-08 indicate that fungal spores and other viable PBA particles account
241 for up to 80% of coarse-mode particles during time periods of weak influence from sources
242 outside of the Basin (U. Pöschl, results to be published). Actively wet spore discharging fungi

243 also emit dissolved inorganic salts like potassium chloride and organic substances like sugar
244 alcohols such as mannitol in aqueous jets, and these jets break up to form droplets that can dry as
245 fine particles [Elbert *et al.*, 2007].

246 2.1.2. Biomass Burning

247 Anthropogenic aerosol particles are generated in great number and mass concentration by
248 biomass burning that is used for deforestation and pasture maintenance [Andreae *et al.*, 1988;
249 Artaxo *et al.*, 1998; Andreae *et al.*, 2002; Artaxo *et al.*, 2002; Guyon *et al.*, 2003b; Andreae *et al.*,
250 2004; Freitas *et al.*, 2005; Guyon *et al.*, 2005; Fuzzi *et al.*, 2007; Yokelson *et al.*, 2007]. The
251 resulting particles are predominantly carbonaceous, consisting largely of organic carbon but also
252 with significant nearly-elemental soot carbon and minor amounts of inorganic materials. The
253 particles consist of a variety of internal and external mixtures, including soot-carbon fractal
254 aggregates, tarry materials (sometimes present as spherical “tar balls”), grains of inorganic salts
255 (KCl, K₂SO₄, KNO₃, (NH₄)₂SO₄, etc.), and ash and char particles [Allen and Miguel, 1995;
256 Andreae *et al.*, 1998; Ferek *et al.*, 1998; Yamasoe *et al.*, 2000; Posfai *et al.*, 2003; 2004].

257 Emission factors from primary deforestation fires and pasture maintenance fires in
258 tropical rain forests range from 6 to 25 g kg⁻¹ for total particulate matter (PM) and 7.5 to 15 g
259 kg⁻¹ for PM smaller than 2.5 μm (PM_{2.5}), expressed as mass of emitted primary particles per
260 mass unit of dry fuel. For Amazonia, the estimates for the emission rates of PM_{2.5} and PM₁₀ are 8
261 and 10 Tg yr⁻¹, respectively [Yokelson *et al.*, 2008]. The pollution plumes can be hundreds of
262 kilometers across and thousands of kilometers long. With an optical thickness regularly
263 exceeding 1.0 in the peak period of biomass burning in the Basin (i.e., September and October),
264 the plumes are clearly visible in satellite images, and biomass burning is the dominant particle
265 source in affected areas. At those times, pollution from biomass smoke typically accounts for
266 >90% of the fine particles and about 50% of the coarse particles. The annual mean of optical
267 thickness over Amazonia and regions nearby is dominated by biomass-burning emissions [Tegen
268 *et al.*, 1997; Tie *et al.*, 2005].

269 Fire counts observed by satellite over the Amazon Basin in Feb, May, Aug, and Nov
270 2007 are shown. As expected, the most numerous and intense fires were in the dry season (i.e.,
271 Aug 2007 in Figure 8) in the southern part of Amazonia. Fires were also important in
272 northeastern Brazil in Nov 2007, and the prevailing flow patterns carried the biomass-burning
273 emissions into the central and southern Amazon Basin (Figure 5). In the wet season, biomass
274 burning took place along the northern rim of the Basin, and at times local meteorological
275 variability transported the biomass-burning emissions into the Basin.

276 In addition to biomass burning within South America, emissions from Africa are
277 imported into the Amazon Basin at all times of the year [Talbot *et al.*, 1990; Andreae *et al.*,
278 1994]. Fires burn in tropical and subtropical Africa year-round, with a maximum early in the
279 year in the Northern Hemisphere and a maximum in the second half of the year in the Southern
280 Hemisphere. At least some of the smoke from these fires is transported across the Atlantic by the
281 trade winds [Andreae *et al.*, 1994]. The particles emitted to the Northern Hemisphere enter
282 northern Amazonia in the wet season and can be important, given the weak baseline production
283 mechanisms of the Amazonian biosphere. In comparison, the contribution from the Southern
284 Hemisphere enters southern Amazonia during the regional burning season of the latter, and the
285 African contribution is therefore typically of less relative importance to Amazonia at that time.
286 In summary, although the highest concentrations of biomass-burning particles are observed in
287 southern Amazonia during the dry season, lower levels of biomass-burning particles can be
288 important intermittently at any time and at most locations of the Basin throughout the year.

289 2.1.3. *African Mineral Dust*

290 Saharan dust is a prominent out-of-Basin particle source. The importance of the
291 transatlantic transport of dust was recognized by Prospero *et al.* [1981] and has been observed in
292 several subsequent measurement campaigns [Swap *et al.*, 1992; Artaxo *et al.*, 1998; Formenti *et*
293 *al.*, 2001]. Imported dust occurs at its highest concentrations in those parts of the Basin that are
294 north of the ITCZ. The maximum dust concentrations at the surface are typically reached around
295 March and April, coinciding with the wet season in the central Basin. A significant fraction of

296 the delivered mineral dust is submicron, as explained by the large transport distance from Africa
297 and the preferential loss of coarse-mode particles along the way. The dust is observed at near-
298 surface stations in pulses of high concentrations that last from one to several days, and, when
299 present, mineral dust often dominates the total particle mass concentrations [*Andreae et al.*,
300 1990b; *Talbot et al.*, 1990; *Formenti et al.*, 2001; *Worobiec et al.*, 2007].

301 2.1.4. *Marine Emissions*

302 Crossing the coast of the Amazon Basin with the trade wind flow, large concentrations of
303 marine aerosol particles are progressively removed by wet and dry deposition as air masses
304 travel deeper into the Basin [*Andreae and Andreae*, 1988; *Talbot et al.*, 1988; *Andreae et al.*,
305 1990a; 1990b; *Talbot et al.*, 1990; *Worobiec et al.*, 2007]. The contribution by marine particles to
306 the total Amazonian particle mass concentration can remain significant even over the central
307 parts of the Basin. Marine aerosol particles consist in large part of primary seaspray particles,
308 which are composed mainly of coarse-mode inorganic salts mixed with lesser amounts of the
309 primary biological material that was partitioned to the ocean's surface [*O'Dowd et al.*, 2004;
310 *Andreae and Rosenfeld*, 2008]. The sodium and chloride content of the coarse fraction of the
311 particle population in the Amazon Basin is explained almost entirely by marine sources. Marine
312 aerosol particles also have a substantial contribution from secondary processes, such as sulfates
313 produced by the oxidation of dimethyl sulfide and organic material produced by the oxidation of
314 volatile organic compounds [*Ceburnis et al.*, 2008]. Much of the secondary material occurs in
315 the fine mode. In the absence of biomass-burning particles or dust, approximately half of the
316 submicron sulfate fraction is attributable to secondary sulfate produced from marine emissions,
317 especially from DMS oxidation [*Andreae et al.*, 1990a; *Worobiec et al.*, 2007].

318 Priorities for progress in identifying the sources of primary particles in the Amazon Basin
319 and quantifying their emissions include:

- 320 • Characterization and quantification of different types of primary biological, biomass burning,
321 mineral dust, and marine aerosol particles, including long-term trends, seasonal cycles, and

- 322 diel differences. Identification of their mixing states, including the relative contributions of
323 primary and secondary components.
- 324 • Discrimination and quantification of the relative fractions of in-Basin and out-of-Basin
325 sources of all particle types.
 - 326 • Improved characterization and understanding of Amazonian aerosol particles by application
327 of a combination of advanced measurement techniques, such as bulk and single-particle mass
328 spectrometry, X-ray microanalysis, fluorescence spectroscopy, electron microscopy, and
329 DNA analysis.
 - 330 • Development of process models describing the emission of primary biological particles from
331 the Amazonian ecosystem and implementation of these process models in regional and global
332 models of atmospheric chemistry, transport, and climate.

333 2.2. Secondary Gas-to-Particle Conversion

334 The production mechanisms for secondary particle components involve many trace gases,
335 in particular biogenic volatile organic compounds, nitrogen oxides (NO_x), ozone (O_3), hydroxyl
336 radical (OH), and sulfur species including dimethyl sulfide (DMS) and sulfur dioxide (SO_2)
337 [Andreae and Andreae, 1988; Jacob and Wofsy, 1988; Andreae et al., 1990a; Browell et al.,
338 1990; Jacob and Wofsy, 1990; Kesselmeier et al., 2000; Andreae et al., 2002]. DMS and SO_2 are
339 oxidized to form particle sulfate. BVOCs react with O_3 and OH to produce oxidized organic
340 products, a fraction of which have low enough volatility to condense and serve as particle
341 components. BVOCs and NO_x together influence the concentrations of O_3 and OH, thereby
342 influencing the production of BVOC oxidation products. Reactions both in the gas phase and in
343 cloud waters are important. BVOCs and NO_x emitted by biomass burning can lead to regional
344 concentrations of these trace gases and O_3 that are similar to those found in industrially polluted
345 regions.

346 For natural conditions (i.e., as defined by the influences in Figure 4), Figure 9 shows that
347 NO_x and O_3 concentrations in the convective boundary layer over Amazonia are low, of the order
348 of 20 - 200 ppb and 5 - 20 ppb, respectively. At night, NO_x concentrations within the canopy

349 space can increase by a factor of approximately 10 because the trace gases emitted from the soils
350 get trapped by the nocturnal inversion (Figure 10). At the same time, O₃ concentrations can drop
351 to nearly 0 ppb in the canopy space, both because of reactions between NO and O₃ and because
352 of deposition to the vegetation surfaces. The enrichment of trace gases in the nocturnal boundary
353 layer, especially in topographic depressions, can persist into the morning hours, thus providing
354 conditions that may be conducive to intense photochemical processes, possibly including the
355 formation of low-volatility products relevant to particle growth.

356 2.2.1. *Biogenic Volatile Organic Compounds*

357 Biogenic volatile organic compounds are emitted from plants during growth,
358 maintenance, decay, and consumption, and average emission rates account for more than 2% of
359 net primary productivity in the Basin and other regions [Zimmerman *et al.*, 1988]. The Amazon
360 Basin contains on the order of 10⁵ plant species, each having unique signatures of BVOC
361 emissions. Estimates of BVOC emissions from the whole of the Amazon Basin represent a
362 challenging but important task. Prior to new studies conducted in the past decade, Amazonian
363 BVOC emission estimates were based on a few measurements conducted by Zimmerman *et al.*
364 [1988]. Kuhn *et al.* [2007] and Karl *et al.* [2007] have shown that under some circumstances for
365 specific compounds models of biogenic emissions accurately simulate measured BVOC fluxes in
366 the region. The substantial progress in understanding Amazonian BVOC emissions and the
367 major remaining uncertainties are described in detail by Kesselmeier *et al.* [2009].

368 The specific BVOC compounds emitted and their relative rates of emissions vary widely
369 by plant species and environmental conditions. Major BVOCs emitted include isoprene (C₅H₈),
370 monoterpenes (i.e., compounds composed of two isoprene moieties), sesquiterpenes (i.e., three
371 isoprene moieties), ethane, and oxygenated VOCs (OVOCs). BVOC emissions typically increase
372 exponentially with temperature, doubling every 5 to 15 K depending on the compound. For some
373 compounds such as isoprene, emissions also increase with available sunlight, and light-
374 dependent monoterpene emissions result in a pronounced diel flux and mixing ratio cycle (Figure
375 11). The controlling mechanisms for emission differ among compounds. Some are emitted

376 immediately following production, and others are stored in plant tissues. Examples of directly
377 emitted compounds include isoprene and some types of monoterpenes and sesquiterpenes.
378 Examples of stored compounds include other types of monoterpenes and sesquiterpenes, some
379 oxygenated terpenes, and some components of plant oils. The factors influencing the emission of
380 stored compounds to the atmosphere are complex, depending on molecular vapor pressure,
381 animal herbivory, and plant phenology, moisture, or stress.

382 Although tropical forests are the dominant global source of atmospheric BVOCs and the
383 Amazon Basin is a major contributor [*Rasmussen and Khalil, 1988*], BVOC emissions have been
384 studied more extensively in temperate regions. The high species diversity in the Amazon Basin is
385 coupled with an ecological complexity and a seasonality, however, that is very different from
386 temperate regions, yielding significantly different emission trends with different forest types. For
387 example, because of the consistently high temperatures over the Amazon Basin, BVOC
388 emissions do not exhibit large seasonality there. Isoprene and monoterpene emissions and
389 concentrations are also strongly correlated in the Amazon (Figure 11), in contrast to their
390 anticorrelated behavior in temperate forests. Isoprene concentrations are highest at midday in
391 temperate forests while monoterpene concentrations are highest at night, corresponding to their
392 emission into a shallow boundary layer. Monoterpene release by plants in those forests is
393 dominated by the emission of stored compounds, and the diel monoterpene emission pattern is
394 therefore significantly different from that of isoprene, which favors released during time period
395 of intense sunlight. The explanation for the different diel monoterpene emission pattern in the
396 Amazon rain forest is not yet fully known.

397 Emissions of BVOCs have been incorporated into global chemical transport models, and
398 the contribution of low-volatility BVOC oxidation products to the mass concentration of organic
399 particles has been predicted [*Chung and Seinfeld, 2002; Tsigaridis and Kanakidou, 2003; Hoyle*
400 *et al., 2007; Heald et al., 2008*]. *Heald et al.* [2008] estimate that the conversion of South
401 American BVOCs into secondary particle mass contributes 40% of the annual global production
402 of this particle component. Simulated concentrations over the Amazon Basin vary from 0.6 to 3

403 $\mu\text{g m}^{-3}$ and peak in the dry season, corresponding to decreased wet deposition during that time
404 period.

405 A significant underestimate by models of ambient organic particle concentrations, as
406 reported for a number of anthropogenically influenced environments, has been attributed to
407 underestimated conversion yields of oxidized VOCs to secondary particle mass [Volkamer *et al.*,
408 2007]. The models employ laboratory-based yields of a few percent for isoprene and 10-15% for
409 most terpenes [Chung and Seinfeld, 2002; Henze and Seinfeld, 2006]. These yields, however,
410 may significantly underestimate what occurs over longer timescales in the atmosphere [Ng *et al.*,
411 2006]. Recent laboratory studies carried out for BVOC concentrations present in the atmosphere
412 have also shown that yields of secondary particle mass are higher than those obtained by the
413 extrapolation of earlier laboratory results carried out at higher BVOC concentrations [Shilling *et al.*,
414 2008], and this finding could potentially bring models and observations into closer
415 agreement. A mass-balance approach, based on analysis of the fate of BVOCs as either (i)
416 oxidation to CO and CO₂ or (ii) deposition with the remainder assumed to form particle mass,
417 leads to much higher production estimates of secondary particle mass [Goldstein and Galbally,
418 2007]. For the Amazon Basin, the underpredictions by chemical transport models of observed
419 concentrations appear significantly less (e.g., 35% reported by Chen *et al.* [2009]) than those
420 reported for anthropogenically influenced regions of the world.

421 Further refinement of the mass-balance approach requires better measurements of BVOC
422 oxidation products and their rates of wet and dry deposition. Studies of plant signaling, defense,
423 and food and flavor chemistry have led to the detection of thousands of individual BVOCs
424 [Hines, 2006], yet only a few of these have been studied for their emission rates, their
425 atmospheric chemistry, and their contribution to secondary particle mass. Given the limitations
426 in knowledge of emissions rates, oxidation pathways, and yields of particle mass, monoterpene,
427 sesquiterpene, and other BVOC emissions are generally lumped into a few categories for both
428 emission modeling and atmospheric chemistry modeling. As analytical techniques have
429 improved in recent decades, a much broader array of highly reactive and oxygenated BVOCs and

430 their oxidation products has been identified [*Holzinger et al.*, 2005; *Lee et al.*, 2006; *Surratt et*
431 *al.*, 2008]. The hardest to measure of the directly emitted compounds tend to be the most
432 reactive, and these same compounds, following multi-step oxidation in the atmosphere, often
433 have the highest potential for the production of secondary particle components [*Ng et al.*, 2006].
434 Sesquiterpenes, which have atmospheric lifetimes of a few minutes or less, are the best known
435 example [*Ciccioli et al.*, 1999].

436 One approach to estimate BVOC emissions for incorporation in chemical transport
437 models is a bottom-up calculation constrained by leaf, branch, or canopy-scale fluxes [*Guenther*
438 *et al.*, 1995]. Bottom-up models are based primarily on enclosure measurements that characterize
439 emissions associated with the foliage of an individual plant species. The resulting emission
440 factors are combined with the distributions of plant species to estimate landscape-level emission
441 rates. This approach, however, is difficult to apply in the Amazon Basin because of the high
442 species diversity. Direct BVOC emission measurements are available for less than 0.2% of the
443 10^5 plant species in the Amazon Basin [*Harley et al.*, 2004], and investigations performed to date
444 consist of only a few measurements per plant species, with analyses that include only a limited
445 subset of all BVOCs. Even so, *Harley et al.* [2004] demonstrate the utility of this approach for
446 specific Amazonian landscapes. Accurate characterization of the species-dependent emission
447 rates typically requires a large number of measurements over different seasons and locations due
448 to the substantial variability. Representative measurements also require sampling of upper
449 canopy foliage, which is often difficult to access in rain forests. Above-canopy flux
450 measurements, summarized by *Kesselmeier et al.* [2009] for Amazonia, provide an alternative
451 approach to the parameterization of bottom-up models.

452 Another approach to estimate BVOC emissions for use in chemical transport models is a
453 top-down calculation driven by satellite observations of the atmospheric distribution of
454 formaldehyde (HCHO). Formaldehyde is a high-yield oxidation product of isoprene and thus
455 enhancements above the methane background can be used as a proxy for this emission source
456 [*Palmer et al.*, 2003]. The first global top-down study by *Shim et al.* [2005] provided a global

457 emission estimate that was approximately 13% higher than the bottom-up estimate of *Guenther*
458 *et al.* [1995]. The estimates by *Shim et al.* [2005], however, were 35% lower for South America
459 than those of *Guenther et al.* [1995]. Recent top-down investigations have begun to specifically
460 focus on the Amazon Basin. *Palmer et al.* [2007] examined seasonal variations and found a good
461 correlation between satellite HCHO column measurements and isoprene concentrations
462 measured above the eastern part of the Basin. A related top-down study by *Barkley et al.* [2008]
463 estimated isoprene emissions that were 35% lower than the bottom-up estimates of *Guenther et*
464 *al.* [2006]. This result differed from that of *Stavrakou et al.* [2009], who estimated in a top-down
465 study that isoprene emissions were slightly higher than the bottom-up emission estimate of
466 *Guenther et al.* [2006]. The bottom-up and top-down models, nevertheless, can be considered to
467 generally agree because each has an uncertainty factor of about two.

468 After emission, some fraction of the oxidized BVOCs yields secondary particle mass.
469 *Went* [1960] first proposed that BVOCs are oxidized creating the blue haze observed in the
470 atmosphere above many forested regions. BVOCs are usually oxidized by OH, O₃, or NO₃ in the
471 atmosphere. The OH pathway is particularly important for BVOC oxidation in the tropics given
472 the high light levels and H₂O concentrations. The NO₃ pathway is minor for the usual conditions
473 in the Amazon Basin because there is low O₃ at night and low NO₂ in the day so that the rate of
474 formation by O₃ + NO₂ is slow. However, knowledge of the composition, the sources, the
475 chemistry, and the role of the secondary organic components of particles in the atmosphere and
476 earth's climate system is still extremely limited. Even for the well-studied compound isoprene,
477 recent analysis suggests that state-of-the-art atmospheric chemistry models greatly underpredict
478 OH concentrations [*Lelieveld et al.*, 2008], possibly implying important missing chemistry [*Karl*
479 *et al.*, 2009]. The OH concentrations measured in flights by *Lelieveld et al.* [2008] over the
480 Amazon forest ($5.6 \pm 1.9 \cdot 10^6$ molec cm⁻³ in the boundary layer and $8.2 \pm 3.0 \cdot 10^6$ molec cm⁻³ in
481 the free troposphere) were significantly higher than anticipated from model calculations,
482 suggesting an overlooked pathway mediated by organic peroxy radicals for OH production
483 (Figure 12). Isoprene emissions estimated using inverse modeling can be artificially low for

484 times during which OH is underestimated [Kuhn *et al.*, 2007]. In addition, higher OH estimates
485 change greatly our understanding of photochemistry in the tropics and the rate of transformations
486 of BVOCs into particle mass.

487 Opportunities for progress in identifying the correct BVOC precursors to secondary
488 particle mass include:

- 489 • *Emissions*: Develop approaches for measuring total non-methane BVOC and improve
490 approaches for measuring total oxidant reactivity (principally OH and O₃); quantify the major
491 contributions of known compounds to these totals and investigate any residuals; and
492 characterize and understand regional and seasonal variations in canopy-scale emissions using
493 airborne and tower networks of above-canopy flux measurements and satellite observations.
- 494 • *Oxidation*: Develop conceptual approaches to utilize mass balance of organic material in the
495 atmosphere as a diagnostic tool to test current understanding, to predict compounds that
496 should be in the atmosphere (oxidation products), and to search for them in a systematic way.
497 Explore by experimental or modeling studies the possible formation of low-volatility
498 products relevant to particle production by intense photochemical processes at daybreak,
499 following the enrichment of trace gases in the nocturnal boundary layer.
- 500 • *Measurements*: Develop laboratory, in situ, and remote sensing techniques to scan the
501 atmosphere for currently unmeasured compounds, to observe sums of compounds by
502 functional classes and compare with measured individual species, to more broadly utilize
503 comprehensive separation technology, and to quantify wet and dry atmospheric deposition of
504 gas-phase and particle-phase organic molecules. Focus on the broad array of semi-volatile
505 organic species present in the atmosphere, the majority of which are likely oxidation products
506 of primary BVOC emissions and can potentially condense at secondary organic components
507 of particles.
- 508 • *Modeling*: Develop models to constrain the chemical influence and fate of products from
509 atmospheric BVOC reactions, to assess the importance of additional organic compounds for
510 atmospheric photochemistry and secondary particle mass, and to represent the full range of

511 BVOCs and their gas-phase and particle-phase products in chemistry and climate
512 simulations.

513 2.2.2. *New Particle Formation*

514 Near-surface measurements of particle number-diameter distributions and ion number-
515 mobility distributions at many terrestrial sites around the globe—but excluding the Amazon
516 Basin—show that new particle formation occurs frequently [Hörrak *et al.*, 1994; Kulmala *et al.*,
517 2004; Laakso *et al.*, 2007; Iida *et al.*, 2008]. Evidence for these events (outside the Amazon) is
518 the appearance of neutral and electrically charged nanoparticles at diameters well below 10 nm
519 and their subsequent growth to larger diameters. Measurements carried out in parallel at sites
520 located distances of several hundred kilometers apart show that the events are often regional
521 [Stanier *et al.*, 2004; Vana *et al.*, 2004; Komppula *et al.*, 2006]. In contrast, measurements in the
522 Amazon Basin provide little evidence for near-surface regional-scale production of new particles
523 [Zhou *et al.*, 2002; Krejci *et al.*, 2003; 2005; Rissler *et al.*, 2004; 2006]. Whereas in other
524 continental locations, 3-nm particles are regularly observed at near-surface measurement sites
525 and also seen to grow into the Aitken mode size range above 30 nm, in the Amazon Basin the
526 smallest particles typically have sizes of 10 to 20 nm and continuous growth to larger diameters
527 is rarely observed. Growth rates for Amazonia under pristine conditions have been reported as 5
528 nm hr⁻¹ in one location for a limited set of measurements [Zhou *et al.*, 2002], implying that the
529 observed ultrafine particles nucleated 2 to 4 hr prior to the observations. Further observations are
530 needed, however, to define the possible variability of growth rates in the Amazon Basin.

531 A constraint on an observable new-particle mode is that freshly nucleated particles must
532 grow more quickly than they are scavenged by coagulation with preexisting larger particles. The
533 likelihood of satisfying this condition increases both with faster growth rates of the nucleated
534 particles and with lower concentrations of preexisting particles [McMurry and Friedlander,
535 1979; Kerminen and Kulmala, 2002; McMurry *et al.*, 2005; Lehtinen *et al.*, 2007]. The constraint
536 on an observable new-particle mode is typically satisfied for growth rates of 1 to 10 nm hr⁻¹. The
537 condensation of sulfuric acid vapor typically accounts for only a fraction of this growth [Fiedler

538 *et al.*, 2005; *Stolzenburg et al.*, 2005], with most of the balance due to condensation of VOC
539 oxidation products [*Smith et al.*, 2008]. Under the assumption that the growth rate of 5 nm hr⁻¹
540 reported by *Zhou et al.* [2005] can be broadly extrapolated to the Amazon Basin (i.e., both
541 geographically and seasonally), the discussed constraint should be satisfied and nucleation events
542 should therefore be observable as new-particle modes, and their absence in observations
543 therefore suggests that near-surface nucleation is not widespread.

544 Some evidence for the Amazon Basin shows that nucleation occurs at high altitudes and
545 that the entrainment of these particles to the near-surface layer explains surface observations of
546 ultrafine particles. Aircraft measurements above Suriname in northern Amazonia observed
547 enhanced ultrafine number concentrations at 2 to 4 km in regions of cloud outflow (Figure 13),
548 suggesting nucleation [*Krejci et al.*, 2003; 2005], which is in broad agreement with observations
549 of cloud outflow from other locations worldwide [*Perry and Hobbs*, 1994; *Clarke et al.*, 1998;
550 1999]. For comparison, measurements of vertical profiles with a tethered balloon in Melpitz,
551 Germany, showed that nucleation first occurred aloft in the residual layer prior to breakup of the
552 nocturnal inversion and then continued in the mixed layer during and after break-up, all in the
553 absence of clouds [*Stratmann et al.*, 2003]. Over the boreal forests of Finland, a similar
554 measurement program showed that nucleation occurred within the boundary layer but not aloft
555 [*Laakso et al.*, 2007]. In the Amazon Basin, nocturnal events of apparent nucleation, which
556 cannot be explained by outflow from deep convective clouds, have also been observed
557 intermittently for short periods [*Rissler et al.*, 2004; 2006]. Diel patterns of intermittent
558 nucleation were similar for a wide variety of conditions, including periods of intensive biomass
559 burning as well as natural conditions. Nucleation-mode particle concentrations were highest at
560 sunrise and sunset, with average concentrations exceeding 1000 cm⁻³ (Figure 14). Similar
561 nocturnal events were observed in an Australian eucalypt forest [*Suni et al.*, 2008].

562 The implications of these observations for the Amazon Basin compared to those
563 worldwide are that several different chemical processes may be capable of separately inducing
564 nucleation and growth and, further, that these different processes may occur in different regions

565 of the atmosphere. Further research, however, may yet succeed in unifying these presently
566 disparate observations by using a mechanistic approach to the problem. A comprehensive
567 analysis of particle formation events recorded at continental locations around the world shows
568 that the nucleation rates J , which quantify the rates at which stable molecular clusters are
569 produced, satisfy the following empirical expression [Riipinen *et al.*, 2007; Kuang *et al.*, 2008]:
570 $J = k[\text{H}_2\text{SO}_4]^p$ for $1 < p < 2$, where $[\text{H}_2\text{SO}_4]$ is the sulfuric acid vapor concentration and k is a
571 kinetic prefactor that varies from location to location. Mechanisms responsible for the variability
572 in k are not yet understood. One hypothesis is that k accounts for the concentrations of species
573 that co-nucleate with sulfuric acid. Laaksonen *et al.* [2008] have proposed that BVOC oxidation
574 products may be important co-nucleating species over forested regions.

575 Of critical importance for the application of this nucleation equation to the near-surface
576 layer over the Amazon Basin are the weak sulfur sources within the Basin, which lead to sulfur
577 dioxide concentrations of typically 20 to 30 ppt [Andreae and Andreae, 1988; Andreae *et al.*,
578 1990a]. This value is more than an order of magnitude lower than the values commonly found
579 under remote conditions over the continents of the Northern Hemisphere. This concentration of
580 SO_2 plausibly implies correspondingly low gas-phase H_2SO_4 concentrations, although no direct
581 observations have ever been made in the Amazon Basin to provide confirmation. Simulated peak
582 daytime near-surface H_2SO_4 concentrations are less than $5 \times 10^5 \text{ cm}^{-3}$ (0.019 ppt) [Spracklen *et*
583 *al.*, 2005]. By the above nucleation equation, this modeled H_2SO_4 concentration is too low to
584 result in near-surface particle formation because preexisting particles will scavenge any incipient
585 molecular clusters before they grow to new particles [Spracklen *et al.*, 2006]. An alternative
586 mechanism to the H_2SO_4 pathway, namely ion-mediated nucleation, is also modeled as an
587 unimportant source of nuclei over the Basin [Yu *et al.*, 2008].

588 In comparison to the absence of predicted new particle formation in the near-surface
589 region of the Basin, models predict that new particle formation upwind or aloft, in particular
590 within the upper troposphere followed by growth and entrainment into the near-surface layer,
591 contributes significantly to the Amazonian particle number concentrations, especially during the

592 wet season [*Spracklen et al.*, 2005]. Although sufficient for nucleation at higher altitudes,
593 modeled H₂SO₄ concentrations are insufficient to explain the subsequent rate of particle growth
594 observed in convective outflow over the Basin, suggesting that other gas-phase species such as
595 the oxidation products of BVOCs may have a role [*Ekman et al.*, 2008].

596 Opportunities for progress to better constrain and quantify mechanisms of new particle
597 formation over the Amazon include:

- 598 • *H₂SO₄ concentration*: Perform simultaneous observations of sulfuric acid vapor
599 concentrations, particle nucleation rates, and particle number-diameter distributions over the
600 Amazon Basin. Establish whether the apparent absence of surface level nucleation in the
601 Basin is consistent with our understanding of the atmospheric conditions that lead to new
602 particle formation in other locations [*McMurry et al.*, 2005].
- 603 • *Particle growth rates*: Observe and explain growth rates of ultrafine particles over the Basin.
604 Quantify the contribution of sulfuric acid and BVOC oxidation products to the observed
605 rates.
- 606 • *Ion mobility distributions*: Deploy an air-ion spectrometer [*Hörrak et al.*, 1994; *Mirme et al.*,
607 2007] in the Basin to give information on very small particles (diameter < 3 nm) and the
608 earliest steps of new particle formation.
- 609 • *Modeling*: Develop models to assess the contribution of different nucleation mechanisms to
610 aerosol particles in the Basin. Evaluate candidate mechanisms by comparisons of model
611 predictions made using these mechanisms against past and newly available observations.

612 **3. Properties**

613 3.1. Mass Concentration

614 The mass concentrations of particles in the Amazon Basin vary strongly with season and
615 location, modulated to the largest extent by the influence of in-Basin biomass burning with other
616 important influences by the episodic long-range transport of African dust and biomass burning
617 [*Artaxo et al.*, 2002; *Guyon et al.*, 2003b]. Figure 15 shows the time series of PM₁₀

618 concentrations for central Amazonia (Balbina), eastern Amazonia (Santarem), and southern
619 Amazonia (Alta Floresta). These locations are influenced seasonally in varying degrees by in-
620 Basin biomass-burning emissions. In Alta Floresta (Aug 1992 to Feb 2005), two different
621 prevailing regimes of mass concentration occur. In the wet season, in the absence of biomass
622 burning, the PM_{10} concentration is 9 to 12 $\mu\text{g m}^{-3}$, with a fine fraction of 2 to 3 $\mu\text{g m}^{-3}$. In the dry
623 season, the PM_{10} concentration approaches 300 to 600 $\mu\text{g m}^{-3}$, producing an optical thickness of
624 more than 4 at 500 nm [Schafer *et al.*, 2008].

625 For comparison, in central Amazonia where the influence of biomass burning is less, the
626 mass concentration is low even in the dry season (Figure 15, Balbina), with an average PM_{10}
627 concentration of 11 $\mu\text{g m}^{-3}$ (Oct 1998 to Feb 2005). The typical concentration of fine particles
628 increases from 2 $\mu\text{g m}^{-3}$ in the wet season to 4 $\mu\text{g m}^{-3}$ in the dry season. The corresponding fine-
629 mode black-carbon-equivalent (BC_e ; cf. §3.3.1) mass concentration ranges from 100 to 150 ng
630 m^{-3} during the wet season and from 600 to 800 ng m^{-3} during the dry season. Abrupt pulses of
631 relatively high mass concentration can occur in both the fine and coarse fractions (Figure 15,
632 Balbina), and these pulses are explained by African dust outflow that reaches the observation
633 site.

634 The influence of biomass burning on mass concentration in Santarem in eastern
635 Amazonia is intermediate compared to Balbina and Alta Floresta. From March 2000 to January
636 2005 in Santarem, the PM_{10} concentration increases from about 10 $\mu\text{g m}^{-3}$ in the wet season to 40
637 $\mu\text{g m}^{-3}$ in the dry season. The fine fraction, typically as low as 2 $\mu\text{g m}^{-3}$ in the wet season,
638 reaches 20 to 30 $\mu\text{g m}^{-3}$ in the dry season. During the wet season, the ratio of the fine-to-coarse
639 fraction is lower for Santarem than Balbina, possibly suggesting an increased relative importance
640 of out-of-Basin coarse-mode particles over eastern Amazonia. Figure 15 also shows that the
641 influence of biomass burning is strongest in the fourth quarter of each year at Santarem whereas
642 it is strongest in the third quarter at Alta Floresta. These seasonal patterns match those of
643 vegetation fires in eastern and southern Amazonia, respectively, as is apparent in Figure 8.

644 Figure 16 (top) shows that the particle mass-diameter distribution is dominated for
645 natural conditions by coarse-mode particles, corresponding to primary biological particles
646 possibly coated by secondary material. These data were obtained by gravimetric analysis of the
647 stages of a Multi-Orifice Uniform Deposit Impactor (MOUDI) during Mar and Apr 2008 as part
648 of AMAZE-08 in central Amazonia [Martin *et al.*, 2009]. The selected data correspond to time
649 periods during which the influence of sources outside of the Amazon Basin was weak [Chen *et*
650 *al.*, 2009]. Artaxo and Hansson [1995] applied principal component analysis to the elemental
651 composition and the mass concentrations recorded on five stages of a cascade impactor for
652 various levels within the canopy and found that the concentrations of potassium and phosphorus,
653 indicative of primary particles, were prominent in the coarse fraction, especially during the night.

654 The mass-diameter distribution shifts from the coarse to the fine fraction during times of
655 strong influence by in-Basin biomass burning. Figure 16 (bottom) shows MOUDI measurements
656 recorded during LBA-SMOCC in southwestern Amazonia (Rondônia). The total particle mass
657 concentration was $154 \mu\text{g m}^{-3}$, emphasizing the overwhelming influence of biomass-burning
658 particles compared to any other types at the time of sampling. The figure shows that the mass-
659 diameter distribution is heavily loaded in the accumulation mode, with a mass median
660 aerodynamic diameter between 0.33 and $0.56 \mu\text{m}$. Also apparent, however, is that the mass
661 concentrations in the coarse fraction are similar between the top and bottom panels of Figure 16.
662 The implication is that the mass concentration of coarse-mode natural biogenic particles
663 (possibly coated by secondary materials) is similar in the dry and wet seasons as well as for
664 different locations within the Amazon Basin.

665 Opportunities for progress to identify and quantify the processes that control mass
666 concentrations and mass-diameter distributions of particles over the Amazon Basin include:

- 667 • Measurements show that long-range transport aerosol particles from Africa and the Atlantic
668 Ocean are present within the Amazon Basin and at times dominate observations, yet the
669 concentrations and distribution of these out-of-Basin particles are not fully understood,
670 especially in relation to the extensive cloud development and opportunities for wet

671 scavenging within the Basin. Long-term measurements at a minimum of three locations
672 across the Basin along an east-west transect are therefore motivated to provide observations
673 of concentration gradients. Such a data set would provide insights into the sources of
674 particles from outside the Basin as well as sinks within the Basin, thereby placing new
675 constraints on the processes affecting particle lifetime.

- 676 • Systematic measurements of the size-segregated composition of aerosol particles are scarce
677 in the Amazon Basin. Such measurements are particularly needed in the remote and
678 seasonally pristine areas in the western half of the Basin.

679 3.2. Number-Diameter Distribution

680 At times of prevailing natural conditions, particle number concentrations in the mixed
681 layer of the Basin are 200 to 400 cm⁻³, lower by 10 to 100 times than continental and urban
682 concentrations elsewhere in the world having high anthropogenic influence. The baseline, low
683 number concentrations in the Basin are derived from biogenic sources that have low variability
684 and low concentration throughout the year, and strong deviations from this norm that are
685 episodically observed at near-surface stations must be explained either by local pollution or long-
686 range transport from distant sources [*Pauliquevis et al.*, 2007].

687 The particle number-diameter distributions have been measured in the Basin by
688 techniques such as mobility particle sizing (diameter range of 3 to 850 nm) (e.g., *Zhou et al.*
689 [2002]), optical particle counting (diameters 100 nm and larger) (e.g., *Krejci et al.* [2003]), and
690 aerodynamic particle sizing (diameters of 500 nm and larger) (e.g., *Rissler et al.* [2006]). A
691 typical number-diameter distribution measured by a differential mobility particle sizer (DMPS)
692 operated during a period of natural conditions (i.e., CLAIRE-98 wet season in central Amazonia)
693 is shown in Figure 17a [*Zhou et al.*, 2002]. Aitken and accumulation modes are apparent at 70
694 and 150 nm, respectively. The histograms of Aitken, accumulation-mode, and particle number
695 concentrations for the observations from CLAIRE-98 are shown in Figures 17b-d, and the modal
696 statistics are summarized in Table 2. The characteristic diameters of the nucleation, Aitken, and
697 accumulation modes are well confined within diameter ranges, and the modes are clearly

698 separated (Table 2). These results of CLAIRE-98 are comparable to those of CLAIRE-01
699 [Rissler *et al.*, 2004] at the same site and of AMAZE-08 about 100 km away (Figure 18, top) (E.
700 Swietlicki, unpublished data).

701 In Figure 17 a Hoppel gap, which is persistent in Amazonian number-size distributions,
702 separates the Aitken and accumulation modes and differentiates to a large extent those particles
703 that have been subjected to in-cloud processing from those that have not. According to *Hoppel et*
704 *al.* [1986; 1994], cloud-droplet activation on Aitken particles is followed by the uptake and
705 reaction of soluble gases to form low-volatility products, and net diameter growth therefore
706 occurs upon evaporation of the cloud droplets. In the Amazon Basin, the soluble gases that
707 become fixed are expected mostly to be BVOCs, which then react in the cloud waters to form at
708 least in part low-volatility BVOC oxidation products (i.e., SOA particle components) [*Blando*
709 *and Turpin*, 2000; *Lim et al.*, 2005; *Carlton et al.*, 2006]. Alternative mechanisms for converting
710 Aitken particles into the accumulation mode, such as out-of-cloud coagulation and
711 condensational growth, are too slow in clean atmospheres such as the green-ocean Amazon.

712 In fair weather, a diel pattern during CLAIRE-98 in the number concentration of the two
713 modes, specifically that the Aitken number concentration decreased while that of the
714 accumulation mode increased as the day progressed, was linked to the diel variation of the lower
715 atmosphere (Figure 6) [*Zhou et al.*, 2002]. Cloud processing above the boundary layer leads to
716 the depletion of Aitken particles and to the growth of accumulation particles in that layer, and
717 this cloud convective layer mixes into the surface layer during the day, strongly influencing
718 observations there. In rainy weather, the behavior is interrupted by strong scavenging of particle
719 number, volume, and mass, followed by quick recovery (due to regional mixing) after rainfall
720 ceases.

721 Measurements in CLAIRE-98 showed that the diameters of both the Aitken and
722 accumulation modes continuously increased from sunrise to sunset, with few exceptions. The
723 growth of the accumulation-mode particles was attributed to cloud processing, with downward
724 mixing of these larger particles throughout the day. The growth of the Aitken particles (ca. 5 nm

725 hr⁻¹) was plausibly explained by the condensation of low-volatility vapors resulting from the
726 photooxidation of BVOCs (i.e., SOA production) (cf. §2.2.1). Much of this particle growth was
727 proposed to occur in the boundary layer itself. Rainfall temporarily halted the steady diameter
728 growth of the Aitken particles. This observation suggested a down-mixing of somewhat smaller
729 Aitken particles in association with cold downdrafts.

730 These near-surface observations in central Amazonia during CLAIRE-98, CLAIRE-01,
731 and AMAZE-08 were made on air masses that had spent several days within the Basin and thus
732 were highly processed (e.g., ecosystem emissions, cloud cycles, and so forth). For contrast,
733 measurements made during flights over Suriname and French Guyana during CLAIRE-98 were
734 taken as air masses first entered the Amazon Basin and to some extent can be considered the
735 initialization conditions for processing within the Basin [Krejci *et al.*, 2003]. The average
736 number-diameter distribution from the lowest flight level within the well-mixed boundary layer
737 (0.2 - 1.2 km) is shown in Figure 18 (bottom). This distribution is depleted in Aitken particles
738 compared to the near-surface measurements described for central Amazonia (e.g., CLAIRE-98)
739 (cf. top and bottom of Figure 18), offering evidence for the formation of Aitken particles within
740 the Basin.

741 The particle number-diameter distribution changes greatly at locations in the Basin that
742 are influenced by in-Basin biomass burning [Reid *et al.*, 2005]. The mode of the distribution is
743 typically between 100 and 160 nm. The higher values are found in more aged pollution plumes
744 as a result of coagulation with other particles and the condensation from the gas phase of low-
745 volatility species. Given these high concentrations, when they are present, aerosol particles
746 resulting from in-Basin biomass burning dominate the overall features of the Amazonian aerosol,
747 tending to minimize the impact of other processes such as particles imported from outside of the
748 Basin or the processes of the natural Amazonian biosphere.

749 Measurements using a DMPS conducted at a ground site during LBA-SMOCC showed a
750 single number median diameter of 135 nm for fresh smoke and average particle number
751 concentrations of 10,500 cm⁻³ for the diameter range of 30 to 850 nm [Rissler *et al.*, 2006]. The

752 number concentration of particles in the nucleation mode (i.e., from 3 to 30 nm diameter) was
753 also relatively high, averaging 800 to 1000 cm⁻³, although their presence was intermittent.
754 Airborne measurements of particle number-diameter distributions were also performed [*Guyon et*
755 *al.*, 2005], and the geometric mean diameters were 110 ± 15 nm in 69 plumes within the
756 boundary layer and 139 ± 17 nm for 50 smoke plumes detrained above the boundary layer,
757 mostly from non-precipitating clouds. Biomass-burning particles that enter higher altitudes and
758 escape wet deposition can be exported out of the Basin, affecting particle number and mass
759 concentrations in distant regions of the Southern Hemisphere.

760 Opportunities for progress in identifying and quantifying the processes that control the
761 number concentration and the number-diameter distribution of particles in the Amazon Basin
762 include:

- 763 • *Processes*: The processes that control the number-diameter distribution over the Amazon
764 Basin are not fully understood. The consistent appearance of the Aitken and accumulation
765 modes in confined diameter windows points to the importance of in-cloud processing, but the
766 source and sink rates of Aitken and accumulation-mode particles, as well as the vertical
767 structure and mixing of particles (including quantifying of cloud convective mixing and
768 associated downdrafts as mechanisms for entraining particles into the boundary layer), must
769 yet be quantified.
- 770 • *Measurements and Instrumentation*: Long-term (i.e., years) size-distribution measurements
771 in a pristine environment using an air ion spectrometer, twin scanning mobility particle
772 sizers, ultraviolet and normal aerodynamic particle sizers, preferably both just over the
773 canopy as well as higher on a tall tower to observe vertical gradients. Size-resolved particle
774 number fluxes by eddy-covariance techniques, preferably at several altitudes in a high tower.
775 Tethered balloons equipped with condensation particle counters having various lower-size
776 cutoff diameters or diffusion batteries to locate altitudes having increased new particle
777 formation. Ground-based long-term LIDAR measurements for vertical profiling. These
778 instruments should be applied to closure studies between (i) number and mass, (ii) the

779 number-diameter distribution and light scattering, and (iii) the number-diameter distribution
780 (including hygroscopic properties) and the concentration of cloud condensation nuclei.

781 3.3. Chemical Composition

782 Aerosol particles in the Amazon Basin are composed mainly (i.e., 80 to 90%) of organic
783 carbon (OC), although African mineral dust and Atlantic sea salt can dominate the mass
784 concentration for short periods [Talbot *et al.*, 1990].

785 3.3.1. Organic Component

786 At times of prevailing natural conditions, typical OC concentrations are $1 \mu\text{g}\cdot\text{C m}^{-3}$ or
787 less in the fine fraction and 1 to 3 $\mu\text{g}\cdot\text{C m}^{-3}$ in the coarse fraction [Formenti *et al.*, 2001; Graham
788 *et al.*, 2003a; Guyon *et al.*, 2003b]. A ratio of 1.6 to 1.7 for OM:OC (i.e., the ratio of organic
789 mass to organic carbon) is estimated from the high-resolution mass spectra of Chen *et al.* [2009]
790 and in agreement with value of Fuzzi *et al.* [2007]. Molecular characterization by
791 chromatography has been carried out for the water-soluble organic fraction [Graham *et al.*,
792 2003b; Claeys *et al.*, 2004]. Results are shown in Figure 19 for particles collected on fine-
793 fraction filters [Decesari *et al.*, 2006]. The identified composition ($0.10 \mu\text{g}\cdot\text{C m}^{-3}$) is less than
794 20% of the water-soluble organic carbon (WSOC; $0.85 \mu\text{g}\cdot\text{C m}^{-3}$), indicating the presence of a
795 wide range of other unidentified organic compounds. Dicarboxylic and hydroxyacids are
796 persistently higher during the day compared to night, consistent with the condensation from the
797 gas phase of BVOC oxidation products [Graham *et al.*, 2003a]. Likewise, the detection of
798 methyltetrols is a signature of isoprene photooxidation [Claeys *et al.*, 2004]. BVOC oxidation
799 products may also explain the elevation of fine-fraction OC concentrations by 1.4 ± 0.2 times
800 during the day compared to the night (i.e., greater plant emissions and greater sunlight during the
801 day), although enhanced convective downward mixing of particles from aloft can also play an
802 important role [Graham *et al.*, 2003a, b; Claeys *et al.*, 2004]. The presence of low levels of
803 anhydrosugars (such as levoglucosan, mannosan, and galactosan) even during time periods for
804 which natural conditions prevail demonstrates an influence of biomass burning [Graham *et al.*,
805 2003a], possibly indicative of the long-range transport of emissions from African fires.

806 For AMAZE-08, *Chen et al.* [2009] report the results of real-time mass spectrometric
807 measurements of submicron particles for a time period of natural conditions. During periods of
808 weak out-of-Basin influence, patterns and identifier peaks in the mass spectra closely resemble
809 those of secondary particle components formed by the oxidation of BVOCs in environmental
810 chambers. Most of the mass concentration of submicron organic particles is attributed for the
811 period of AMAZE-08 to the condensation of BVOC oxidation products as secondary particle
812 components, with a smaller amount to highly oxidized materials representative of humic-like
813 substances (HULIS) that arrive by long-range transport (e.g., aged biomass burning emissions
814 from Africa).

815 In regard to the coarse fraction, *Graham et al.* [2003a] quantify sugars, sugar alcohols,
816 and fatty acids, providing strong evidence for the release of primary biological particles into the
817 forest atmosphere. Trehalose, mannitol, arabitol, and the fatty acids are more prevalent at night,
818 coinciding with a nocturnal biological activity that increases the release rates of yeasts and other
819 small fungal spores. Glucose, fructose, and sucrose are persistently higher during the day,
820 coinciding with a daytime increase in large fungal spores, fern spores, pollen grains, and, to a
821 lesser extent, plant fragments, as driven by lowered relative humidity and enhanced wind speeds
822 and convective activity during the day. Although mass emissions are reduced at night, coarse-
823 fraction OC concentrations are, nevertheless, elevated at night compared to day by a mean factor
824 of 1.9 ± 0.4 , which is explained by trapping of emitted particles in the nocturnal boundary layer.

825 The organic compounds constituting the natural particles can be light absorbing.
826 Although black-carbon-equivalent (BC_e) concentration is typically less than $0.1 \mu\text{g m}^{-3}$ and
827 represents under 5% of the total carbon concentration [*Formenti et al.*, 2001; *Graham et al.*,
828 2003b], the BC_e concentrations are higher than those of elemental carbon (EC), implying that
829 organic components are contributing to the absorption of light [*Guyon et al.*, 2003a; c]. Because
830 these compounds have a steep increase of light absorption with decreasing wavelength, resulting
831 in a brown color of the filter samples, they have been termed “brown carbon” [*Andreae and*
832 *Gelencser*, 2006]. The light-absorbing material is mainly in the coarse fraction and can be

833 explained mostly by chromophores present in primary biological particles and certain BVOC
834 oxidation products.

835 In addition to natural particles, the composition of biomass-burning particles in the Basin
836 has also been extensively studied [*Penner et al.*, 1991; *Andreae*, 1993; *Falkovich et al.*, 2005;
837 *Decesari et al.*, 2006; *Fuzzi et al.*, 2007]. The biomass-burning particles, found mostly in the fine
838 size fraction (cf. §3.2), are most predominant in southern Amazonia and downwind of it (Figure
839 5), but even remote areas in northern and central Amazonia are subject to the large-scale
840 transport of biomass-burning emissions [*Pauliquevis et al.*, 2007]. The particles consist of 85 to
841 90% organic carbon [*Talbot et al.*, 1990; *Yamasoe et al.*, 2000; *Soto-García et al.*, 2009], of
842 which more than half is water soluble [*Graham et al.*, 2002; *Mayol-Bracero et al.*, 2002;
843 *Decesari et al.*, 2006]. The balance of 10 to 15% has typically been operationally defined as
844 black carbon (i.e., apparent elemental carbon). The OC and BC_e concentrations have a diel
845 variability arising both from variations in the thickness of the atmospheric boundary layer and
846 the frequency of fires [*Soto-García et al.*, 2009]. Biomass-burning particles also are an important
847 source of water-soluble organic nitrogen, including urea and several amino acids. For example,
848 *Mace et al.* [2003] report concentrations of 0.9 µg·N m⁻³ in the dry season compared to 0.05
849 µg·N m⁻³ in the wet season (i.e., a 20-fold difference).

850 Figure 20 shows the composition of biomass-burning particles in Amazonia, as obtained
851 from a synthesis of data from multiple complementary techniques (e.g., gas-chromatography
852 mass spectrometry (GC-MS) and high-pressure liquid chromatography (HPLC)) [*Mayol-Bracero*
853 *et al.*, 2002]. The organic compounds are a complex mixture of differing molecular structures,
854 physical properties, and reactivities [*Jacobson et al.*, 2000]. Molecular speciation using GC-MS
855 accounts for about 10% of the WSOC. The identified species are mostly pyrolysis products of
856 cellulose, hemicellulose, and lignin [*Graham et al.*, 2002; *Zdrahal et al.*, 2002; *Claeys et al.*,
857 2004; *Trebs et al.*, 2005; *Decesari et al.*, 2006]. Levoglucosan, a primary vegetation combustion
858 product, is the single most abundant compound identified [*Schkolnik et al.*, 2005; *Fuzzi et al.*,
859 2007]. It is enriched in samples collected at night compared to those from the day, reflecting the

860 shift from flaming fires in the day to smoldering fires at night [Fuzzi *et al.*, 2007]. The balance of
861 about 90% that eludes analysis by molecular chromatography is expected to be chemical
862 compounds that are predominantly of high molecular weight [Mayol-Bracero *et al.*, 2002; Hoffer
863 *et al.*, 2006]. The HPLC results show that neutral molecules, mono and dicarboxylic acids, and
864 polycarboxylic acids represent about 70% of the WSOC. Decesari *et al.* [2006] propose model
865 compounds to reproduce quantitatively the average chemical structure of the WSOC, with the
866 intention that these model compounds can be used as best-guess surrogates in microphysical
867 models.

868 *Fuzzi et al.* [2006] provide excellent recommendations in a general context on the
869 research needs for improved chemical characterization of organic aerosol particles. For the
870 specific context of the Amazon, high-priority opportunities for increased chemical
871 characterization include:

- 872 • *Sampling*: Use of sampling techniques such as denuders and real-time measurements that
873 reduce positive and negative sampling artifacts. As necessary, the development of new
874 techniques for these purposes. Control of sampling with wind direction to facilitate the
875 interpretation of the results. Increased use of size-segregated sampling.
- 876 • *Analysis*: Development of analytical methods to improve chemically resolved mass balance,
877 including both at the molecular and common-property levels. Innovation of new methods for
878 the analysis of high-molecular weight compounds. Determination of the composition of light-
879 absorbing material. Development of analytical techniques for airborne measurements having
880 a similar capability as ground-level measurements.
- 881 • *Composition*: Identification and characterization of the molecules and the molecular families
882 constituting the water-insoluble organic fraction. Determination of the relative contributions
883 of humic-like substances and BVOC oxidation products to OC mass concentrations during
884 the wet and dry seasons. Characterization and improvement of the understanding of water-
885 soluble organic nitrogen in biomass-burning particles.

886 3.3.2. *Inorganic Component*

887 Table 3 summarizes the composition of the inorganic component of Amazonian aerosol
888 particles. The inorganic component typically constitutes 10 to 20% of the total mass in the fine
889 fraction and less in the coarse fraction, with the balance largely from organic components and at
890 times African dust and African and South American biomass burning.

891 The two most commonly applied techniques for the study of the inorganic composition
892 have been off-line ion chromatography and proton-induced X-ray emission (PIXE). Samples
893 were collected for at least half a day up to several days on a single filter to obtain sufficient
894 quantities for analysis. The SMOCC campaign (Table 1) introduced the use of selective semi-
895 continuous measurements of water-soluble inorganic compounds (i.e., SO_4^{2-} , NO_3^- , Cl^- , and
896 NH_4^+) via a steam-jet aerosol collector and their gaseous precursors (i.e., SO_2 , HNO_3 , HCl , NH_3)
897 by use of a rotating wet-annular denuder [Trebs *et al.*, 2004; 2005]. In AMAZE-08, an Aerodyne
898 aerosol mass spectrometer was employed to quantify non-refractory sulfate, nitrate, and
899 ammonium every 5 min (Figure 21) [Chen *et al.*, 2009].

900 Sulfate is the major water-soluble inorganic anion and is primarily distributed in the fine
901 fraction. Sulfate is formed by the reactions of DMS, H_2S , and CS_2 emitted by plants and
902 microorganisms within the Amazon Basin, and the in-Basin source contributes approximately
903 $0.05 \mu\text{g m}^{-3}$ to the sulfate mass concentration [Andreae *et al.*, 1990a]. Even for natural
904 conditions, however, sulfate concentrations averaged over several weeks are 3 to 5 times greater
905 than this in-Basin contribution (Table 3) [Artaxo *et al.*, 1990; Formenti *et al.*, 2001; Artaxo *et al.*,
906 2002; Fuzzi *et al.*, 2007]. Part of the explanation is that marine DMS is persistently imported
907 from the Atlantic Ocean, representing an integral part of the natural processes of the Basin.
908 Episodic importation of sulfate included as part of African dust and biomass-burning particles
909 also occurs, and these episodes can increase sulfate concentrations in both the fine and coarse
910 fractions by factors of 2 to 3 or more [Talbot *et al.*, 1990; Formenti *et al.*, 2001]. In the dry
911 season, the average sulfate concentration increases by a factor of 2 to 3 (Table 3), even for
912 nominally clean conditions. The increase is attributed in large part to the reduction in wet

913 deposition as well as to the presence of biomass-burning particles diluted throughout the Basin,
914 rather than a change in biogenic gaseous emissions or shifts in the patterns of imported sulfate
915 precursors [Artaxo *et al.*, 1988; Talbot *et al.*, 1988; Artaxo *et al.*, 2002; Graham *et al.*, 2003a;
916 Fuzzi *et al.*, 2007]. At sampling locations and times repeatedly and heavily affected by biomass
917 burning (e.g., in plumes), the average sulfate concentration can increase by a factor of 10 or
918 more.

919 Nitrate is found predominately in the coarse fraction of natural Amazonian aerosol
920 particles and, compared to sulfate, is a minor component by mass (Table 3) [Talbot *et al.*, 1988;
921 1990; Graham *et al.*, 2003a; Fuzzi *et al.*, 2007]. The ambient HNO₃ concentrations are too low
922 and the temperatures typical of the Basin are too high to favor elevated concentrations of
923 particle-phase nitrate. Significant nitrate enrichment to the fine fraction occurs, however, for
924 locations affected by in-Basin biomass burning, which can be attributed to increased NO_x
925 emissions followed by oxidation and subsequent condensation of HNO₃ [Trebs *et al.*, 2004]. At
926 these times, the nitrate and sulfate concentrations can be nearly equal [Trebs *et al.*, 2005].

927 Ammonium concentrations in the fine and coarse fractions are typically more than
928 sufficient during both the wet and dry seasons to neutralize the nitrate and sulfate in each size
929 fraction, particularly at night [Talbot *et al.*, 1988; Graham *et al.*, 2003a; Mace *et al.*, 2003; Fuzzi
930 *et al.*, 2007]. Trebs *et al.* [2005] conclude that organic anions, such as formate, acetate, and
931 oxalate, must balance the ammonium in the particles, implying the presence of species such as
932 ammonium oxalate. At locations having a strong influence from biomass burning, ammonium
933 concentrations, along with those of nitrate, are elevated in the fine fraction during the night
934 [Trebs *et al.*, 2004], which can be explained by higher concentrations of HNO₃ and NH₃ in the
935 vicinity of biomass burning and favorable gas-to-particle partitioning for higher relative
936 humidity and cooler temperatures.

937 The mass concentrations of crustal elements (e.g., Al, Si, Fe, Ti, and Mn) are low in the
938 absence of episodic inputs of African mineral dust (cf. Table 3) [Talbot *et al.*, 1990; Formenti *et*
939 *al.*, 2001; Graham *et al.*, 2003a; Trebs *et al.*, 2005]. During episodes, their concentrations are

940 equally distributed in the fine and coarse fractions, representing a fractionation that is
941 significantly different from the dominance of the coarse mode near source regions. After long
942 transport over the Atlantic Ocean, the coarse mode of these elements is lost preferentially over
943 the fine mode, resulting in the observations seen for the Amazon, including an important tail into
944 the accumulation mode. Wet-season concentrations are several times higher than those during
945 the dry season, which is explained by a shift of the ITCZ that favors the transport to the Amazon
946 Basin of air from the Saharan region (cf. §1).

947 The mass concentrations of P, K, and Zn in the coarse fraction are derived primarily from
948 PBA particles [*Artaxo and Hansson, 1995; Guyon et al., 2003c*], as evidenced by their greater
949 concentrations underneath the forest canopy compared to above it and by their greater
950 concentrations at night than day (cf. §3.3.1). Although the long-range transport of phosphorus
951 from African dust is important for the fertilization of the Basin on long timescales [*Mahowald et*
952 *al., 2005*], the incremental concentrations are low and difficult to detect compared to the in-
953 Basin cycling of phosphorus by PBA particle emission and deposition [*Formenti et al., 2001*].
954 For natural conditions, K occurs almost exclusively in the coarse fraction, but a fine mode
955 appears at times of biomass-burning influence, increasing the mass concentration by a factor of
956 10 or more in fresh plumes downwind of biomass burning [*Echalar et al., 1998; Yamasoe et al.,*
957 *2000; Artaxo et al., 2002; Maenhaut et al., 2002; Trebs et al., 2005*]. Fire is therefore important
958 for the recycling of K throughout the Basin, and its presence in the fine fraction is often used as a
959 marker for the influence of biomass burning.

960 High priorities for the improved chemical characterization of the inorganic component of
961 Amazonian aerosol particles include:

- 962 • *Instrumentation*: Positive and negative sampling artifacts can be introduced, especially for
963 semi-volatile compounds like ammonium nitrate, by swings in temperature and relative
964 humidity during the long sampling periods required by many techniques when employed at
965 the low particle mass concentrations characteristic of the Amazon Basin [*Solomon and*

966 *Sioutas, 2008; Trebs et al., 2008*]. The development of instrumentation that can overcome
967 these obstacles in tropical environments is needed.

- 968 • *Processes*: Investigations are needed (1) that target the influence of water-soluble organic
969 compounds on the gas-particle partitioning of inorganic species, (2) that test for the possible
970 presence of organosulfates and understand how these compounds may be tracers for the
971 sources of particle components, and (3) that focus on connections between inorganic
972 compounds and their physical properties (e.g., their activity as cloud condensation nuclei).
- 973 • *Measurements*: Long-term measurements are needed for a better understanding of seasonal
974 and annual variability, especially the effects of the long-range transport of African dust and
975 biomass-burning emissions. Measurements with improved geographic coverage are needed
976 from the eastern edge of the Basin into central parts to quantify gradients in African and
977 Atlantic aerosol particles and thereby to understand better their influence.

978 3.4. Hygroscopicity

979 The hygroscopic properties of submicron Amazonian aerosol particles have been studied
980 both for the wet season during periods of weak out-of-Basin influence [*Zhou et al., 2002*] and for
981 the dry season at times of strong in-Basin biomass burning [*Rissler et al., 2004; 2006*].
982 Irrespective of season and the air mass type, the hygroscopic diameter growth factor measured at
983 90% relative humidity (RH) by use of a tandem differential mobility analyzer is typically 1.05 to
984 1.35, with few exceptions. This finding differentiates Amazonia from rural sites on other
985 continents, for which “highly hygroscopic” particles having growth factors of 1.7 that approach
986 those of inorganic salts are observed for at least a fraction of the particles [*Swietlicki et al.,*
987 2008]. The consistently high organic fraction of submicron particles (cf. §3.3) can explain the
988 absence of highly hygroscopic particles in the Amazon Basin. As the exception, “highly
989 hygroscopic” particles have, however, been observed in the Amazon Basin at least once [*Zhou et*
990 *al., 2002*], plausibly corresponding to the presence of marine particles imported with an Atlantic
991 air mass [*Formenti et al., 2001*]

992 For natural conditions, “moderately hygroscopic” particles dominate the submicron
993 particle population in the Amazon Basin [Zhou *et al.*, 2002; Rissler *et al.* 2004]. Zhou *et al.*
994 [2002] find that the hygroscopic growth factor increases from 1.17 at 35 nm to 1.32 at 264 nm.
995 Ammonium-bisulfate dry-volume fractions ranging from 0.17 at 35 nm to 0.27 at 265 nm, with
996 the balance of the dry-volume fraction corresponding to an insoluble core, can equivalently
997 represent the observed hygroscopic growth. This equivalent representation must not, however, be
998 interpreted as implying that the water-soluble components of the real particles are composed
999 solely of ammonium and bisulfate ions. Water-soluble organic compounds constitute a large
1000 fraction of the particle components (cf. Figures 19 and 20), although many of the substances,
1001 such as the larger dicarboxylic acids, fulvic acids [Svenningsson *et al.*, 2006], or humic-like
1002 substances (HULIS) [Ziese *et al.*, 2008], have low water uptake. There is also the important
1003 possibility of a difference between the WSOC fraction measured for relatively dilute aqueous
1004 solution (i.e., as represented in Figures 19 and 20) and that relevant to the lower water activity of
1005 90% RH at which measurements of hygroscopic growth have been made.

1006 For locations strongly influenced by fresh in-Basin biomass burning, an external mixture
1007 of “moderately hygroscopic” (as described above) and “barely hygroscopic” particles is observed
1008 [Rissler *et al.*, 2006; Vestin *et al.*, 2007]. “Barely hygroscopic” particles have growth factors
1009 from 1.06 at 20 nm to 1.12 at 440 nm, corresponding to inorganic-equivalent dry-volume
1010 fractions of approximately 0.07. The “barely hygroscopic” particles can dominate the number
1011 balance of the external mixture by a factor of 5 to 10, with a larger fraction at smaller particle
1012 sizes. Open-air biomass burning produces particles largely composed of organic components that
1013 have a limited propensity for water uptake.

1014 In an advance compared to the inorganic-equivalent representation for hygroscopicity,
1015 Mircea *et al.* [2005] provide a more comprehensive treatment of chemical hygroscopic closure
1016 that includes a treatment of the organic component. Growth factors observed during LBA-
1017 SMOCC (i.e., moderately to strongly influenced by biomass burning) for particles 420 nm and
1018 smaller were compared with the predictions of a water-uptake model that incorporated the size-

1019 segregated chemical composition. Inorganic components accounted for about 10% of the size-
1020 segregated mass concentrations, and the balance was carbonaceous [Fuzzi *et al.*, 2007]. Water-
1021 soluble organic compounds constituted 50 to 60% of the mass concentrations [Decesari *et al.*,
1022 2006]. The water-uptake model used for the hygroscopic closure simplified the organic
1023 composition by choosing nine model compounds derived from functional group analyses and
1024 other analytical techniques (cf. further description in §3.3). The model, combined with the size-
1025 segregated chemical composition, accurately predicted the measured growth factors for an
1026 assumption of limited solubility of the organic compounds at 90% RH. Closure could not be
1027 obtained for other candidate assumptions, including complete solubility or complete insolubility.

1028 Priorities for progress to better constrain the hygroscopic behavior of Amazonian aerosol
1029 particles include:

- 1030 • Amazonian biomass-burning particles were extensively characterized during the LBA-
1031 SMOCC experiment in 2002. Less is known about natural particles in pristine rain forest
1032 environments, essentially based on 18 days of data from Zhou *et al.* [2002] and 2 days of data
1033 from Rissler *et al.* [2004], and further measurements are therefore highly motivated with a
1034 focus on understanding temporal and geographic variability.
- 1035 • The links between size-segregated chemical composition and hygroscopic behavior, for
1036 instance by implementing new instrumentation such as aerosol mass spectrometers, as well
1037 as between hygroscopic growth and radiative properties, for instance by conducting
1038 experiments using humidity-controlled nephelometers, should be pursued.
- 1039 • The presence of PBA particles should be quantified by utilizing the ability of a tandem
1040 differential mobility analyzer to determine the hygroscopic growth of individual particles.
1041 The intermittent appearance of externally mixed “barely hygroscopic” particles during the
1042 wet-season of Zhou *et al.* [2002] (i.e., a frequency of occurrence of 5-9% in the Aitken mode
1043 and 11-14% in the accumulation mode) might indicate an influence from PBA particle
1044 sources, although anthropogenic pollution was not entirely ruled out as having an influence
1045 during that study.

1046 3.5. Cloud Condensation Nuclei

1047 The concentrations of cloud condensation nuclei (CCN), in the absence of an influence
1048 by in-Basin biomass burning, are 250-300 cm⁻³ for 1% supersaturation [Roberts *et al.*, 2001;
1049 2002; Andreae *et al.*, 2004; Gunthe *et al.*, 2009]. These CCN concentrations are comparable to
1050 the total particle concentration, and they are lower than typically observed for other rural sites
1051 worldwide, a finding which is indicative of strong anthropogenic influence for most other
1052 continental locations worldwide [Andreae, 2007; 2009]. The CCN properties for natural
1053 conditions can be described in large part by an effective CCN hygroscopicity parameter $\kappa = 0.15$
1054 [Gunthe *et al.*, 2009]. Such particles are sufficiently hygroscopic that they activate at
1055 supersaturations of 0.1 to 0.3% that typically occur within clouds [Zhou *et al.*, 2002;
1056 Svenningsson *et al.*, 2006]. For comparison, κ is larger by a factor of two for other continental
1057 locations [Andreae and Rosenfeld, 2008]. The lower κ in Amazonia is consistent with properties
1058 of SOA material reported in laboratory studies [King *et al.*, 2007, 2009; Prenni *et al.*, 2007] and
1059 with the report of Chen *et al.* [2009] for AMAZE-08 that high proportions of secondary organic
1060 matter constitute the components of submicron particles. Supermicron PBA particles present at
1061 relatively low number concentrations can also be important under some circumstances by serving
1062 as “giant” CCN, which activate at supersaturations below 0.1% because of their large diameters
1063 and enhance the collision-coalescence stage of precipitation formation, especially under polluted
1064 conditions [Yin *et al.*, 2000]. PBA particles are also an important source of ice nuclei in the Basin
1065 [Prenni *et al.*, 2009].

1066 In stark contrast to the low CCN concentrations observed for natural conditions, regions
1067 affected by in-Basin biomass burning can have CCN concentrations on the order of 10,000 cm⁻³
1068 or more (Figure 22) [Roberts *et al.*, 2003; Rissler *et al.*, 2004; Vestin *et al.*, 2007]. The
1069 contribution to CCN number concentration arises not only from the increase in particle number
1070 concentration but also by increases in mode diameter and in water-soluble fraction, both of
1071 which further favor CCN activation [Mayol-Bracero *et al.*, 2002; Decesari *et al.*, 2006; Fuzzi *et*
1072 *al.*, 2007]. For the observations of LBA-SMOCC, Mircea *et al.* [2005] show that CCN closure in

1073 the supersaturated regime (i.e., greater than 100% RH) is best achieved by assuming complete
1074 water solubility of organic species at the high water activities of CCN activation.

1075 The difference in CCN concentrations between natural and anthropogenically influenced
1076 conditions has significant consequences on the microphysical properties of clouds, particularly
1077 the average droplet diameter, the maximum in-cloud supersaturation, and the precipitation
1078 dynamics. Microphysical properties are most susceptible to increasing CCN concentrations for
1079 low base concentrations, such as those of natural Amazonia. The input of additional particles
1080 from biomass burning greatly alters the pathways of cloud development. When smoke plumes
1081 spread over large areas, shallow clouds are inhibited, causing a reduction in cloud cover [*Koren*
1082 *et al.*, 2004; 2005]. *Feingold et al.* [2005] and *Jiang and Feingold* [2006] also suggest that the
1083 extinction of radiation by elevated particle concentrations in the middle troposphere reduces the
1084 surface heat flux, thereby stabilizing the boundary layer and further reducing cloud cover.
1085 Furthermore, enhanced CCN concentrations that result from biomass burning reduce the cloud
1086 droplet diameter below the collision-coalescence threshold, an effect which reduces warm-cloud
1087 precipitation [*Andreae et al.*, 2004]. Another potential effect of increased CCN concentrations,
1088 which is to enhance cloud albedo as a result of both the smaller droplet diameter and more
1089 numerous droplets [*Twomey*, 1977], is small or negligible in the Amazon Basin because the
1090 clouds are usually already optically thick [*Platnick and Twomey*, 1994; *Roberts et al.*, 2003].

1091 The range of microphysical regimes observed in the Basin, including blue-ocean, green-
1092 ocean, smoky-, and pyro- clouds, is illustrated by the mass-diameter distributions of liquid water
1093 content (Figure 23) [*Andreae et al.*, 2004]. There is a narrowing of the distributions and a
1094 slowing of their rate of broadening with increasing height for the progressively more particle-
1095 rich regimes from *A* to *D*. For low CCN concentrations (panels *A* and *B*), the droplet distributions
1096 over the ocean and the Amazon Basin grow and broaden in a similar manner as the parcel rises.
1097 In contrast, for the very high CCN concentrations of pyroclouds (panel *D*) that form in the
1098 invigorated updrafts of the smoke plume over an active fire, the droplet distribution stops
1099 growing once the air parcel rises above a critical altitude (e.g., above 2800 m in Figure 23). The

1100 stunted growth is explained by reduced in-cloud supersaturation that inhibits droplet growth and
1101 has the consequence of suppressing precipitation. Pyro-clouds embedded in a smoky atmosphere
1102 also reduce ground heating by blocking sunlight, and they therefore reduce convective vigor and
1103 precipitation. Smoky clouds (panel C) represent an intermediate case compared to green-ocean
1104 and pyro- clouds. In smoky clouds, the altitude for the onset of precipitation, which corresponds
1105 to a modal diameter of the liquid water content that is greater than an approximate threshold of
1106 24 μm , shifts higher compared to natural conditions. The consequence is that rain either does not
1107 occur or occurs from higher altitudes, including more precipitation initiated through the ice
1108 phase. Consequently, there is a vertical redistribution of released heat and, in the case of ice,
1109 invigorated convection and lightning [Rosenfeld *et al.*, 2008].

1110 The need to understand how the CCN activity of organic particles evolves and the
1111 associated implications for direct and indirect radiative forcing have been highlighted in general
1112 reviews [Kanakidou *et al.*, 2005; Fuzzi *et al.*, 2006; McFiggans *et al.*, 2006; Andreae and
1113 Rosenfeld, 2008]. Specific priorities for better understanding and predicting the CCN activity of
1114 Amazonian aerosol particles and their effects on climate include:

- 1115 • *Measurements*: Data sets of CCN activity in the Basin are restricted both in time and space
1116 and can be considered sparse in comparison to the seasonal and spatial heterogeneities of the
1117 Amazon Basin. Most measurements in available data sets were carried out during intensive
1118 campaigns lasting a short number of weeks at a single location [Roberts *et al.*, 2002; Rissler
1119 *et al.*, 2004; Vestin *et al.*, 2007; Gunthe *et al.*, 2009]. Long-term and distributed
1120 measurements are therefore highly motivated so that seasonal and annual cycles can be
1121 related to other particle parameters, such as chemistry and diameter, as well as to different
1122 meteorological conditions.
- 1123 • *Organic components of CCN activity*: Particle CCN activity depends both on the physical
1124 and chemical properties of the particles, especially particle diameter and the water-soluble
1125 fraction [McFiggans *et al.*, 2006]. The understanding of the mechanistic processes of CCN
1126 activation, in particular variability in space and time of the organic chemical speciation and

1127 most importantly in the water-soluble fraction of Amazonian aerosol particles (cf. §3.3.1),
1128 must be improved.

1129 • *Modeling*: Models should be developed that connect the sources of particles and their
1130 components with CCN activity, including the further evolution as cloud droplets. This
1131 approach should be added to cloud-resolving models, such as the Regional Atmosphere
1132 Modeling System (RAMS) or the Weather and Research Forecasting (WRF) model, and
1133 large-eddy simulations should be carried out. Interconnections among vegetation, particle
1134 chemistry, cloud microphysics, and dynamics can thereby be investigated.

1135 **4. Outlook and Future Priorities**

1136 The Amazonian rain forest, having a present area of greater than 5×10^6 km² and
1137 corresponding to 85% of its original extent, is the largest on Earth. The future of Amazonian
1138 aerosol particles is directly connected to the fate of this forest. Before the 1950s, less than 1% of
1139 the area had been deforested. At present, approximately 0.02×10^6 km² is deforested annually,
1140 mostly accomplished by biomass burning. The fate of the forest depends largely on two
1141 interacting factors, including (1) the rates and the types of human development in the region and
1142 (2) the influences of ongoing global climate change [*Laurance and Williamson, 2001; Cardoso*
1143 *et al., 2003; Lewis, 2006; Soares-Filho et al., 2006; Malhi et al., 2008; Nepstad et al., 2008*]. At
1144 present, deforestation is largely illegal and driven by business opportunities for cattle ranchers
1145 and soy farmers. In the near future, several of the ruling governments plan to construct a network
1146 of crisscrossing roads through the forest (Figure 24). Road access, especially when paved, is the
1147 first step of illegal deforestation [*Laurance et al., 2001; Nepstad et al., 2008*], and forest
1148 fragmented by human activities is also more susceptible to drying and subsequent burning,
1149 resulting in positive feedbacks between deforestation and wildfire [*Laurance and Williamson,*
1150 *2001; Cardoso et al., 2003*]. Deforestation by business-as-usual policies is expected to drive
1151 cumulative loss of the forest from 15% in 2003 of the original extent to 40% by 2050 (Figure 25)
1152 [*Soares-Filho et al., 2006*]. Even with good governance, however, ongoing global climate
1153 change might nevertheless induce large-scale drought [e.g., *Marengo et al., 2008*]. According to

1154 some coupled climate-carbon models, droughts might lead to the loss of most forest in the
1155 Amazon Basin and to the conversion to savanna grassland or desert, regardless of activities
1156 within political Brazil [Betts *et al.*, 2004; Saleska *et al.*, 2007; Cox *et al.*, 2008]. Some models
1157 also indicate that a “tipping point” might occur for a fractional forest cover that falls below a
1158 critical value, meaning that deforestation might serve as a feedback that decreases rainfall and
1159 thus accelerates deforestation and further reductions in rainfall (i.e., at present total
1160 evapotranspiration approaches 50% of total precipitation), ultimately transforming the region
1161 from a rain forest biome to a savanna [Silva Dias *et al.*, 2002; Oyama and Nobre, 2003; Nobre *et al.*,
1162 2004]. A possible fate of large-scale forest loss has paradigm-shifting implications for the
1163 concentrations and the properties of particles as well as for meteorology, cloud type, and rainfall
1164 intensity.

1165 According to projections by Streets [2007] of the future of the Amazon Basin, emissions
1166 of both sulfur dioxide and organic carbon can be expected to grow by 2050 because of high
1167 population and economic growth and limited environmental protection. Heald *et al.* [2008]
1168 project that a 20% mass increase in the production of particle components resulting from BVOC
1169 oxidation can be expected for a business-as-usual scenario, mostly because of changes in climate
1170 and emissions over South America by 2100. Nevertheless, this estimate includes uncertain
1171 changes in BVOC emissions and land-use patterns. Deforestation, for example, might reduce
1172 BVOC emissions and therefore reduce some of the increases projected by Heald *et al.* Another
1173 possibility is that future anthropogenic pollution, in addition to the direct release of primary
1174 particles, might enhance oxidant levels and thereby might significantly alter prevailing BVOC
1175 oxidation pathways, including the fraction of low-volatility products formed (i.e., those that go
1176 on to compose particle components). Changes in land cover might also influence the abundance
1177 and the properties of PBA particles and components. The CCN concentrations in the Basin might
1178 increase year-round, driven both by higher anthropogenic emissions of primary particles and by
1179 greater concentrations of sulfuric acid that might increase the rate of new particle formation.
1180 These many possible changes in the climatology of Amazonian aerosol particles might in turn

1181 significantly influence aerosol-cloud-climate interactions in the Basin, providing a feedback to
1182 rainfall, vegetation, and climate change [Barth *et al.*, 2005; Andreae and Rosenfeld, 2008].
1183 Davidson and Artaxo [2004] review the links in the Amazon Basin between biology and
1184 atmospheric properties.

1185 Potentially of particular importance presently and in the future is the export of particles
1186 from the Amazon Basin to other parts of the world following deep convection [Andreae *et al.*,
1187 2001], especially during the dry season when biomass-burning particle concentrations can be
1188 very high. Because the ITCZ cuts across the Basin and the general circulation involves the
1189 meridional transport of air towards it from both north and south, a substantial fraction of the
1190 Amazonian particles experiences deep convection. Whatever does not get scavenged in that
1191 process can be injected into the middle and upper troposphere, and some particles can also enter
1192 the stratosphere via the Brewer-Dobson circulation. These particles can be transported over long
1193 distances between continents. Transported microorganisms provide a clear demonstration of
1194 these processes [Griffin *et al.*, 2006; Griffin, 2008]. The Amazon Basin may thus represent a
1195 significant global source both of particles and of gaseous precursors to secondary particle
1196 components, and these contributions may have a strong influence on the pristine, low
1197 concentrations that are typically found in the upper troposphere and the stratosphere.

1198 4.1. Priorities for Improved Models

1199 Few modeling studies have focused on aerosol particles in the Amazon Basin. In
1200 particular, very little effort has been invested in understanding particle sources for natural
1201 conditions when concentrations are low. Regional models have been used, however, to
1202 characterize the importance of biomass burning to particle number and mass concentrations as
1203 well as to climate [Freitas *et al.*, 2005; Liu, 2005; Martins and Pereira, 2006]. An
1204 intercomparison of global models showed that model skill over Amazonia for the annual-average
1205 aerosol optical thickness misleadingly appeared to be good because of compensation by an
1206 underestimate of optical thickness for regions influenced by biomass burning and an
1207 overestimate for periods during which natural conditions prevail [Kinne *et al.*, 2003]. The

1208 overestimate for natural conditions was puzzling because the global models did not include
1209 emissions of PBA particles and had very rudimentary descriptions, if any, of the production of
1210 particle-phase BVOC oxidation products. The bias for natural conditions must therefore be
1211 attributed to one or more of an overestimate of out-of-Basin particles into the region, an
1212 underestimate of their in-Basin deposition rates, or a poor characterization of the optical
1213 properties of Amazonian particles. *Kanakidou et al.* [2005] summarized the uncertainties and
1214 challenges related to global climate model simulations of organic aerosol particles. There is a
1215 critical need to validate these model estimates with observations over the Amazon Basin.

1216 High priority research needs for improving the regional modeling of Amazonian aerosol
1217 particles include:

- 1218 • Development of model schemes for emissions of PBA particles in the Amazon Basin. These
1219 schemes are entirely absent in state-of-the-art chemical transport models.
- 1220 • Investigation and implementation of models at the scale necessary to capture how vegetative
1221 heterogeneity within the rain forest canopy affects BVOC and PBA particle emissions.
- 1222 • Inclusion in models at the level of compounds or families of all BVOC emissions
1223 contributing to the secondary components of particles. Amazonian vegetation might be a
1224 more or less efficient emitter of specific compounds compared to other locations for which
1225 standard emissions in models have been calibrated.
- 1226 • Incorporation in models of new BVOC chemistry, such as the reactions of isoprene in the
1227 chemical regimes prevailing in the Amazon Basin (e.g., pristine low NO_x) and more
1228 generally of organic peroxy radicals.
- 1229 • Attention in models, validated by measurements, of how efficiently particles over the
1230 Amazon Basin are removed by precipitation and how this sink term is affected by the
1231 processing and alteration of particles during their residence in the atmosphere.

1232 4.2. Priorities for Improved Measurements

1233 There have been many technological advances in the past 10 years for the
1234 characterization of aerosol particles globally, yet many of the new instruments have yet to be

1235 deployed in the Amazon Basin. The logistical difficulties there have constrained measurements
1236 temporally, spatially, and technically to levels insufficient for fully accurate descriptions of
1237 Amazonian aerosol particles and the processes affecting them. These difficulties
1238 notwithstanding, new instruments, defined in the context of Amazonian aerosol particles both as
1239 truly new instruments in the broad scientific community and as more familiar instruments that
1240 have never been deployed before for studies in the Basin, should be used to obtain more precise
1241 and accurate measurements of key properties of Amazonian particles. The instruments should be
1242 deployed to provide complementary information on complex properties, especially related to
1243 particle chemical composition. Chemical information can be employed, in conjunction with
1244 models, to understand mechanisms of particle formation and subsequent aging processes. In
1245 addition, more creative applications should be made of realtime displays and airborne remote
1246 sensors for better placement of aircraft, particularly when sampling clouds.

1247 The particle properties that are the most uncertain and thus limit our ability to assess their
1248 environmental effects include (1) the molecular composition of the organic component as
1249 quantitative tracers of sources and age, (2) the hygroscopicity and mixing states as affected by
1250 atmospheric processing, (3) the activity as cloud and ice nuclei, and (4) the optical activity (i.e.,
1251 extinction, absorption, single scattering albedo, and asymmetry factor). State-of-the-art
1252 measurement systems have the potential to significantly reduce the uncertainties surrounding
1253 these properties. For example, improvements in measuring the chemistry of aerosol particles
1254 with instruments like aerosol mass spectrometers [*Prather et al.*, 1994; *Jayne et al.*, 2000] have
1255 provided a detailed look into the chemistry of particles, yet an aerosol mass spectrometer went to
1256 Amazonia for the first time only recently as part of AMAZE-08 [*Chen et al.*, 2009]. Similarly,
1257 single- and multi- column continuous flow CCN counters, including instruments with size-
1258 selective inlets, have opened new possibilities for measurements of particle hygroscopicity
1259 [*Roberts and Nenes*, 2005], and these state-of-the-art instruments also were deployed in
1260 Amazonia for the first time in 2008 as part of AMAZE-08. There are many other instruments
1261 that are currently under development or that have only just been made operational. For example,

1262 the single-particle soot photometer is the first instrument to measure the mass in single, light-
1263 absorbing carbon particles, and its use in studying biomass-burning particles could improve the
1264 connections between the particle chemistry and the particle radiation field [*Baumgardner et al.*,
1265 2004; *Schwarz et al.*, 2006; *Moteki et al.*, 2007]. When an aircraft is available, airborne lidar and
1266 radar with realtime displays could be used to locate optimum areas for cloud penetration to
1267 understand and quantify the effects of cloud processing on particles.

1268 The use of measurements to understand the evolution of Amazonian aerosol particles,
1269 their interactions with clouds and radiation, and their impacts on climate must ultimately be
1270 facilitated by global climate models coupled with chemical transport models. These models,
1271 however, must be based on particle properties and processes defined by a combination of
1272 laboratory and ambient measurements, and significant uncertainties exist in the treatments
1273 presently employed for Amazonian aerosol particles (cf. §4.1), traceable in part to an insufficient
1274 set of measurements. As an example of how the uncertainty in measurements propagates into
1275 models, the aerosol optical thickness predicted by several different global models varies by more
1276 than 50% in the Basin [*Kinne et al.*, 2003; 2006]. These intermodel differences are primarily
1277 attributable to uncertainties in the parameterization of particle composition and residence time
1278 [*Textor et al.*, 2006]. Reductions in these uncertainties requires knowledge, to be gained through
1279 well-designed measurement programs, that focus on closure studies, where the closure to be
1280 achieved is between the predicted and measured particle properties, particularly their
1281 hygroscopic, chemical, microphysical, and optical properties. Another important closure, though
1282 less precisely defined, is that of particle residence time, meaning an evaluation of sources and
1283 sinks.

1284 As one example, obtaining closure for CCN activation requires measuring the chemical
1285 composition of particles as a function of size and accurately predicting the number-diameter
1286 distribution of CCN as a function of supersaturation that is measured with in situ
1287 instrumentation. *Gunthe et al.* [2009] provide one example for studies in the Amazon Basin. The
1288 results of this type of closure should be incorporated into modules of cloud microphysics to

1289 improve how the aerosol indirect effect is forecast. As a second example, radiation closure
1290 requires ground-based and satellite measurements of optical thickness, at multiple wavelengths,
1291 to compare with the optical thickness derived from measurements of the chemical and optical
1292 properties of particles over a range of altitudes and geographic locations.

1293 The ideal field program to implement the above closure studies would span several dry
1294 and wet seasons in the Basin, would require in situ and remote sensing measurements from
1295 multiple ground-based and airborne platforms, and would be complemented by satellite
1296 observations from which particle and trace gas properties are derived. Autonomous, unmanned
1297 aerial vehicles have recently been used to make measurements of particles over the Indian Ocean
1298 [*Corrigan et al.*, 2008] and could be extremely useful in the Basin. Establishment of a tall-tower
1299 atmospheric observatory could greatly facilitate long-term, high-quality measurements of particle
1300 properties, gaseous tracers, and meteorological variables. A tall tower could additionally provide
1301 the opportunity to measure vertical profiles through the atmosphere over the forest.

1302 *Concluding remarks.* The goal of this review was to provide a synthesis of aerosol
1303 research in the Amazon Basin, most of which has been published during the last two decades. By
1304 integrating the information that has been published by a considerable number of authors in a
1305 variety of journals over a long span of time, we sought to provide a comprehensive picture about
1306 what is known about Amazonian aerosol particles and to bring together the various aspects that
1307 are now scattered throughout the literature. We intend that this effort will facilitate an
1308 understanding of the current state of knowledge on Amazonian aerosol particles specifically and
1309 tropical continental aerosol particles in general and will thereby enhance future research in this
1310 area. This review barely touched on the transformations of particles by interactions between
1311 particle- and gas-phase species (i.e., condensation, evaporation, and reactions), reactions within
1312 the particles, and cloud processing of particles. These processes, though undoubtedly important,
1313 have hardly been researched in the Amazon Basin. We therefore identify future studies designed
1314 toward these ends as being urgently needed.

Acknowledgments.

The authors of this article were part of *The International Workshop: Aerosol particles in the Amazon - Changes and their Consequences from Past and Future Human Activities*, which took place February 18 - 22, 2008, at ARIAÚ Amazon Towers, north of Manaus, Brazil. Special thanks go to C.H. Martin. Support was received from the U.S. National Science Foundation (NSF) (OISE-0651836), the Brazil LBA Millennium Institute, the Harvard Brazil Studies Program of the David Rockefeller Center of Latin American Studies, the Atmospheric Composition Change: The European Network of Excellence (ACCENT), the Max Planck Society, and the Brazilian Large-Scale Biosphere-Atmosphere (LBA) Experiment. STM also acknowledges support from a Humboldt Research Fellowship: Summer Research Fellowship for U.S. Scientists and Scholars and a Visiting Researcher Award, State of São Paulo Research Foundation (FAPESP), Brazil.

References Cited

- Allen, A. G., and A. H. Miguel, Biomass burning in the Amazon - Characterization of the ionic component of aerosols generated from flaming and smoldering rain-forest and savanna, *Environ. Sci. Technol.*, *29*, 486-493, 1995.
- Andreae, M. O., and T. W. Andreae, The cycle of biogenic sulfur-compounds over the Amazon Basin. 1. Dry season, *J. Geophys. Res.*, *93*, 1487-1497, 1988.
- Andreae, M. O., E. V. Browell, M. Garstang, G. L. Gregory, R. C. Harriss, G. F. Hill, D. J. Jacob, M. C. Pereira, G. W. Sachse, A. W. Setzer, P. L. S. Dias, R. W. Talbot, A. L. Torres, and S. C. Wofsy, Biomass-burning emissions and associated haze layers over Amazonia, *J. Geophys. Res.*, *93*, 1509-1527, 1988.
- Andreae, M. O., H. Berresheim, H. Bingemer, D. J. Jacob, B. L. Lewis, S. M. Li, and R. W. Talbot, The atmospheric sulfur cycle over the Amazon Basin. 2. Wet season, *J. Geophys. Res.*, *95*, 16813-16824, 1990a.
- Andreae, M. O., R. W. Talbot, H. Berresheim, and K. M. Beecher, Precipitation chemistry in Central Amazonia, *J. Geophys. Res.*, *95*, 16987-16999, 1990b.
- Andreae, M. O. (1993), The influence of tropical biomass burning on climate and the atmospheric environment, in *Biogeochemistry of Global Change: Radiatively Active Trace Gases*, edited by R. S. Oremland, pp. 113-150, Chapman & Hall, New York..
- Andreae, M. O., B. E. Anderson, D. R. Blake, J. D. Bradshaw, J. E. Collins, G. L. Gregory, G. W. Sachse, and M. C. Shipham, Influence of plumes from biomass burning on atmospheric chemistry over the equatorial Atlantic during CITE-3, *J. Geophys. Res.*, *99*, 12,793-712,808, 1994.
- Andreae, M. O., and P. J. Crutzen, Atmospheric aerosols: Biogeochemical sources and role in atmospheric chemistry, *Science*, *276*, 1052-1058, 1997.
- Andreae, M. O., T. W. Andreae, H. Annegarn, J. Beer, H. Cachier, P. le Canut, W. Elbert, W. Maenhaut, I. Salma, F. G. Wienhold, and T. Zenker, Airborne studies of aerosol emissions from savanna fires in southern Africa: 2. Aerosol chemical composition, *J. Geophys. Res.*, *103*, 32119-32128, 1998.
- Andreae, M. O., P. Artaxo, H. Fischer, S. R. Freitas, J. M. Gregoire, A. Hansel, P. Hoor, R. Kormann, R. Krejci, L. Lange, J. Lelieveld, W. Lindinger, K. Longo, W. Peters, M. de Reus, B. Scheeren, M. Dias, J. Strom, P. F. J. van Velthoven, and J. Williams, Transport of biomass burning smoke to the upper troposphere by deep convection in the equatorial region, *Geophys. Res. Lett.*, *28*, 951-954, 2001.
- Andreae, M. O., P. Artaxo, C. Brandao, F. E. Carswell, P. Ciccioli, A. L. da Costa, A. D. Culf, J. L. Esteves, J. H. C. Gash, J. Grace, P. Kabat, J. Lelieveld, Y. Malhi, A. O. Manzi, F. X. Meixner, A. D. Nobre, C. Nobre, M. Ruivo, M. A. Silva-Dias, P. Stefani, R. Valentini, J. von Jouanne, and M. J. Waterloo, Biogeochemical cycling of carbon, water, energy, trace gases, and aerosols in Amazonia: The LBA-EUSTACH experiments, *J. Geophys. Res.*, *107*, 8066, 2002.
- Andreae, M. O., D. Rosenfeld, P. Artaxo, A. A. Costa, G. P. Frank, K. M. Longo, and M. A. F. Silva-Dias, Smoking rain clouds over the Amazon, *Science*, *303*, 1337-1342, 2004.

- Andreae, M. O., and A. Gelencser, Black carbon or brown carbon? The nature of light-absorbing carbonaceous aerosols, *Atmos. Chem. Phys.*, *6*, 3131-3148, 2006.
- Andreae, M. O., Aerosols before pollution, *Science*, *315*, 50-51, 2007.
- Andreae, M. O., Correlation between cloud condensation nuclei concentration and aerosol optical thickness in remote and polluted regions, *Atmos. Chem. Phys.*, *9*, 543-556, 2009.
- Andreae, M. O., and D. Rosenfeld, Aerosol-cloud-precipitation interactions. Part 1. The nature and sources of cloud-active aerosols, *Earth Sci. Rev.*, *89*, 13-41, 2008.
- Artaxo, P., H. Storms, F. Bruynseels, R. Van Grieken, and W. Maenhaut, Composition and sources of aerosols from the Amazon Basin, *J. Geophys. Res.*, *93*, 1605-1615, 1988.
- Artaxo, P., W. Maenhaut, H. Storms, and R. Van Grieken, Aerosol characteristics and sources for the Amazon Basin during the wet season, *J. Geophys. Res.*, *95*, 16971-16985, 1990.
- Artaxo, P., and H. C. Hansson, Size distribution of biogenic aerosol-particles from the Amazon Basin, *Atmos. Environ.*, *29*, 393-402, 1995.
- Artaxo, P., E. T. Fernandes, J. V. Martins, M. A. Yamasoe, P. V. Hobbs, W. Maenhaut, K. M. Longo, and A. Castanho, Large-scale aerosol source apportionment in Amazonia, *J. Geophys. Res.*, *103*, 31837-31847, 1998.
- Artaxo, P., J. V. Martins, M. A. Yamasoe, A. S. Procopio, T. M. Pauliquevis, M. O. Andreae, P. Guyon, L. V. Gatti, and A. M. C. Leal, Physical and chemical properties of aerosols in the wet and dry seasons in Rondonia, Amazonia, *J. Geophys. Res.*, *107*, 8081, 2002.
- Avissar, R., P. L. S. Dias, M. Dias, and C. Nobre, The Large-Scale Biosphere-atmosphere Experiment in Amazonia (LBA): Insights and future research needs, *J. Geophys. Res.*, *107*, 8086, 2002.
- Barkley, M. P., P. I. Palmer, U. Kuhn, J. Kesselmeier, K. Chance, T. P. Kurosu, R. V. Martin, D. Helmig, and A. Guenther, Net ecosystem fluxes of isoprene over tropical South America inferred from GOME observations of HCHO columns, *J. Geophys. Res.*, *113*, D20304, 2008.
- Barth, M., J. P. McFadden, J. L. Sun, C. Wiedinmyer, P. Chuang, B. Collins, R. Griffin, M. Hannigan, T. Karl, S. W. Kim, S. Lasher-Trapp, S. Levis, M. Litvak, N. Mahowald, K. Moore, S. Nandi, E. Nemitz, A. Nenes, M. Potosnak, T. M. Raymond, J. Smith, C. Still, and C. Stroud, Coupling between land ecosystems and the atmospheric hydrologic cycle through biogenic aerosol pathways, *Bull. Amer. Meteorol. Soc.*, *86*, 1738-1742, 2005.
- Barregard, L., G. Sallsten, P. Gustafson, L. Andersson, L. Johansson, S. Basu, and L. Stigendal, Experimental exposure to wood-smoke particles in healthy humans: Effects on markers of inflammation, coagulation, and lipid peroxidation, *Inhal. Toxic.*, *18*, 845-853, 2006.
- Baumgardner, D., G. Kok, and G. Raga, Warming of the Arctic lower stratosphere by light absorbing particles, *Geophys. Res. Lett.*, *31*, L06117, 2004.
- Betts, R. A., P. M. Cox, M. Collins, P. P. Harris, C. Huntingford, and C. D. Jones, The role of ecosystem-atmosphere interactions in simulated Amazonian precipitation decrease and forest dieback under global climate warming, *Theo. Appl. Clim.*, *78*, 157-175, 2004.
- Blando, J. D., and B. J. Turpin, Secondary organic aerosol formation in cloud and fog droplets: a literature evaluation of plausibility, *Atmos. Environ.*, *34*, 1623-1632, 2000.
- Browell, E. V., G. L. Gregory, R. C. Harriss, and V. Kirchhoff, Ozone and aerosol distributions over the Amazon Basin during the wet season, *J. Geophys. Res.*, *95*, 16887-16901, 1990.

- Cardoso, M. F., G. C. Hurtt, B. Moore, C. A. Nobre, and E. M. Prins, Projecting future fire activity in Amazonia, *Glob. Change Biol.*, *9*, 656-669, 2003.
- Carlton, A. G., B. J. Turpin, H.-J. Lim, K. E. Altieri, and S. P. Seitzinger, Link between isoprene and secondary organic aerosol (SOA): Pyruvic acid oxidation yields low volatility organic acids in clouds, *Geophys. Res. Lett.*, *33*, L06822, 2006.
- Ceburnis, D., C. D. O'Dowd, G. S. Jennings, M. C. Facchini, L. Emblico, S. Decesari, S. Fuzzi, and J. Sakalys, Marine aerosol chemistry gradients: Elucidating primary and secondary processes and fluxes, *Geophys. Res. Lett.*, *35*, L07804, 2008.
- Chand, D., P. Guyon, P. Artaxo, O. Schmid, G. P. Frank, L. V. Rizzo, O. L. Mayol-Bracero, L. V. Gatti, and M. O. Andreae, Optical and physical properties of aerosols in the boundary layer and free troposphere over the Amazon Basin during the biomass burning season, *Atmos. Chem. Phys.*, *6*, 2911-2925, 2006.
- Chen, Q., Farmer, D. K., Schneider, J., Zorn, S. R., Heald, C. L., Karl, T. G., Guenther, A., Allan, J. D., Robinson, N., Coe, H., Kimmel, J. R., Pauliquevis, T., Borrmann, S., Pöschl, U., Andreae, M. O., Artaxo, P., Jimenez, J. L., and Martin S. T.: Mass spectral characterization of submicron biogenic organic particles in the Amazon basin, *Geophys. Res. Lett.*, **submitted**, 2009.
- Chung, S. H., and J. H. Seinfeld, Global distribution and climate forcing of carbonaceous aerosols, *J. Geophys. Res.*, *107*, 4407, 2002.
- Ciccioli, P., E. Brancaleoni, M. Frattoni, V. Di Palo, R. Valentini, G. Tirone, G. Seufert, N. Bertin, U. Hansen, O. Csiky, R. Lenz, and M. Sharma, Emission of reactive terpene compounds from orange orchards and their removal by within-canopy processes, *J. Geophys. Res.*, *104*, 8077-8094, 1999.
- Claeys, M., B. Graham, G. Vas, W. Wang, R. Vermeylen, V. Pashynska, J. Cafmeyer, P. Guyon, M. O. Andreae, P. Artaxo, and W. Maenhaut, Formation of secondary organic aerosols through photooxidation of isoprene, *Science*, *303*, 1173-1176, 2004.
- Clarke, A. D., J. L. Varner, F. Eisele, R. L. Mauldin, D. Tanner, and M. Litchy, Particle production in the remote marine atmosphere: Cloud outflow and subsidence during ACE 1, *J. Geophys. Res.*, *103*, 16397-16409, 1998.
- Clarke, A. D., V. N. Kapustin, F. L. Eisele, R. J. Weber, and P. H. McMurry, Particle production near marine clouds: Sulfuric acid and predictions from classical binary nucleation, *Geophys. Res. Lett.*, *26*, 2425-2428, 1999.
- Corrigan, C. E., G. C. Roberts, M. V. Ramana, D. Kim, and V. Ramanathan, Capturing vertical profiles of aerosols and black carbon over the Indian Ocean using autonomous unmanned aerial vehicles, *Atmos. Chem. Phys.*, *8*, 737-747, 2008.
- Cox, P. M., P. P. Harris, C. Huntingford, R. A. Betts, M. Collins, C. D. Jones, T. E. Jupp, J. A. Marengo, and C. A. Nobre, Increasing risk of Amazonian drought due to decreasing aerosol pollution, *Nature*, *453*, 212-217, 2008.
- Davidson, E. A., and P. Artaxo, Globally significant changes in biological processes of the Amazon Basin: results of the Large-scale Biosphere-Atmosphere Experiment, *Glob. Change Biol.*, *10*, 519-529, 2004.
- Decesari, S., S. Fuzzi, M. C. Facchini, M. Mircea, L. Emblico, F. Cavalli, W. Maenhaut, X. Chi, G. Schkolnik, A. Falkovich, Y. Rudich, M. Claeys, V. Pashynska, G. Vas, I. Kourtchev, R. Vermeylen, A. Hoffer, M. O. Andreae, E. Tagliavini, F. Moretti, and P. Artaxo,

- Characterization of the organic composition of aerosols from Rondonia, Brazil, during the LBA-SMOCC 2002 experiment and its representation through model compounds, *Atmos. Chem. Phys.*, *6*, 375-402, 2006.
- Despres, V. R., J. F. Nowoisky, M. Klose, R. Conrad, M. O. Andreae, and U. Pöschl, Characterization of primary biogenic aerosol particles in urban, rural, and high-alpine air by DNA sequence and restriction fragment analysis of ribosomal RNA genes, *Biogeosci.*, *4*, 1127-1141, 2007.
- Echalar, F., P. Artaxo, J. V. Martins, M. Yamasoe, F. Gerab, W. Maenhaut, and B. Holben, Long-term monitoring of atmospheric aerosols in the Amazon Basin: Source identification and apportionment, *J. Geophys. Res.*, *103*, 31849-31864, 1998.
- Ekman, A. M. L., R. Krejci, A. Engstrom, J. Strom, M. de Reus, J. Williams, and M. O. Andreae, Do organics contribute to small particle formation in the Amazonian upper troposphere?, *Geophys. Res. Lett.*, *35*, L17810, 2008.
- Elbert, W., P. E. Taylor, M. O. Andreae, and U. Pöschl, Contribution of fungi to primary biogenic aerosols in the atmosphere: wet and dry discharged spores, carbohydrates, and inorganic ions, *Atmos. Chem. Phys.*, *7*, 4569-4588, 2007.
- Falkovich, A. H., E. R. Graber, G. Schkolnik, Y. Rudich, W. Maenhaut, and P. Artaxo, Low molecular weight organic acids in aerosol particles from Rondonia, Brazil, during the biomass-burning, transition and wet periods, *Atmos. Chem. Phys.*, *5*, 781-797, 2005.
- Feingold, G., H. L. Jiang, and J. Y. Harrington, On smoke suppression of clouds in Amazonia, *Geophys. Res. Lett.*, *32*, L02804, 2005.
- Ferek, R. J., J. S. Reid, P. V. Hobbs, D. R. Blake, and C. Liousse, Emission factors of hydrocarbons, halocarbons, trace gases and particles from biomass burning in Brazil, *J. Geophys. Res.*, *103*, 32107-32118, 1998.
- Fiedler, V., M. Dal Maso, M. Boy, H. Aufmhoff, J. Hoffmann, T. Schuck, W. Birmili, M. Hanke, J. Uecker, F. Arnold, and M. Kulmala, The contribution of sulphuric acid to atmospheric particle formation and growth: a comparison between boundary layers in Northern and Central Europe, *Atmos. Chem. Phys.*, *5*, 1773-1785, 2005.
- Fisch, G., J. Tota, L. A. T. Machado, M. Dias, R. F. D. Lyra, C. A. Nobre, A. J. Dolman, and J. H. C. Gash, The convective boundary layer over pasture and forest in Amazonia, *Theo. Appl. Clim.*, *78*, 47-59, 2004.
- Formenti, P., M. O. Andreae, L. Lange, G. Roberts, J. Cafmeyer, I. Rajta, W. Maenhaut, B. N. Holben, P. Artaxo, and J. Lelieveld, Saharan dust in Brazil and Suriname during the Large-Scale Biosphere-Atmosphere Experiment in Amazonia (LBA) - Cooperative LBA Regional Experiment (CLAIRE) in March 1998, *J. Geophys. Res.*, *106*, 14919-14934, 2001.
- Freitas, S. R., M. A. F. S. Dias, and P. L. S. Dias, Modeling the convective transport of trace gases by deep and moist convection, *Hyb. Meth. Eng.*, *2*, 317-330, 2000.
- Freitas, S. R., K. M. Longo, M. Diasb, P. L. S. Diasb, R. Chatfield, E. Prins, P. Artaxo, G. A. Grell, and F. S. Recuero, Monitoring the transport of biomass-burning emissions in South America, *Env. Fluid Mech.*, *5*, 135-167, 2005.
- Fuzzi, S., M. O. Andreae, B. J. Huebert, M. Kulmala, T. C. Bond, M. Boy, S. J. Doherty, A. Guenther, M. Kanakidou, K. Kawamura, V. M. Kerminen, U. Lohmann, L. M. Russell, and U. Pöschl, Critical assessment of the current state of scientific knowledge, terminology, and

- research needs concerning the role of organic aerosols in the atmosphere, climate, and global change, *Atmos. Chem. Phys.*, *6*, 2017-2038, 2006.
- Fuzzi, S., S. Decesari, M. C. Facchini, F. Cavalli, L. Emblico, M. Mircea, M. O. Andreae, I. Trebs, A. Hoffer, P. Guyon, P. Artaxo, L. V. Rizzo, L. L. Lara, T. Pauliquevis, W. Maenhaut, N. Raes, X. G. Chi, O. L. Mayol-Bracero, L. L. Soto-Garcia, M. Claeys, I. Kourtchev, J. Rissler, E. Swietlicki, E. Tagliavini, G. Schkolnik, A. H. Falkovich, Y. Rudich, G. Fisch, and L. V. Gatti, Overview of the inorganic and organic composition of size-segregated aerosol in Rondonia, Brazil, from the biomass-burning period to the onset of the wet season, *J. Geophys. Res.*, *112*, D01201, 2007.
- Garreaud, R. D., and J. M. Wallace, Summertime incursions of midlatitude air into subtropical and tropical South America, *Mon. Wea. Rev.*, *126*, 2713-2733, 1998.
- Gerab, F., P. Artaxo, R. Gillett, and G. Ayers, PIXE, PIGE and ion chromatography of aerosol particles from northeast Amazon Basin, *Nuc. Inst. Meth. Phys. Res.*, *137*, 955-960, 1998.
- Giglio, L., I. Csiszar, and C. O. Justice, Global distribution and seasonality of active fires as observed with the Terra and Aqua MODIS sensors, *J. Geophys. Res.*, *111*, G02016, 2006.
- Gilbert, G. S., Nocturnal fungi: Airborne spores in the canopy and understory of a tropical rain forest, *Biotropica*, *37*, 462-464, 2005.
- Goldstein, A. H., and I. E. Galbally, Known and unexplored organic constituents in the earth's atmosphere, *Environ. Sci. Technol.*, *41*, 1514-1521, 2007.
- Graham, B., O. L. Mayol-Bracero, P. Guyon, G. C. Roberts, S. Decesari, M. C. Facchini, P. Artaxo, W. Maenhaut, P. Koll, and M. O. Andreae, Water-soluble organic compounds in biomass burning aerosols over Amazonia - 1. Characterization by NMR and GC-MS, *J. Geophys. Res.*, *107*, 8047, 2002.
- Graham, B., P. Guyon, W. Maenhaut, P. E. Taylor, M. Ebert, S. Matthias-Maser, O. L. Mayol-Bracero, R. H. M. Godoi, P. Artaxo, F. X. Meixner, M. A. L. Moura, C. Rocha, R. Van Grieken, M. M. Glovsky, R. C. Flagan, and M. O. Andreae, Composition and diurnal variability of the natural Amazonian aerosol, *J. Geophys. Res.*, *108*, 4765, 2003a.
- Graham, B., P. Guyon, P. E. Taylor, P. Artaxo, W. Maenhaut, M. M. Glovsky, R. C. Flagan, and M. O. Andreae, Organic compounds present in the natural Amazonian aerosol: Characterization by gas chromatography-mass spectrometry, *J. Geophys. Res.*, *108*, 4766, 2003b.
- Griffin, D. W., D. L. Westphal, and M. A. Gray, Airborne microorganisms in the African desert dust corridor over the mid-Atlantic ridge, Ocean Drilling Program, Leg 209, *Aerobiologia*, *22*, 211-226, 2006.
- Griffin, D. W., Non-spore forming eubacteria isolated at an altitude of 20,000 m in Earth's atmosphere: extended incubation periods needed for culture-based assays, *Aerobiologia*, *24*, 19-25, 2008.
- Guenther, A., C. N. Hewitt, D. Erickson, R. Fall, C. Geron, T. Graedel, P. Harley, L. Klinger, M. Lerdau, W. A. McKay, T. Pierce, B. Scholes, R. Steinbrecher, R. Tallamraju, J. Taylor, and P. Zimmerman, A global model of natural volatile organic compound emissions, *J. Geophys. Res.*, *100*, 8873-8892, 1995.
- Guenther, A., T. Karl, P. Harley, C. Wiedinmyer, P. I. Palmer, and C. Geron, Estimates of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols from Nature), *Atmos. Chem. Phys.*, *6*, 3181-3210, 2006.

- Gunthe, S. S., King, S. M., Rose, D., Chen, Q., Roldin, P., Farmer, D. K., Jimenez, J. L., Artaxo, P., Andreae, M. O., Martin, S. T., and Pöschl, U.: Cloud condensation nuclei in pristine tropical rainforest air of Amazonia: size-resolved measurements and modeling of atmospheric aerosol composition and CCN activity, *submitted*, 2009
- Guyon, P., O. Boucher, B. Graham, J. Beck, O. L. Mayol-Bracero, G. C. Roberts, W. Maenhaut, P. Artaxo, and M. O. Andreae, Refractive index of aerosol particles over the Amazon tropical forest during LBA-EUSTACH 1999, *J. Aerosol Sci.*, *34*, 883-907, 2003a.
- Guyon, P., B. Graham, J. Beck, O. Boucher, E. Gerasopoulos, O. L. Mayol-Bracero, G. C. Roberts, P. Artaxo, and M. O. Andreae, Physical properties and concentration of aerosol particles over the Amazon tropical forest during background and biomass burning conditions, *Atmos. Chem. Phys.*, *3*, 951-967, 2003b.
- Guyon, P., B. Graham, G. C. Roberts, O. L. Mayol-Bracero, W. Maenhaut, P. Artaxo, and M. O. Andreae, In-canopy gradients, composition, sources, and optical properties of aerosol over the Amazon forest, *J. Geophys. Res.*, *108*, 4591, 2003c.
- Guyon, P., G. P. Frank, M. Welling, D. Chand, P. Artaxo, L. Rizzo, G. Nishioka, O. Kolle, M. Dias, L. V. Gatti, A. M. Cordova, and M. O. Andreae, Airborne measurements of trace gas and aerosol particle emissions from biomass burning in Amazonia, *Atmos. Chem. Phys.*, *5*, 2989-3002, 2005.
- Harley, P., P. Vasconcellos, L. Vierling, C. C. D. Pinheiro, J. Greenberg, A. Guenther, L. Klinger, S. S. De Almeida, D. Neill, T. Baker, O. Phillips, and Y. Malhi, Variation in potential for isoprene emissions among Neotropical forest sites, *Glob. Change Biol.*, *10*, 630-650, 2004.
- Harriss, R. C., S. C. Wofsy, M. Garstang, E. V. Browell, L. C. B. Molion, R. J. McNeal, J. M. Hoell, R. J. Bendura, S. M. Beck, R. L. Navarro, J. T. Riley, and R. L. Snell, The Amazon Boundary-Layer Experiment (ABLE-2A) - dry season 1985, *J. Geophys. Res.*, *93*, 1351-1360, 1988.
- Harriss, R. C., M. Garstang, S. C. Wofsy, S. M. Beck, R. J. Bendura, J. R. B. Coelho, J. W. Drewry, J. M. Hoell, P. A. Matson, R. J. McNeal, L. C. B. Molion, R. L. Navarro, V. Rabine, and R. L. Snell, The Amazon Boundary-Layer Experiment - Wet Season 1987, *J. Geophys. Res.*, *95*, 16721-16736, 1990.
- Heald, C. L., D. K. Henze, L. W. Horowitz, J. Feddema, J.-F. Lamarque, A. Guenther, P. G. Hess, F. Vitt, J. H. Seinfeld, A. H. Goldstein, and I. Fung, Predicted change in global secondary organic aerosol concentrations in response to future climate, emissions, and land use change, *J. Geophys. Res.*, *113*, D05211, 2008.
- Henze, D. K., and J. H. Seinfeld, Global secondary organic aerosol from isoprene oxidation, *Geophys. Res. Lett.*, *33*, L09812, 2006.
- Hines, P. J., The invisible bouquet - Introduction, *Science*, *311*, 803-803, 2006.
- Hoffer, A., A. Gelencser, M. Blazso, P. Guyon, P. Artaxo, and M. O. Andreae, Diel and seasonal variations in the chemical composition of biomass burning aerosol, *Atmos. Chem. Phys.*, *6*, 3505-3515, 2006.
- Holzinger, R., A. Lee, K. T. Paw, and A. H. Goldstein, Observations of oxidation products above a forest imply biogenic emissions of very reactive compounds, *Atmos. Chem. Phys.*, *5*, 67-75, 2005.

- Hoppel, W. A., G. M. Frick, and R. E. Larson, Effects of non-precipitating clouds on the aerosol size distribution in the marine boundary layer, *Geophys. Res. Lett.*, *13*, 125-128, 1986.
- Hoppel, W. A., G. M. Frick, J. W. Fitzgerald and R. E. Larson, Marine boundary layer measurements of new particle formation and the effects of nonprecipitating clouds on the aerosol size distribution, *J. Geophys. Res.*, *99*, 14443-14459, 1994.
- Hörrak, U., H. Iher, A. Luts, J. Salm, and H. Tammet, Mobility spectrum of air ions at Tahkuse Observatory, *J. Geophys. Res.*, *99*, 10697-10700, 1994.
- Hoyle, C. R., T. Berntsen, G. Myhre, and I. S. A. Isaksen, Secondary organic aerosol in the global aerosol - chemical transport model Oslo CTM2, *Atmos. Chem. Phys.*, *7*, 5675-5694, 2007.
- Ignotti, E., J. Valente, S. Hacon, K. Longo, and S. Freitas, Proportional mortality due to respiratory diseases (RD) and exposure to PM_{2.5} in Amazon region, *Epidemiology*, *18*, S94-S94, 2007.
- Iida, K., M. R. Stolzenburg, P. H. McMurry, and J. N. Smith, Estimating nanoparticle growth rates from size-dependent charged fractions - Analysis of new particle formation events in Mexico City, *J. Geophys. Res.*, *113*, D05207, 2008.
- IPCC (2007), Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change, edited by S. Solomon, et al., p. 996, Cambridge University Press, Cambridge and New York.
- Jacob, D. J., and S. C. Wofsy, Photochemistry of biogenic emissions over the Amazon forest, *J. Geophys. Res.*, *93*, 1477-1486, 1988.
- Jacob, D. J., and S. C. Wofsy, Budgets of reactive nitrogen, hydrocarbons, and ozone over the Amazon-forest during the wet season, *J. Geophys. Res.*, *95*, 16737-16754, 1990.
- Jacobson, M. C., H.-C. Hansson, K. J. Noone, and R. J. Charlson, Organic atmospheric aerosols: Review and state of the science, *Rev. Geophys.*, *38*, 267-294, 2000.
- Jaenicke, R., Abundance of cellular material and proteins in the atmosphere, *Science*, *308*, 73-73, 2005.
- Jayne, J. T., D. C. Leard, X. F. Zhang, P. Davidovits, K. A. Smith, C. E. Kolb, and D. R. Worsnop, Development of an aerosol mass spectrometer for size and composition analysis of submicron particles, *Aerosol Sci. Technol.*, *33*, 49-70, 2000.
- Jiang, H. L., and G. Feingold, Effect of aerosol on warm convective clouds: Aerosol-cloud-surface flux feedbacks in a new coupled large eddy model, *J. Geophys. Res.*, *111*, D01202, 2006.
- Jones, C., and L. M. V. Carvalho, Active and break phases in the south American Monsoon system, *J. Clim.*, *15*, 905-914, 2002.
- Kalnay, E., M. Kanamitsu, R. Kistler, W. Collins, D. Deaven, L. Gandin, M. Iredell, S. Saha, G. White, J. Woollen, Y. Zhu, M. Chelliah, W. Ebisuzaki, W. Higgins, J. Janowiak, K. C. Mo, C. Ropelewski, J. Wang, A. Leetmaa, R. Reynolds, R. Jenne, and D. Joseph, The NCEP/NCAR 40-year reanalysis project, *Bull. Amer. Meteorol. Soc.*, *77*, 437-471, 1996.
- Kanakidou, M., J. H. Seinfeld, S. N. Pandis, I. Barnes, F. J. Dentener, M. C. Facchini, R. Van Dingenen, B. Ervens, A. Nenes, C. J. Nielsen, E. Swietlicki, J. P. Putaud, Y. Balkanski, S. Fuzzi, J. Horth, G. K. Moortgat, R. Winterhalter, C. E. L. Myhre, K. Tsigaridis, E. Vignati,

- E. G. Stephanou, and J. Wilson, Organic aerosol and global climate modelling: a review, *Atmos. Chem. Phys.*, *5*, 1053-1123, 2005.
- Karl, T., A. Guenther, R. J. Yokelson, J. Greenberg, M. Potosnak, D. R. Blake, and P. Artaxo, The tropical forest and fire emissions experiment: Emission, chemistry, and transport of biogenic volatile organic compounds in the lower atmosphere over Amazonia, *J. Geophys. Res.*, *112*, D18302, 2007.
- Karl, T., A. Guenther, A. Turnipseed, P. Artaxo, and S. Martin, Rapid formation of isoprene photo-oxidation products observed in Amazonia, *Atmos. Chem. Phys.*, submitted, 2009.
- Kaufman, Y. J., P. V. Hobbs, V. Kirchhoff, P. Artaxo, L. A. Remer, B. N. Holben, M. D. King, D. E. Ward, E. M. Prins, K. M. Longo, L. F. Mattos, C. A. Nobre, J. D. Spinhirne, Q. Ji, A. M. Thompson, J. F. Gleason, S. A. Christopher, and S. C. Tsay, Smoke, Clouds, and Radiation - Brazil (SCAR-B) experiment, *J. Geophys. Res.*, *103*, 31783-31808, 1998.
- Keller, M., M. Bustamante, J. Gash, and P. Dias (Eds) (2009), *Amazonia and Global Change*, American Geophysical Union, Washington, DC, **in press**.
- Kerminen, V. M., and M. Kulmala, Analytical formulae connecting the "real" and the "apparent" nucleation rate and the nuclei number concentration for atmospheric nucleation events, *J. Aerosol Sci.*, *33*, 609-622, 2002.
- Kesselmeier, J., U. Kuhn, A. Wolf, M. O. Andreae, P. Ciccioli, E. Brancaleoni, M. Frattoni, A. Guenther, J. Greenberg, P. D. Vasconcellos, T. de Oliva, T. Tavares, and P. Artaxo, Atmospheric volatile organic compounds (VOC) at a remote tropical forest site in central Amazonia, *Atmos. Environ.*, *34*, 4063-4072, 2000.
- Kesselmeier, J., A. Guenther, T. Hoffmann, and J. Warnke, Natural volatile organic compound (VOC) emissions from plants and their roles in oxidant balance and particle formation, in *Amazonia and Global Change*, edited by M. Keller, et al., p. **in press**, 2009.
- King, S. M., T. Rosenoern, J. E. Shilling, Q. Chen, and S. T. Martin, Cloud condensation nucleus activity of secondary organic aerosol particles mixed with sulfate, *Geophys. Res. Lett.*, *34*, L24806, 2007.
- King, S. M., T. Rosenoern, J.E. Shilling, Q. Chen, and S.T. Martin, "Increased cloud activation potential of secondary organic aerosol for atmospheric mass loadings," *Atmos. Chem. Phys.*, *9*, 2959-2972, 2009.
- Kinne, S., U. Lohmann, J. Feichter, M. Schulz, C. Timmreck, S. Ghan, R. Easter, M. Chin, P. Ginoux, T. Takemura, I. Tegen, D. Koch, M. Herzog, J. Penner, G. Pitari, B. Holben, T. Eck, A. Smirnov, O. Dubovik, I. Slutsker, D. Tanre, O. Torres, M. Mishchenko, I. Geogdzhayev, D. A. Chu, and Y. Kaufman, Monthly averages of aerosol properties: A global comparison among models, satellite data, and AERONET ground data, *J. Geophys. Res.*, *108*, 4634, 2003.
- Kinne, S., M. Schulz, C. Textor, S. Guibert, Y. Balkanski, S. E. Bauer, T. Berntsen, T. F. Berglen, O. Boucher, M. Chin, W. Collins, F. Dentener, T. Diehl, R. Easter, J. Feichter, D. Fillmore, S. Ghan, P. Ginoux, S. Gong, A. Grini, J. E. Hendricks, M. Herzog, L. Horowitz, L. Isaksen, T. Iversen, A. Kirkavag, S. Kloster, D. Koch, J. E. Kristjansson, M. Krol, A. Lauer, J. F. Lamarque, G. Lesins, X. Liu, U. Lohmann, V. Montanaro, G. Myhre, J. E. Penner, G. Pitari, S. Reddy, O. Seland, P. Stier, T. Takemura, and X. Tie, An AeroCom initial assessment - optical properties in aerosol component modules of global models, *Atmos. Chem. Phys.*, *6*, 1815-1834, 2006.

- Kodama, Y., Large-scale common features of subtropical precipitation zones (the Baiu frontal zone, the SPCZ, and the SACZ). 1. Characteristics of subtropical frontal zones, *J. Met. Soc. Japan*, *70*, 813-836, 1992.
- Komppula, M., S. L. Sihto, H. Korhonen, H. Lihavainen, V. M. Kerminen, M. Kulmala, and Y. Viisanen, New particle formation in air mass transported between two measurement sites in Northern Finland, *Atmos. Chem. Phys.*, *6*, 2811-2824, 2006.
- Koren, I., Y. J. Kaufman, L. A. Remer, and J. V. Martins, Measurement of the effect of Amazon smoke on inhibition of cloud formation, *Science*, *303*, 1342-1345, 2004.
- Koren, I., Y. J. Kaufman, D. Rosenfeld, L. A. Remer, and Y. Rudich, Aerosol invigoration and restructuring of Atlantic convective clouds, *Geophys. Res. Lett.*, *32*, L14828, 2005.
- Krejci, R., J. Strom, M. de Reus, P. Hoor, J. Williams, H. Fischer, and H. C. Hansson, Evolution of aerosol properties over the rain forest in Surinam, South America, observed from aircraft during the LBA-CLAIRE 98 experiment, *J. Geophys. Res.*, *108*, 4561, 2003.
- Krejci, R., J. Strom, M. de Reus, J. Williams, H. Fischer, M. O. Andreae, and H. C. Hansson, Spatial and temporal distribution of atmospheric aerosols in the lowermost troposphere over the Amazonian tropical rainforest, *Atmos. Chem. Phys.*, *5*, 1527-1543, 2005.
- Kuang, C., P. H. McMurry, A. V. McCormick, and F. L. Eisele, Dependence of nucleation rates on sulfuric acid vapor concentrations in diverse atmospheric locations, *J. Geophys. Res.*, *113*, D10209, 2008.
- Kuhn, U., M. O. Andreae, C. Ammann, A. C. Araujo, E. Brancaleoni, P. Ciccioli, T. Dindorf, M. Frattoni, L. V. Gatti, L. Ganzeveld, B. Kruijt, J. Lelieveld, J. Lloyd, F. X. Meixner, A. D. Nobre, U. Poschl, C. Spirig, P. Stefani, A. Thielmann, R. Valentini, and J. Kesselmeier, Isoprene and monoterpene fluxes from Central Amazonian rainforest inferred from tower-based and airborne measurements, and implications on the atmospheric chemistry and the local carbon budget, *Atmos. Chem. Phys.*, *7*, 2855-2879, 2007.
- Kulmala, M., H. Vehkamäki, T. Petaja, M. Dal Maso, A. Lauri, V. M. Kerminen, W. Birmili, and P. H. McMurry, Formation and growth rates of ultrafine atmospheric particles: a review of observations, *J. Aerosol Sci.*, *35*, 143-176, 2004.
- Laakso, L., T. Grönholm, L. Kulmala, S. Haapanala, A. Hirsikko, e. R. Lovejoy, J. Kazil, T. Kurten, M. Boy, E. D. Nilsson, A. Sogachev, I. Riipinen, F. Stratmann, and M. Kulmala, Hot-air balloon as a platform for boundary layer profile measurements during particle formation, *Boreal Environ. Res.*, *12*, 279-294, 2007.
- Laaksonen, A., M. Kulmala, C. D. O'Dowd, J. Joutsensaari, P. Vaattovaara, S. Mikkonen, K. E. J. Lehtinen, L. Sogacheva, M. Dal Maso, P. Aalto, T. Petäjä, A. Sogachev, Y. J. Yoon, H. Lihavainen, D. Nilsson, M. C. Facchini, F. Cavalli, S. Fuzzi, T. Hoffmann, F. Arnold, M. Hanke, K. Sellegri, B. Umann, W. Junkermann, H. Coe, J. D. Allan, M. R. Alfarra, D. R. Worsnop, M. -L. Riekkola, T. Hyötyläinen, and Y. Viisanen, The role of VOC oxidation products in continental new particle formation, *Atmos. Chem. Phys.*, *8*, 2657-2665, 2008.
- Laurance, W. F., M. A. Cochrane, S. Bergen, P. M. Fearnside, P. Delamonica, C. Barber, S. D'Angelo, and T. Fernandes, Environment - The future of the Brazilian Amazon, *Science*, *291*, 438-439, 2001.
- Laurance, W. F., and G. B. Williamson, Positive feedbacks among forest fragmentation, drought, and climate change in the Amazon, *Cons. Bio.*, *15*, 1529-1535, 2001.

- Lee, A., A. H. Goldstein, J. H. Kroll, N. L. Ng, V. Varutbangkul, R. C. Flagan, and J. H. Seinfeld, Gas-phase products and secondary aerosol yields from the photooxidation of 16 different terpenes, *J. Geophys. Res.*, *111*, D17305, 2006.
- Lehtinen, K. E. J., M. Dal Maso, M. Kulmala, and V. M. Kerminen, Estimating nucleation rates from apparent particle formation rates and vice versa: Revised formulation of the Kerminen-Kulmala equation, *J. Aerosol Sci.*, *38*, 988-994, 2007.
- Lelieveld, J., T. M. Butler, J. N. Crowley, T. J. Dillon, H. Fischer, L. Ganzeveld, H. Harder, M. G. Lawrence, M. Martinez, D. Taraborrelli, and J. Williams, Atmospheric oxidation capacity sustained by a tropical forest, *Nature*, *452*, 737-740, 2008.
- Lewis, S. L., Tropical forests and the changing earth system, *Philos. Trans. R. Soc. Lond. B*, *361*, 195-210, 2006.
- Lim, H.-J., A. G. Carlton, and B. J. Turpin, Isoprene forms secondary organic aerosol through cloud processing: Model simulations, *Environ. Sci. Technol.*, *39*, 4441-4446, 2005.
- Liu, Y. Q., Atmospheric response and feedback to radiative forcing from biomass burning in tropical South America, *Agric. Forest Meteorol.*, *133*, 40-53, 2005.
- Mace, K. A., P. Artaxo, and R. A. Duce, Water-soluble organic nitrogen in Amazon Basin aerosols during the dry (biomass burning) and wet seasons, *J. Geophys. Res.*, *108*, 4512, 2003.
- Maenhaut, W., M. T. Fernandez-Jimenez, I. Rajta, and P. Artaxo, Two-year study of atmospheric aerosols in Alta Floresta, Brazil: Multielemental composition and source apportionment, *Nucl. Instr. Meth. Phys. Res. B*, *189*, 243-248, 2002.
- Mahowald, N. M., P. Artaxo, A. R. Baker, T. D. Jickells, G. S. Okin, J. T. Randerson, and A. R. Townsend, Impacts of biomass-burning emissions and land use change on Amazonian atmospheric phosphorus cycling and deposition, *Glo. Biogeochem. Cycles*, *19*, GB4030, 2005.
- Malhi, Y., J. T. Roberts, R. A. Betts, T. J. Killeen, W. H. Li, and C. A. Nobre, Climate change, deforestation, and the fate of the Amazon, *Science*, *319*, 169-172, 2008.
- Marengo, J. A., C. A. Nobre, and A. D. Culf, Climatic impacts of "friagens" in forested and deforested areas of the Amazon Basin, *J. Appl. Meteorol.*, *36*, 1553-1566, 1997.
- Marengo, J. A., C. A. Nobre, J. Tomasella, M. D. Oyama, G. S. De Oliveira, R. De Oliveira, H. Camargo, L. M. Alves, and I. F. Brown, The drought of Amazonia in 2005, *J. Clim.*, *21*, 495-516, 2008.
- Martin, S. T., et al, Amazonian Aerosol Characterization Experiment 2008 (AMAZE-08), *Atmos. Chem. Phys.*, **in preparation**, 2009.
- Martins, F. R., and E. B. Pereira, Parameterization of aerosols from burning biomass in the Brazil-SR radiative transfer model, *Solar Energy*, *80*, 231-239, 2006.
- Mayol-Bracero, O. L., P. Guyon, B. Graham, G. Roberts, M. O. Andreae, S. Decesari, M. C. Facchini, S. Fuzzi, and P. Artaxo, Water-soluble organic compounds in biomass burning aerosols over Amazonia - 2. Apportionment of the chemical composition and importance of the polyacidic fraction, *J. Geophys. Res.*, *107*, 8091, 2002.
- McFiggans, G., P. Artaxo, U. Baltensperger, H. Coe, M. C. Facchini, G. Feingold, S. Fuzzi, M. Gysel, A. Laaksonen, U. Lohmann, T. F. Mentel, D. M. Murphy, C. D. O'Dowd, J. R. Snider,

- and E. Weingartner, The effect of physical and chemical aerosol properties on warm cloud droplet activation, *Atmos. Chem. Phys.*, *6*, 2593-2649, 2006.
- McMurry, P. H., and S. K. Friedlander, New particle formation in the presence of an aerosol, *Atmos. Environ.*, *13*, 1635-1651, 1979.
- McMurry, P. H., M. A. Fink, H. Sakurai, M. R. Stolzenburg, L. Mauldin, K. Moore, J. N. Smith, F. L. Eisele, S. Sjostedt, D. Tanner, L. G. Huey, J. B. Nowak, E. Edgerton, and D. Voisin, A criterion for new particle formation in the sulfur-rich Atlanta atmosphere, *J. Geophys. Res.*, *110*, D22S02, 2005.
- Mircea, M., M. C. Facchini, S. Decesari, F. Cavalli, L. Emblico, S. Fuzzi, A. Vestin, J. Rissler, E. Swietlicki, G. Frank, M. O. Andreae, W. Maenhaut, Y. Rudich, and P. Artaxo, Importance of the organic aerosol fraction for modeling aerosol hygroscopic growth and activation: a case study in the Amazon Basin, *Atmos. Chem. Phys.*, *5*, 3111-3126, 2005.
- Mirme, A., E. Tamm, G. Mordas, M. Vana, J. Uin, S. Mirme, T. Bernotas, L. Laakso, A. Hirsikko, and M. Kulmala, A wide-range multi-channel air ion spectrometer, *Boreal Environ. Res.*, *12*, 247-264, 2007.
- Moteki, N., Y. Kondo, Y. Miyazaki, N. Takegawa, Y. Komazaki, G. Kurata, T. Shirai, D. R. Blake, T. Miyakawa, and M. Koike, Evolution of mixing state of black carbon particles: Aircraft measurements over the western Pacific in March 2004, *Geophys. Res. Lett.*, *34*, L11803, 2007.
- Nepstad, D. C., C. M. Stickler, B. Soares, and F. Merry, Interactions among Amazon land use, forests and climate: prospects for a near-term forest tipping point, *Philos. Trans. R. Soc. Lond. B*, *363*, 1737-1746, 2008.
- Nogues-Paegle, J., and K. C. Mo, Alternating wet and dry conditions over South America during summer, *Mon. Weath. Rev.*, *125*, 279-291, 1997.
- Ng, N. L., J. H. Kroll, M. D. Keywood, R. Bahreini, V. Varutbangkul, R. C. Flagan, J. H. Seinfeld, A. Lee, and A. H. Goldstein, Contribution of first- versus second-generation products to secondary organic aerosols formed in the oxidation of biogenic hydrocarbons, *Environ. Sci. Technol.*, *40*, 2283-2297, 2006.
- Nobre, C. A., M. A. Silva Dias, A. D. Culf, J. A. Polcher, J. H. C. Gash, J. A. Marengo, and R. Avissar (2004), The Amazonian climate, in *Vegetation, Water, Humans and the Climate: A New Perspective on an Interactive System (IGBP Series)* edited by P. Kabat, et al., Springer, Berlin..
- O'Dowd, C. D., M. C. Facchini, F. Cavalli, D. Ceburnis, M. Mircea, S. Decesari, S. Fuzzi, Y. J. Yoon, and J. P. Putaud, Biogenically driven organic contribution to marine aerosol, *Nature*, *431*, 676-680, 2004.
- Oyama, M. D., and C. A. Nobre, A new climate-vegetation equilibrium state for tropical South America, *Geophys. Res. Lett.*, *30*, 2199, 2003.
- Palmer, P. I., D. J. Jacob, A. M. Fiore, R. V. Martin, K. Chance, and T. P. Kurosu, Mapping isoprene emissions over North America using formaldehyde column observations from space, *J. Geophys. Res.*, *108*, 4180, 2003.
- Palmer, P. I., M. P. Barkley, T. P. Kurosu, A. C. Lewis, J. E. Saxton, K. Chance, and L. V. Gatti, Interpreting satellite column observations of formaldehyde over tropical South America, *Phil. Trans. R. Soc. A*, *365*, 1741-1751, 2007.

- Pauliquevis, T., L. L. Lara, M. L. Antunes, and P. Artaxo, Aerosol and precipitation chemistry in a remote site in Central Amazonia: the role of biogenic contribution, *Atmos. Chem. Phys. Discuss.*, *7*, 11465-11509, 2007.
- Penner, J. E., S. J. Ghan, and J. J. Walton (1991), The role of biomass burning in the budget and cycle of carbonaceous soot aerosols and their climate impact, in *Global Biomass Burning*, edited by J. Levine, pp. 432-438, MIT press, Cambridge.
- Perry, K. D., and P. V. Hobbs, Further evidence for particle nucleation in clear air adjacent to marine cumulus clouds, *J. Geophys. Res.*, *99*, 22803-22818, 1994.
- Prenni, A. J., Petters, M. D., Kreidenweis, S. M., and DeMott, P. J.: Cloud droplet activation of secondary organic aerosol, *J. Geophys. Res.*, *112*, D10223, 2007
- Platnick, S., and S. Twomey, Determining the Susceptibility of Cloud Albedo to Changes in Droplet Concentration with the Advanced Very High-Resolution Radiometer, *J. Appl. Meteor.*, *33*, 334-347, 1994.
- Pope, C. A., and D. W. Dockery, Health effects of fine particulate air pollution: Lines that connect, *J. Air Waste Man. Assoc.*, *56*, 709-742, 2006.
- Pöschl, U., Atmospheric aerosols: Composition, transformation, climate and health effects, *Ang. Chemie*, *44*, 7520-7540, 2005.
- Posfai, M., R. Simonics, J. Li, P. V. Hobbs, and P. R. Buseck, Individual aerosol particles from biomass burning in southern Africa: 1. Compositions and size distributions of carbonaceous particles, *J. Geophys. Res.*, *108*, 8483, 2003.
- Posfai, M., A. Gelencser, R. Simonics, K. Arato, J. Li, P. V. Hobbs, and P. R. Buseck, Atmospheric tar balls: Particles from biomass and biofuel burning, *J. Geophys. Res.*, *109*, D06213, 2004.
- Prather, K. A., T. Nordmeyer, and K. Salt, Real-time characterization of individual aerosol-particles using time-of-flight mass-spectrometry, *Anal. Chem.*, *66*, 1403-1407, 1994.
- Prenni, A. J., M. D. Petters, S. M. Kreidenweis, C. L. Heald, S. T. Martin, P. Artaxo, R. M. Garland, A. G. Wollny, U. Poeschl, Relative roles of biogenic emissions and Saharan dust as ice nuclei in the Amazon basin, *Nature Geo.*, *2*, 402-405, 2009.
- Prospero, J. M., R. A. Glaccum, and R. T. Nees, Atmospheric transport of soil dust from Africa to South America, *Nature*, *289*, 570-572, 1981.
- Rasmussen, R. A., and M. A. K. Khalil, Isoprene over the Amazon Basin, *J. Geophys. Res.*, *93*, 1417-1421, 1988.
- Reid, J. S., R. Koppmann, T. F. Eck, and D. P. Eleuterio, A review of biomass-burning emissions part II: intensive physical properties of biomass-burning particles, *Atmos. Chem. Phys.*, *5*, 799-825, 2005.
- Reinhardt, T. E., R. D. Ottmar, and C. Castilla, Smoke impacts from agricultural burning in a rural Brazilian town, *J. Air Waste Man. Assoc.*, *51*, 443-450, 2001.
- Riipinen, I., S.-L. Sihto, M. Kulmala, F. Arnold, M. Dal Maso, W. Birmili, K. Saarnio, K. Teinila, V.-M. Kerminen, A. Laaksonen, and K. E. J. Lehtinen, Connections between atmospheric sulphuric acid and new particle formation during QUEST III - IV campaigns in Hyytiälä and Heidelberg, *Atmos. Phys. Chem.*, *7*, 1899-1914, 2007.
- Rissler, J., E. Swietlicki, J. Zhou, G. Roberts, M. O. Andreae, L. V. Gatti, and P. Artaxo, Physical properties of the sub-micrometer aerosol over the Amazon rain forest during the

- wet-to-dry season transition - comparison of modeled and measured CCN concentrations, *Atmos. Chem. Phys.*, *4*, 2119-2143, 2004.
- Rissler, J., A. Vestin, E. Swietlicki, G. Fisch, J. Zhou, P. Artaxo, and M. O. Andreae, Size distribution and hygroscopic properties of aerosol particles from dry-season biomass burning in Amazonia, *Atmos. Chem. Phys.*, *6*, 471-491, 2006.
- Roberts, G., A. Nenes, J. Seinfeld, and M. Andreae, Impact of biomass burning on cloud properties in the Amazon Basin, *J. Geophys. Res.*, *108*, 4062, 2003.
- Roberts, G. C., M. O. Andreae, J. C. Zhou, and P. Artaxo, Cloud condensation nuclei in the Amazon Basin: "Marine" conditions over a continent?, *Geophys. Res. Lett.*, *28*, 2807-2810, 2001.
- Roberts, G. C., P. Artaxo, J. C. Zhou, E. Swietlicki, and M. O. Andreae, Sensitivity of CCN spectra on chemical and physical properties of aerosol: A case study from the Amazon Basin, *J. Geophys. Res.*, *107*, 8070, 2002.
- Roberts, G. C., and A. Nenes, A continuous-flow streamwise thermal-gradient CCN chamber for atmospheric measurements, *Aerosol Sci. Technol.*, *39*, 206-221, 2005.
- Rosenfeld, D., U. Lohmann, G. B. Raga, C. D. O'Dowd, M. Kulmala, S. Fuzzi, A. Reissell, and M. O. Andreae, Flood or drought: How do aerosols affect precipitation?, *Science*, *321*, 1309-1313, 2008.
- Sachse, G. W., R. C. Harriss, J. Fishman, G. F. Hill, and D. R. Cahoon, Carbon-monoxide over the Amazon Basin during the 1985 dry season, *J. Geophys. Res.*, *93*, 1422-1430, 1988.
- Saleska, S. R., K. Didan, A. R. Huete, and H. R. da Rocha, Amazon forests green-up during 2005 drought, *Science*, *318*, 612-612, 2007.
- Satyamurty, P., C. A. Nobre, and P. S. Dias, South America, in *Meteorology of the Southern Hemisphere*, edited by D. Karoly and D.G. Vincent, American Meteorological Society, Meteorological Monograph, Vol 27, pp. 119-139, 1998.
- Schafer, J. S., T. F. Eck, B. N. Holben, P. Artaxo, and A. F. Duarte, Characterization of the optical properties of atmospheric aerosols in Amazonia from long-term AERONET monitoring (1993-1995 and 1999-2006), *J. Geophys. Res.*, *113*, D04204, 2008.
- Schkolnik, G., A. H. Falkovich, Y. Rudich, W. Maenhaut, and P. Artaxo, New analytical method for the determination of levoglucosan, polyhydroxy compounds, and 2-methylerythritol and its application to smoke and rainwater samples, *Environ. Sci. Technol.*, *39*, 2744-2752, 2005.
- Schwartz, J., F. Laden, and A. Zanobetti, The concentration-response relation between PM_{2.5} and daily deaths, *Env. Health Per.*, *110*, 1025-1029, 2002.
- Schwarz, J. P., R. S. Gao, D. W. Fahey, D. S. Thomson, L. A. Watts, J. C. Wilson, J. M. Reeves, M. Darbeheshti, D. G. Baumgardner, G. L. Kok, S. H. Chung, M. Schulz, J. Hendricks, A. Lauer, B. Karcher, J. G. Slowik, K. H. Rosenlof, T. L. Thompson, A. O. Langford, M. Loewenstein, and K. C. Aikin, Single-particle measurements of midlatitude black carbon and light-scattering aerosols from the boundary layer to the lower stratosphere, *J. Geophys. Res.*, *111*, D16207, 2006.
- Shilling, J. E., Q. Chen, S. M. King, T. Rosenoern, J. H. Kroll, D. R. Worsnop, K. A. McKinney, and S. T. Martin, Particle mass yield in secondary organic aerosol formed by the dark ozonolysis of alpha-pinene, *Atmos. Chem. Phys.*, *8*, 2073-2088, 2008.

- Shim, C., Y. H. Wang, Y. Choi, P. I. Palmer, D. S. Abbot, and K. Chance, Constraining global isoprene emissions with Global Ozone Monitoring Experiment (GOME) formaldehyde column measurements, *J. Geophys. Res.*, *110*, D24301, 2005.
- Silva Dias, M. A. F., S. Rutledge, P. Kabat, P. L. S. Dias, C. Nobre, G. Fisch, A. J. Dolman, E. Zipser, M. Garstang, A. O. Manzi, J. D. Fuentes, H. R. Rocha, J. Marengo, A. Plana-Fattori, L. D. A. Sa, R. C. S. Alvala, M. O. Andreae, P. Artaxo, R. Gielow, and L. Gatti, Cloud and rain processes in a biosphere-atmosphere interaction context in the Amazon Region, *J. Geophys. Res.*, *107*, 8072, 2002.
- Sinha, B. W., et al., Composition and mixing state of wet season fine mode aerosol collected in the Amazonian tropical rain forest (Manaus, Brazil), EGU General Assembly, Abstract EGU2009-4814, 2009.
- Smith, J. N., M. J. Dunn, T. M. VanReken, K. Iida, M. R. Stolzenburg, P. H. McMurry, and L. G. Huey, Chemical composition of atmospheric nanoparticles formed from nucleation in Tecamac, Mexico: Evidence for an important role for organic species in nanoparticle growth, *Geophys. Res. Lett.*, *35*, L04808, 2008.
- Soares-Filho, B. S., D. C. Nepstad, L. M. Curran, G. C. Cerqueira, R. A. Garcia, C. A. Ramos, E. Voll, A. McDonald, P. Lefebvre, and P. Schlesinger, Modelling conservation in the Amazon Basin, *Nature*, *440*, 520-523, 2006.
- Solomon, P. A., and C. Sioutas, Continuous and semicontinuous monitoring techniques for particulate matter mass and chemical components: A synthesis of findings from EPA's particulate matter supersites program and related studies, *J. Air Waste Man. Assoc.*, *58*, 164-195, 2008.
- Soto-García, L. L., O. L. Mayol-Bracero, M. O. Andreae, P. Artaxo, W. Maenhaut, T. Kirchstetter, and T. Novakov, Evaluation of different methods for the determination of BC and OC during biomass burning in the Brazilian Amazon, **in preparation**, 2009.
- Spracklen, D. V., K. J. Pringle, K. S. Carslaw, M. P. Chipperfield, and G. W. Mann, A global off-line model of size-resolved aerosol microphysics: I. Model development and prediction of aerosol properties, *Atmos. Chem. Phys.*, *5*, 2227-2252, 2005.
- Spracklen, D. V., K. S. Carslaw, M. Kulmala, V. M. Kerminen, G. W. Mann, and S. L. Sihto, The contribution of boundary layer nucleation events to total particle concentrations on regional and global scales, *Atmos. Chem. Phys.*, *6*, 5631-5648, 2006.
- Stanier, C. O., A. Y. Khlystov, and S. N. Pandis, Nucleation events during the Pittsburgh air quality study: Description and relation to key meteorological, gas phase, and aerosol parameters, *Aerosol Sci. Technol.*, *38*, 253-264, 2004.
- Stavrakou, T., J.-F. Müller, I. D. Smedt, M. V. Roozendaal, G. R. v. d. Werf, L. Giglio, and A. Guenther, Global emissions of non-methane hydrocarbons deduced from SCIAMACHY formaldehyde columns through 2003-2006, *Atmos. Chem. Phys.*, *9*, 3663-3679, 2009.
- Stolzenburg, M. R., P. H. McMurry, H. Sakurai, J. N. Smith, R. L. Mauldin, F. L. Eisele, and C. F. Clement, Growth rates of freshly nucleated atmospheric particles in Atlanta, *J. Geophys. Res.*, *110*, D22S05, 2005.
- Stratmann, F., H. Siebert, G. Spindler, B. Wehner, D. Althausen, J. Heintzenberg, O. Hellmuth, R. Rinke, U. Schmieder, C. Seidel, T. Tuch, U. Uhrner, A. Wiedensohler, U. Wandinger, M. Wendisch, D. Schell, and A. Stohl, New-particle formation events in a continental boundary layer: first results from the SATURN experiment, *Atmos. Chem. Phys.*, *3*, 1445-1459, 2003.

- Streets, D. G., Dissecting future aerosol emissions: Warming tendencies and mitigation opportunities, *Clim. Change*, *81*, 313-330, 2007.
- Suni, T., M. Kulmala, A. Hirsikko, T. Bergman, L. Laakso, P. P. Aalto, R. Leuning, H. Cleugh, S. Zegelin, D. Hughes, E. van Gorsel, M. Kitchen, M. Vana, U. Horrak, S. Mirme, A. Mirme, S. Sevanto, J. Twining, and C. Tardos, Formation and characteristics of ions and charged aerosol particles in a native Australian Eucalypt forest, *Atmos. Chem. Phys.*, *8*, 129-139, 2008.
- Surratt, J. D., Y. Gómez-González, A. W. H. Chan, R. Vermeylen, M. Shahgholi, T. E. Kleindienst, E. O. Edney, J. H. Offenberg, M. Lewandowski, M. Jaoui, W. Maenhaut, M. Claeys, R. C. Flagan, and J. H. Seinfeld, Organosulfate Formation in Biogenic Secondary Organic Aerosol, *J. Phys. Chem. A*, *112*, 8345-8378, 2008.
- Svenningsson, B., J. Rissler, E. Swietlicki, M. Mircea, M. Bilde, M. C. Facchini, S. Decesari, S. Fuzzi, J. Zhou, J. Monster, and T. Rosenorn, Hygroscopic growth and critical supersaturations for mixed aerosol particles of inorganic and organic compounds of atmospheric relevance, *Atmos. Chem. Phys.*, *6*, 1937-1952, 2006.
- Swap, R., M. Garstang, S. Greco, R. Talbot, and P. Kållberg, Saharan dust in the Amazon Basin, *Tellus*, *44B*, 133-149, 1992.
- Swietlicki, E., H. C. Hansson, K. Hameri, B. Svenningsson, A. Massling, G. McFiggans, P. H. McMurry, T. Petaja, P. Tunved, M. Gysel, D. Topping, E. Weingartner, U. Baltensperger, J. Rissler, A. Wiedensohler, and M. Kulmala, Hygroscopic properties of submicrometer atmospheric aerosol particles measured with H-TDMA instruments in various environments - a review, *Tellus B*, *60*, 432-469, 2008.
- Talbot, R. W., M. O. Andreae, T. W. Andreae, and R. C. Harriss, Regional aerosol chemistry of the Amazon Basin during the dry season, *J. Geophys. Res.*, *93*, 1499-1508, 1988.
- Talbot, R. W., M. O. Andreae, H. Berresheim, P. Artaxo, M. Garstang, R. C. Harriss, K. M. Beecher, and S. M. Li, Aerosol chemistry during the wet season in central Amazonia - the influence of long-range transport, *J. Geophys. Res.*, *95*, 16955-16969, 1990.
- Tegen, I., P. Hollrig, M. Chin, I. Fung, D. Jacob, and J. Penner, Contribution of different aerosol species to the global aerosol extinction optical thickness: Estimates from model results, *J. Geophys. Res.*, *102*, 23895-23915, 1997.
- Textor, C., M. Schulz, S. Guibert, S. Kinne, Y. Balkanski, S. Bauer, T. Berntsen, T. Berglen, O. Boucher, M. Chin, F. Dentener, T. Diehl, R. Easter, H. Feichter, D. Fillmore, S. Ghan, P. Ginoux, S. Gong, J. E. Kristjansson, M. Krol, A. Lauer, J. F. Lamarque, X. Liu, V. Montanaro, G. Myhre, J. Penner, G. Pitari, S. Reddy, O. Seland, P. Stier, T. Takemura, and X. Tie, Analysis and quantification of the diversities of aerosol life cycles within AeroCom, *Atmos. Chem. Phys.*, *6*, 1777-1813, 2006.
- Tie, X. X., S. Madronich, S. Walters, D. P. Edwards, P. Ginoux, N. Mahowald, R. Y. Zhang, C. Lou, and G. Brasseur, Assessment of the global impact of aerosols on tropospheric oxidants, *J. Geophys. Res.*, *110*, D03204, 2005.
- Trebs, I., F. X. Meixner, J. Slanina, R. Otjes, P. Jongejan, and M. O. Andreae, Real-time measurements of ammonia, acidic trace gases and water-soluble inorganic aerosol species at a rural site in the Amazon Basin, *Atmos. Chem. Phys.*, *4*, 967-987, 2004.
- Trebs, I., S. Metzger, F. X. Meixner, G. N. Helas, A. Hoffer, Y. Rudich, A. H. Falkovich, M. A. L. Moura, R. S. da Silva, P. Artaxo, J. Slanina, and M. O. Andreae, The $\text{NH}_4^+\text{NO}_3^- \text{-Cl}^- \text{-SO}_4^{2-}$

- H₂O aerosol system and its gas phase precursors at a pasture site in the Amazon Basin: How relevant are mineral cations and soluble organic acids?, *J. Geophys. Res.*, *110*, D07303, 2005.
- Trebs, I., M. O. Andreae, W. Elbert, O. L. Mayol-Bracero, L. L. Soto-Garcia, Y. Rudich, A. H. Falkovich, W. Maenhaut, P. Artaxo, R. Otjes, and J. Slanina, Aerosol inorganic composition at a tropical site: Discrepancies between filter-based sampling and a semi-continuous method, *Aerosol Sci. Technol.*, *42*, 255-269, 2008.
- Tsigaridis, K., and M. Kanakidou, Global modelling of secondary organic aerosol in the troposphere: a sensitivity analysis, *Atmos. Chem. Phys.*, *3*, 1849-1869, 2003.
- Twomey, S., The influence of pollution on the short-wave albedo of clouds, *J. Atmos. Sci.*, *34*, 1149-1152, 1977.
- Unger, N., D. T. Shindell, D. M. Koch, and D. G. Streets, Air pollution radiative forcing from specific emissions sectors at 2030, *J. Geophys. Res.*, *113*, D02306, 2008.
- Vana, M., M. Kulmala, M. Dal Maso, U. Horrak, and E. Tamm, Comparative study of nucleation mode aerosol particles and intermediate air ions formation events at three sites, *J. Geophys. Res.*, *109*, D17201, 2004.
- Vestin, A., J. Rissler, E. Swietlicki, G. P. Frank, and M. O. Andreae, Cloud-nucleating properties of the Amazonian biomass burning aerosol: Cloud condensation nuclei measurements and modeling, *J. Geophys. Res.*, *112*, D14201, 2007.
- Volkamer, R., F. S. Martini, L. T. Molina, D. Salcedo, J. L. Jimenez, and M. J. Molina, A missing sink for gas-phase glyoxal in Mexico City: Formation of secondary organic aerosol, *Geophys. Res. Lett.*, *34*, L19807, 2007.
- Wang, H., and R. Fu, Cross-equatorial flow and seasonal cycle of precipitation over South America, *J. Clim.*, *15*, 1591-1608, 2002.
- Watson, J. G., Visibility: Science and regulation, *J. Air Waste Man. Assoc.*, *52*, 628-713, 2002.
- Went, F. W., Blue hazes in the atmosphere, *Nature*, *187*, 641-643, 1960.
- Williams, E., D. Rosenfeld, N. Madden, J. Gerlach, N. Gears, L. Atkinson, N. Dunnemann, G. Frostrom, M. Antonio, B. Biazon, R. Camargo, H. Franca, A. Gomes, M. Lima, R. Machado, S. Manhaes, L. Nachtigall, H. Piva, W. Quintiliano, L. Machado, P. Artaxo, G. Roberts, N. Renno, R. Blakeslee, J. Bailey, D. Boccippio, A. Betts, D. Wolff, B. Roy, J. Halverson, T. Rickenbach, J. Fuentes, and E. Avelino, Contrasting convective regimes over the Amazon: Implications for cloud electrification, *J. Geophys. Res.*, *108*, 8082, 2002.
- Worobiec, A., I. Szaloki, J. Osan, W. Maenhaut, E. A. Stefaniak, and R. Van Grieken, Characterisation of Amazon Basin aerosols at the individual particle level by X-ray microanalytical techniques, *Atmos. Environ.*, *41*, 9217-9230, 2007.
- Yamasoe, M. A., P. Artaxo, A. H. Miguel, and A. G. Allen, Chemical composition of aerosol particles from direct emissions of vegetation fires in the Amazon Basin: water-soluble species and trace elements, *Atmos. Environ.*, *34*, 1641-1653, 2000.
- Yin, Y., Z. Levin, T. G. Reisin, and S. Tzivion, The effects of giant cloud condensation nuclei on the development of precipitation in convective clouds - a numerical study, *Atmos. Res.*, *53*, 91-116, 2000.

- Yokelson, R. J., T. Karl, P. Artaxo, D. R. Blake, T. J. Christian, D. W. T. Griffith, A. Guenther, and W. M. Hao, The Tropical Forest and Fire Emissions Experiment: overview and airborne fire emission factor measurements, *Atmos. Chem. Phys.*, *7*, 5175-5196, 2007.
- Yokelson, R. J., T. J. Christian, T. G. Karl, and A. Guenther, The tropical forest and fire emissions experiment: laboratory fire measurements and synthesis of campaign data, *Atmos. Chem. Phys.*, *8*, 3509-3527, 2008.
- Yu, F., Z. Wang, G. Luo, and R. Turco, Ion-mediated nucleation as an important global source of tropospheric aerosols, *Atmos. Chem. Phys.*, *8*, 2537-2554, 2008.
- Zdrahal, Z., J. Oliveira, R. Vermeylen, M. Claeys, and W. Maenhaut, Improved method for quantifying levoglucosan and related monosaccharide anhydrides in atmospheric aerosols and application to samples from urban and tropical locations, *Environ. Sci. Technol.*, *36*, 747-753, 2002.
- Zhang, Y., R. Fu, H. B. Yu, Y. Qian, R. Dickinson, M. Dias, P. L. D. Dias, and K. Fernandes, Impact of biomass burning aerosol on the monsoon circulation transition over Amazonia, *Geophys. Res. Lett.*, *36*, L10814, 2009.
- Zhou, J. C., E. Swietlicki, H. C. Hansson, and P. Artaxo, Submicrometer aerosol particle size distribution and hygroscopic growth measured in the Amazon rain forest during the wet season, *J. Geophys. Res.*, *107*, 8055, 2002.
- Zhou, J. Y., and K. M. Lau, Does a monsoon climate exist over South America?, *J. Clim.*, *11*, 1020-1040, 1998.
- Ziese, M., H. Wex, E. Nilsson, I. Salma, R. Ocskay, T. Hennig, A. Massling, and F. Stratmann, Hygroscopic growth and activation of HULIS particles: experimental data and a new iterative parameterization scheme for complex aerosol particles, *Atmos. Chem. Phys.*, *8*, 1855-1866, 2008.
- Zimmerman, P. R., J. P. Greenberg, and C. E. Westberg, Measurements of atmospheric hydrocarbons and biogenic emission fluxes in the Amazon boundary layer, *J. Geophys. Res.*, *93*, 1407-1416, 1988.

List of Tables

Table 1. Major international measurement campaigns of Amazonian aerosol particles.

Beginning in the 1980s, large-scale measurement campaigns focusing on the atmospheric sciences began in the Amazon Basin, and many of them had at least a component devoted to aerosol research. Over the same time span, these limited-term campaigns have been complemented by some extended measurement series at fixed ground stations.

Table 2. Statistics of particle number-diameter distributions measured during CLAIRE-98.

Table 3. Study-average concentrations (ng m^{-3}) measured for fine- and coarse-mode inorganic species during the wet and dry seasons of the Amazon Basin. Time periods of the studies are categorized as weakly, moderately, or strongly affected by biomass burning. Time periods are also categorized for the relative influence of sources from outside the Basin. Table entries are rounded to two significant figures for presentation. *Notes: Fine fraction is defined as aerodynamic diameter under $1 \mu\text{m}$ for studies b and o, under $2 \mu\text{m}$ for studies a, c, f, g, h, i, j, k, & m, and under $2.5 \mu\text{m}$ for studies n and p. Coarse fraction is defined as aerodynamic diameters from 2 to $10 \mu\text{m}$ for studies f, g, h, i, j, k, & m, of 2 to $15 \mu\text{m}$ for studies a & c, under $10 \mu\text{m}$ for study l, and greater than $1 \mu\text{m}$ for study b. Total suspended particulate matter is used as an approximation of coarse-mode mass for studies d and e. For studies n and p, medians rather than means are listed. The sulfate column includes that measured by ion chromatography as well as that inferred on an equivalent basis from PIXE measurements of elemental sulfur (i.e., elemental sulfur attributed entirely to sulfate in the analysis). References: a. Artaxo et al. [1988]; b. Talbot et al. [1988]; c. Artaxo et al [1990]; d. Talbot et al. [1990]; e. Artaxo et al [1998]; f. Echalar et al. [1998]; g. Gerab et al.[1998]; h. Formenti et al. [2001]; i. Artaxo et al. [2002]; j. Maenhaut et al. [2002]; k. Guyon et al. [2003c]; l. Mace et al. [2003]; m. Graham et al. [2003a]; n. Trebs et al. [2005]; o. Fuzzi et al. [2007]; p. Trebs et al. [2008].*

List of Figures

- Figure 1.** Map showing the geographic boundaries of the Amazon forest (red line), the Amazon-Tocantins River Basins (purple line), and Brazilian Legal Amazon (orange line). The political boundaries of South America are shown as black lines. Cities often mentioned in the literature of Amazonian aerosol particles are indicated. Specific research sites for some of the campaigns listed in Table 1 are also highlighted, including (1) FNS and RBJ in Rondônia in southwestern Amazonia, (2) K34, TT34, and C14 north of Manaus in central Amazonia, (3) Balbina in central Amazonia, and (4) Caxiuanã in eastern Amazonia.
- Figure 2.** Scanning electron micrographs of primary biological particles collected in the Amazon Basin. Source: unpublished results of the Max Planck Institute for Chemistry, Mainz, Germany, for particles collected during the project European Studies on Trace Gases and Atmospheric Chemistry as a Contribution to the Large-Scale Biosphere-Atmosphere Experiment in the Amazon Basin (LBA-EUSTACH) in 1999.
- Figure 3.** Illustration of vertical mixing processes that affect the particle number-diameter distribution of aerosol particles in the Amazon Basin. The lowest 4 km represents daytime conditions with a fully developed mixing layer and shallow convection in the transition layer. Upward transport is controlled by the deep convection and by the fair-weather cumulus clouds. Subsidence dominates the large-scale downward transport. Shown also are wind roses from *Krejci et al.* [2003]. The original figure in *Krejci et al.* [2003] was prepared to describe observations over Suriname but the processes depicted are applicable to the wider Amazon Basin.
- Figure 4.** Source classification scheme for Amazonian aerosol particles. Although emissions from the Amazonian biosphere are active at all times and have low variability year-round, they are relatively weak, and particles and their components can be dominated at times by influences from outside of the Amazon Basin, such as from Saharan dust,

African biomass burning, or Atlantic marine emissions. The dust and marine emissions are a natural contribution because they were present in the year 1750. At other times, anthropogenic influences such as in-Basin biomass burning in the dry season can dominate the type, the number and, the mass concentrations of Amazonian aerosol particles. In this figure, “favored” suggests conditions of greater probability, although all influences are possible under most conditions. For example, in the wet season biomass burning can still influence some observations, such as sampling sites downwind of the border area of Brazil and Guyana and Suriname, which engage in biomass burning during the northern Amazonian wet season (cf. Figure 8).

Figure 5. (a) December-through-February (DJF) wind vector and wind speed (m s^{-1}) at 1000 hPa for South America. (b) Same as *a* but averaged for June through August (JJA). (c) DJF outgoing longwave radiation (W m^{-2}). (d) Same as *c* but for JJA. The horizontal black line shows the position of the equator. Data represent the average reanalysis of 1988-2007 from the National Centers for Environmental Prediction (NCEP).

Figure 6. Illustration of the typical diel evolution of the lower troposphere in the Amazon Basin. Adapted from *Rissler et al.* [2006].

Figure 7. (A) Number-diameter and (B) volume-diameter distributions for giant Amazon aerosol particles ($> 4 \mu\text{m}$), as determined by light microscopic analysis of samples collected on glass slides. Shown are particle totals, biological particle totals, and percentage contribution of biological particles to the total. Data are from M. Andreae (personal communication).

Figure 8. Fire locations in Feb, May, Aug, and Nov 2007 based on the MODIS Collection-5 Active Fire Product [*Giglio et al.*, 2006]. Red (scaled from 0 to 300) is the number of fire pixels, with corrections for cloud cover. Some boxes have more than 300 fire pixels (e.g., the maximum value for the scenes shown is 1144 fire pixels). Figure prepared by S.T. Martin and C.L. Heald.

Figure 9. (top) Mean diel NO, NO₂, and O₃ concentrations at the LBA-EUSTACH primary forest site Reserva Biológica Jarú (RBJ) and at the LBA-EUSTACH pasture site Fazenda Nossa Senhora Aparecida (FNS) in 1999. Both sites are located in Rondônia (cf. Figure 1). Measurements were taken 3.5 m above ground at the pasture site and 20 m above the rain forest canopy. Data are presented as 1-hour medians over 27 days of the LBA-EUSTACH-1 campaign (mostly natural conditions) and 46 days of the LBA-EUSTACH-2 campaign (strongly influenced by biomass burning). (bottom, left) A midday CO profile taken from a flight out of Manaus on July 18 over remote forest in 1985 for natural conditions (ABLE-2A). (bottom, right) Temporal variation of the CO mixing ratio at 30 min intervals at the FNS surface station during LBA-SMOCC in 2002. Adapted from *Sachse et al.* [1988], *Andreae et al.* [2002], and *Chand et al.* [2006].

Figure 10. Mean vertical profiles of NO, NO₂, and O₃ in the forest canopy. Profiles represent an average over 43 days in Rondônia at 14:00 local time during LBA-EUSTACH-2. Data points of concentrations are medians (0.5-quantile), and corresponding variations are indicated by horizontal bars (left end: 0.25-quantile, right end: 0.75-quantile). Variation of the NO concentration above 5 m is smaller than symbol size. Adapted from *Andreae et al.* [2002].

Figure 11. (top) Diel fluxes and (bottom) mixing ratios of isoprene and monoterpenes measured in central Amazonia (tower C14; cf. Figure 1) between 14 and 29 September 2004. Black lines represent the study mean and gray area represents the standard deviation. Adapted from *Karl et al.* [2007].

Figure 12. Scatter plots between the amount of OH observed from aircraft and that modeled for the boundary layer over Suriname in October 2005. (a) The standard model. (b) An updated model including the role of organic peroxy radicals. The solid lines indicate ideal agreement and the dashed lines the $\pm 40\%$ range, based on the measurement accuracy. (c) Percentage difference in the annual mean OH, as calculated using the

updated model compared to the standard model (the arrow indicates the location of Suriname). The aircraft measurements were performed in October 2005 over the pristine forests of Suriname, Guyana, and Guyane (French Guiana). Adapted from *Lelieveld et al.* [2008]. [permission pending]

Figure 13. Particle number-diameter distributions observed over southern Suriname. N_6 , N_{18} , and N_{120} represent the number concentration of particles larger than 6, 18, and 120 nm, respectively. The difference N_{6-18} , equal to $N_6 - N_{18}$, quantifies the concentration of nucleation-mode particles. The arrows on the left mark the altitude where the average number-diameter distributions on the right were measured. The error bars on the average distributions represent lower and upper quartiles. Adapted from *Krejci et al.* [2005].

Figure 14. Diel variation in the number concentration of nucleation-mode (viz. < 25 nm) particles. Shown are averages for periods of LBA-SMOCC that were weakly, moderately, and strongly influenced by biomass burning. Adapted from *Rissler et al.* [2006].

Figure 15. Time series of particle mass concentrations in Balbina (central Amazonia), Santarem (eastern Amazonia), and Alta Floresta (southern Amazonia). Data are shown as stack bar plots of fine (red; $< 2 \mu\text{m}$) and coarse (blue; 2 to $10 \mu\text{m}$) fractions. The PM_{10} concentration is the sum of the two. The measurement protocol follows that of the US EPA for weighing filters, and some water was therefore also included in the mass measurement. The relative mass of water, however, was small because of the low hygroscopic growth factors of Amazonian aerosol particles. Adapted and updated from *Artaxo et al.* [2002].

Figure 16. Particle mass-diameter distributions from gravimetric analysis of MOUDI stages. (top) Examples of three distributions collected during AMAZE-08 at times when natural conditions prevailed. Each distribution represents one week of data (10 to 16 Mar 2008; 22 to 31 Mar 2008; and 9 to 17 Apr 2008). (bottom) Representative

distribution collected during LBA-SMOCC during time periods strongly influenced by biomass burning. A bin label such as “3.2 to 10 μm ” represents a mass filter having 50% cutpoints at 3.2 and 10 μm . Data are from P. Artaxo (personal communication).

Figure 17. (A) A typical particle number-diameter distribution observed for natural conditions in central Amazonia during CLAIRE-98. (B-D) Whole-campaign histograms of Aitken, accumulation-mode, and total particle number concentrations. Natural conditions mostly prevailed. Adapted from *Zhou et al.* [2002].

Figure 18. Average particle number-diameter distributions observed in the Amazon Basin for periods during which natural conditions prevailed. (top) CLAIRE-98 (Balbina, 18 days of data) [*Zhou et al.*, 2002], CLAIRE-01 (Balbina, two days of data) [*Rissler et al.*, 2004], and AMAZE-08 (approximately equidistant to Manaus and Balbina, 22 Feb - 12 Mar 2008) [E. Swietlicki, personal communication]. (bottom) Airborne measurements over Suriname during CLAIRE-98 [*Krejci et al.*, 2003]. Altitude ranges were from (1) 0.2 to 1.2 km and (2) 1.2 to 2.4 km.

Figure 19. Mean composition of the (identified) water-soluble organic carbon. Fine-mode filter samples collected in the different periods of the LBA-SMOCC campaign were examined by chromatography. Key: monocarboxylic acids (MCA), oxalic acid, C₃-C₆ dicarboxylic acids (DCA C3-C6), tricarboxylic acids (TCA), aromatic acids (ArAcids), aromatic aldehydes (ArAld), anhydrosugars (AnSugars), sugar-alcohols, sugars, and 2-methyltetrols (MeTetrols). Adapted from *Decesari et al.* [2006].

Figure 20. Total-carbon apportionment for biomass-burning particles collected during the dry season (LBA EUSTACH). Total carbon is divided into black carbon and organic carbon; organic carbon is partitioned into water-insoluble and water-soluble fractions; the portion of the water-soluble fraction that is elutable and identifiable by HPLC is indicated; the fraction of that eluate that is identifiable by GC-MS is indicated; and the partitioning of that fraction into the chemical species is shown. This final fraction

is also represented in Figure 19. “BC water” is based on thermal analysis for black carbon after washing the sample with water. Adapted from *Mayol-Bracero et al.* [2002].

Figure 21. Two-day time series of chemically apportioned, submicron particle mass concentrations measured on February 17 and 18, 2008, in central Amazonia as part of AMAZE-08. Measurements were made using an Aerodyne high-resolution time-of-flight aerosol mass spectrometer (HR-TOF-AMS) during a time period for which natural conditions prevailed. (a) Time series of organic mass fraction. (b) Time series of submicron mass concentrations of organic material (green), sulfate (red), ammonium (orange), nitrate (blue), and chloride (magenta). The measurement window of the AMS is approximately 60 to 600 nm in vacuum aerodynamic diameter. Data are from S.T. Martin (personal communication).

Figure 22. Spectra of CCN activation in the Amazon Basin for a range of natural to very polluted conditions. The highest concentrations occur for locations heavily influenced by in-Basin biomass burning. Adapted from *Roberts et al.* [2003].

Figure 23. The evolution of the mass-diameter distribution of cloud drops for increasing height in growing convective clouds, in regimes of (A) blue ocean, off the northeast Brazilian coast (4°S, 38°W), (B) green ocean (i.e., natural conditions), at the western tip of the Amazon (6°S, 73°W), (C) smoky clouds in Rondonia (10°S, 62°W), and (D) pyro-clouds. The lowest-altitude distribution in each plot represents conditions at cloud base, except in *D* for which a distribution for large ash particles outside of the cloud is also shown. Adapted from *Andreae et al.* [2004]. [permission pending]

Figure 24. Amazon highways. The government of Brazil plans that the core region of the Amazon will be accessible by all-weather highways (yellow) in the future. Although these investments in transportation promise to lower the production costs of ranching and farming, they potentially threaten to stimulate deforestation. Source: D.C. Nepstad, *A Report to the World Wide Fund for Nature (WWF)*, 2007.

Figure 25. Simulations of forest cover for the year 2050. (a) Scenario of business as usual. (b) Scenario of good governance. Adapted from *Soares-Filho et al.* [2006]. [permission pending].

Dates	Campaign	Location & Description	Ground	Airborne	Science Focus	Reference	
Jul-Aug 1985	ABLE-2A	<i>Amazon Boundary-Layer Experiment - dry season in Amazonia</i> <i>dry season</i>	Y (Manaus)	Y (ER-2 and Electra)	trace gas and meteorological measurements	[Harriss et al., 1988]	
Apr-May 1987	ABLE-2B	<i>Amazon Boundary-Layer Experiment - wet season in Amazonia</i> <i>wet season</i>	Y (Manaus)	Y (ER-2 and Electra)	trace gas and meteorological measurements	[Harriss et al., 1990]	
Aug-Sep 1995	SCAR-B	<i>The Smoke, Clouds, and Radiation-Brazil</i>	dry season in disturbed southern Amazonia (Brasilia and Alta Floresta in southern Amazonia)	Y	Y (ER-2 and Convair)	effects of biomass burning on climate (direct effect)	[Kaufman et al., 1998]
Mar-Apr 1998	CLAIRE-98	<i>The Cooperative LBA Airborne Regional Experiment</i>	wet season in undisturbed central Amazonia (Balbina in central Amazonia)	Y		chemistry and physics of biogenic aerosols	[Avisar et al., 2002]
Apr-May 1999 Sep-Oct 1999	LBA-EUSTACH	<i>European Studies on Trace Gases and Atmospheric Chemistry as a Contribution to the Large-Scale Biosphere-Atmosphere Experiment in Amazonia</i>	wet season in disturbed western Amazonia (FNS and RBJ in Rondônia in southwestern Amazonia; K34 and C14 in central Amazonia; Caxiuana in eastern Amazonia)	Y	Y (Bandeirante)	chemistry and physics of wet season biogenic aerosols; trace gas emissions and reactivity	[Andreae et al., 2002]
Jul 2001	CLAIRE-01	<i>The Cooperative LBA Airborne Regional Experiment</i>	transition of wet-to-dry season in undisturbed central Amazonia (Balbina)	Y	Y (Bandeirante)	chemistry and physics of biogenic aerosols; trace gas emissions and reactivity	
Sep-Nov 2002	LBA-SMOCC	<i>The Large-Scale Biosphere-Atmosphere Experiment in Amazonia: Smoke, Aerosols, Clouds, Rainfall, and Climate field campaign</i>	dry season biomass burning experiment in pasture and forest sites in disturbed Amazonia (FNS and RBJ Rondônia)	Y	Y (Bandeirante)	effects of biomass burning on clouds (indirect effect)	[Fuzzi et al., 2007]
Aug-Sep 2004	TROFFEE	<i>Tropical Forest and Fire Emissions Experiment</i>	dry season in disturbed southern Amazonia (Alta Floresta)		Y (Bandeirante)	measurements of aerosol and trace gas emission factors from biomass burning	[Yokelson et al., 2007]
Feb-Mar 2008	AMAZE-08	<i>Amazonian Aerosol Characterization Experiment</i>	wet season in undisturbed central Amazonia (TT34 north of Manaus in central Amazonia)	Y	N	chemistry and physics of biogenic aerosols; trace gas emissions and reactivity	<i>Martin et al.</i> [in preparation]

Table 1

Mode	Frequency of Occurrence (%)	Number Concentration (# cm ⁻³)			Geometric Mean Diameter (nm)	Geometric Standard Deviation
		Mean ± Standard Deviation	Geometric Mean	Median		
Ultrafine	18	92±99	55	48	24±10	1.31±0.15
Aitken	100	239±154	200	200	68±12	1.40±0.14
Accumulation	100	177±115	137	146	151±22	1.40±0.10

Table 2

		Out-of-Basin																
	In-Basin Biomass Burning	Dust, Marine, African Biomass Burning	SO ₄ ²⁻	NO ₃ ⁻	NH ₄ ⁺	K	P	Ca	Zn	Cl	Na	Al	Si	Ti	Mn	Fe	Reference	
Wet Season																		
Fine	weakly	weakly	170-260	60-140	31-340	18-34	2.8-7.4	5.0-12	0.25-0.78	0.75-65	26-49	23-62	38-160	1.5-5.0	0.22-0.86	12-37	c,h,i,k,n,o,p	
	weakly	strongly	690	-	-	200	13	120	1.6	180	270	690	1400	46	5.5	340	h	
	moderately	moderately	510-1000	21	140	80-220	1.9-14	8.0-40	1.2-2.3	1.0-37	51-190	33-130	27-300	2.1-11	0.53-3.6	24-82	f,g,j	
Coarse	weakly	weakly	38-140	25-36	5.0-13	39-110	8.0-38	11-39	0.60-1.6	8.9-160	54-91	44-92	50-220	4.0-8.6	0.40-1.9	25-64	c,d,h,i,k,l,o	
	weakly	strongly	430	-	-	260	37	280	1.9	1100	660	1100	2100	72	10	540	h	
	moderately	moderately	200-420	-	-	140-270	7.6-87	44-230	2.2-7.4	14-1100	82-2100	100-450	230-970	17-46	2.4-59	190-790	f,g,j	
Dry Season																		
Fine	weakly	weakly	320-780	14-68	5.0-180	35-160	4.0-6.6	4.0-9.2	0.93-2.6	6.0-8.2	46	8.0-16	25	0.90-1.5	0.19-0.80	6.5-11	a,b,m	
	strongly	weakly	1600-3300	1000-1400	860-1600	510-1100	27-33	25-51	4.2-8.3	20-62	-	89-270	120-360	6.0-33	1.9-5.0	28-160	i,k,n,o	
	strongly	strongly	1200-2900	-	-	440-1300	4.7-32	1.5-53	5.3-8.6	12-110	96	100-640	230-540	11-28	1.5-3.8	190-270	f,j	
Coarse	weakly	weakly	51-200	110-210	27-40	33-100	8.9-32	11-18	0.48-1.8	51-81	-	19-35	33-66	1.4-3.5	0.32-0.60	11-23	a,b,m	
	strongly	weakly	180-470	48-520	97-170	88-240	47-68	14-52	2.9-4.3	10-28	-	79-500	90-880	8.0-65	3.9-16	48-460	i,k,l,o	
	strongly	strongly	480-3600	-	-	390-1600	17-140	0.86-1300	4.7-11	24-220	-	1200-3100	2100-3100	110-140	14-68	1100-1800	e,f,j	

Table 3

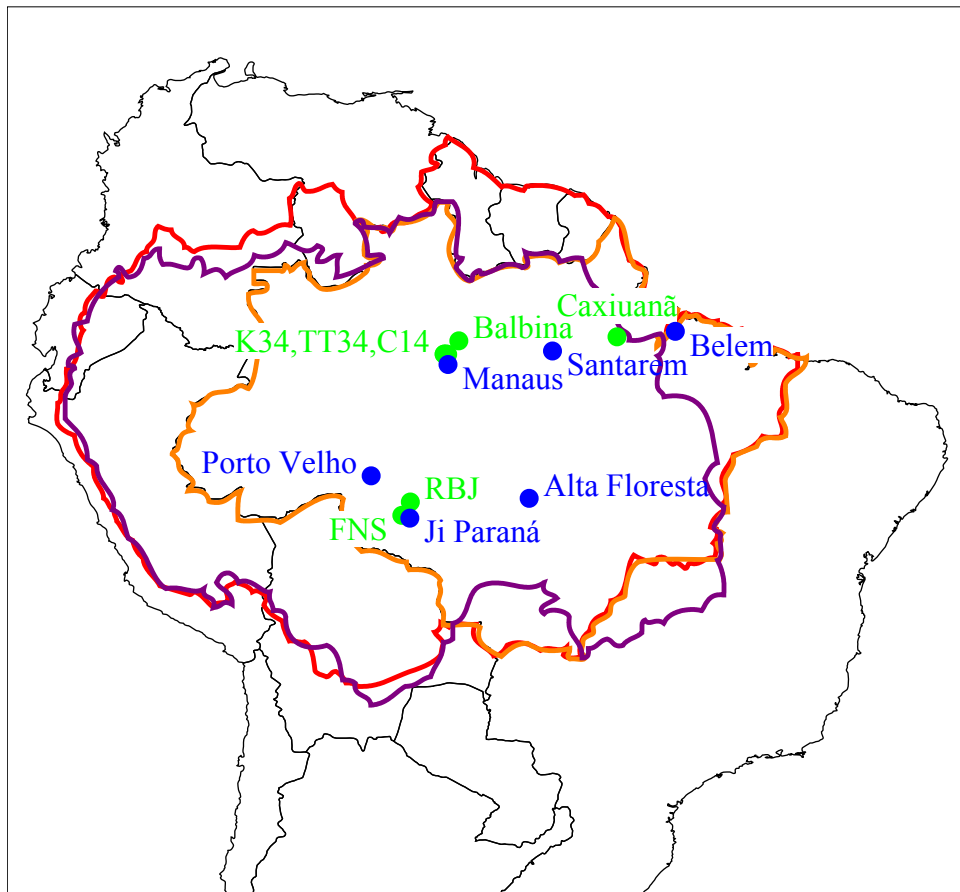


Figure 1

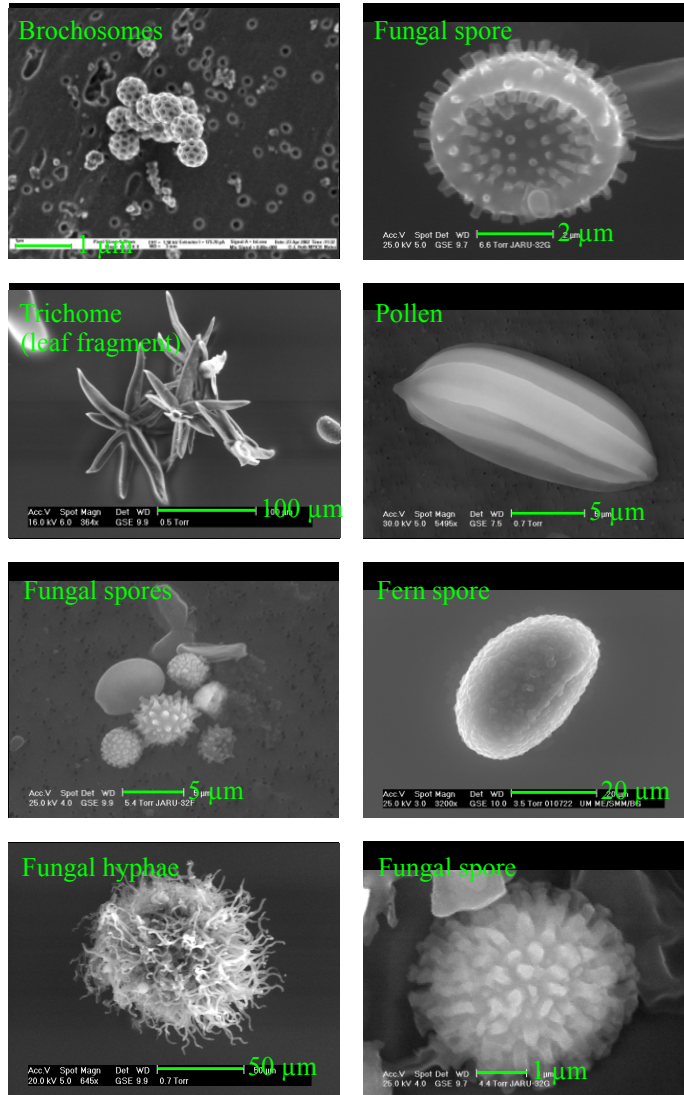


Figure 2

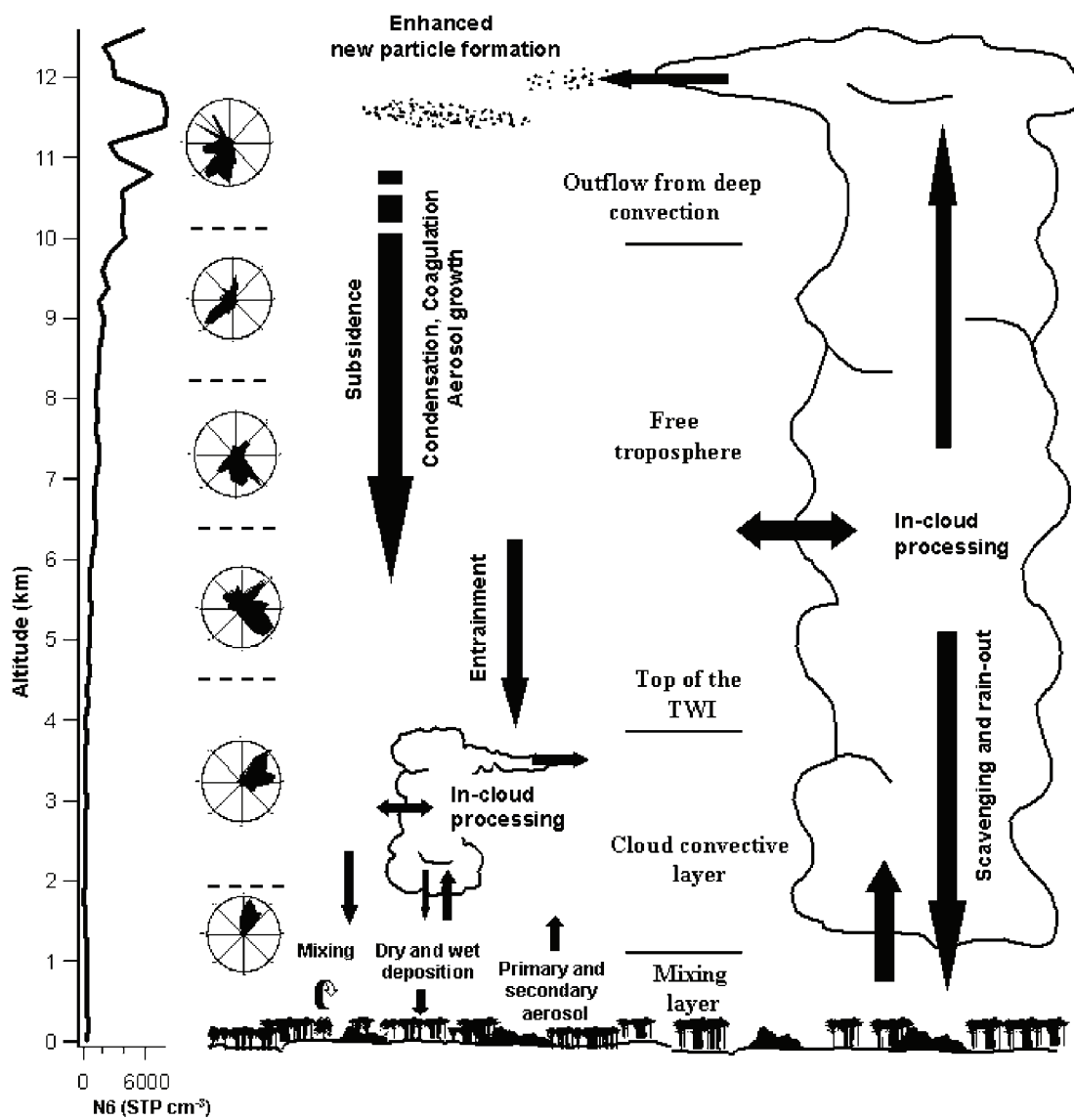


Figure 3

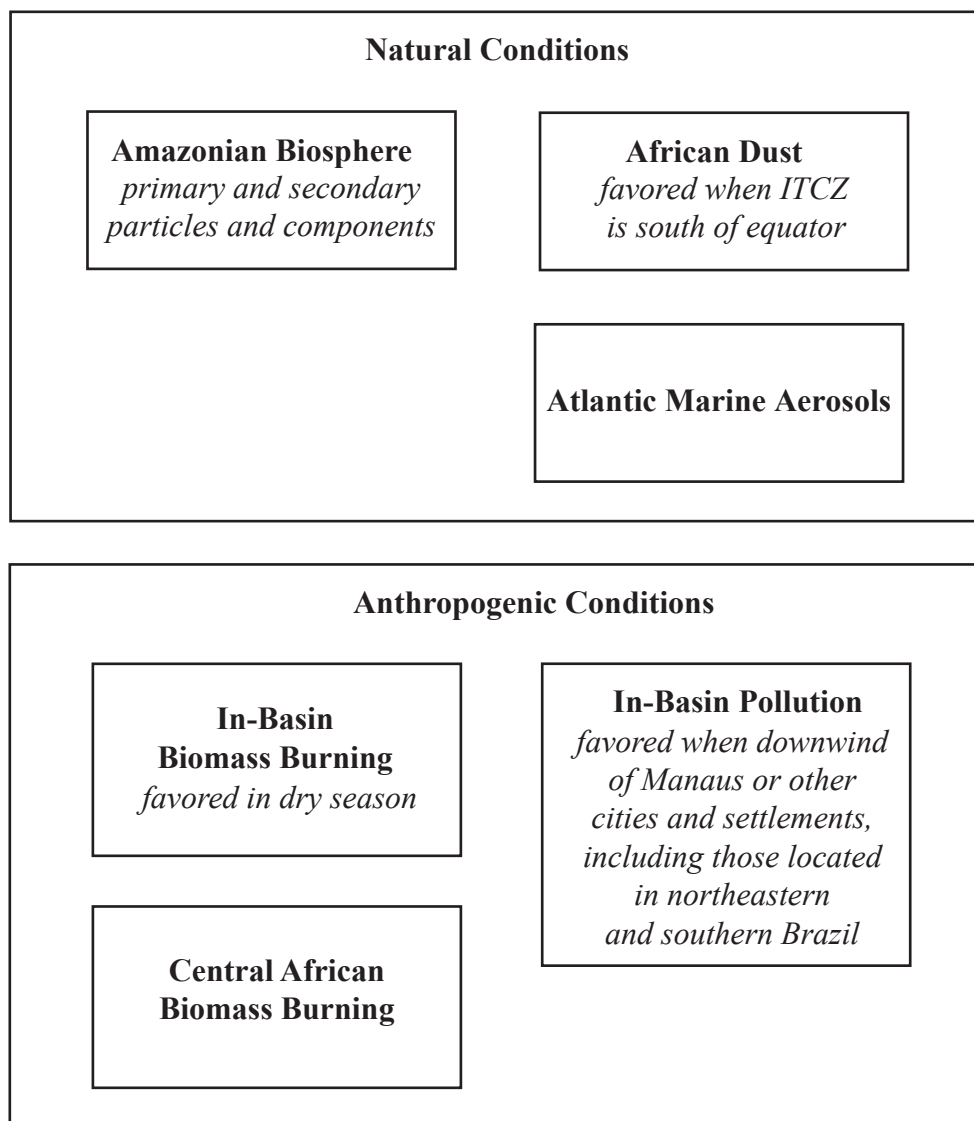


Figure 4

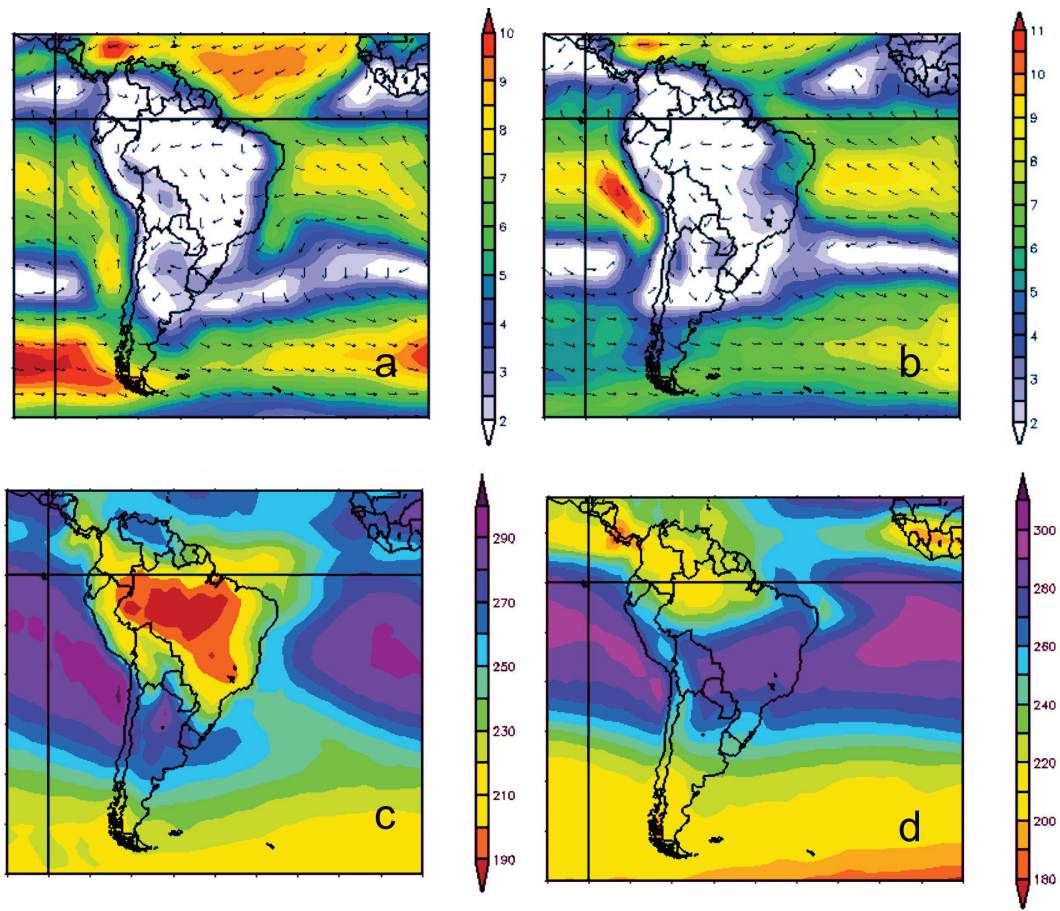


Figure 5

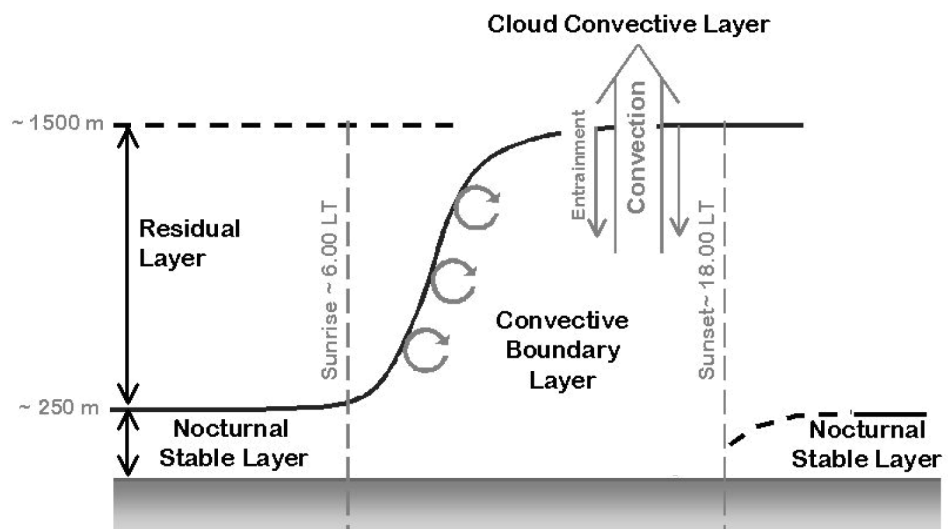


Figure 6

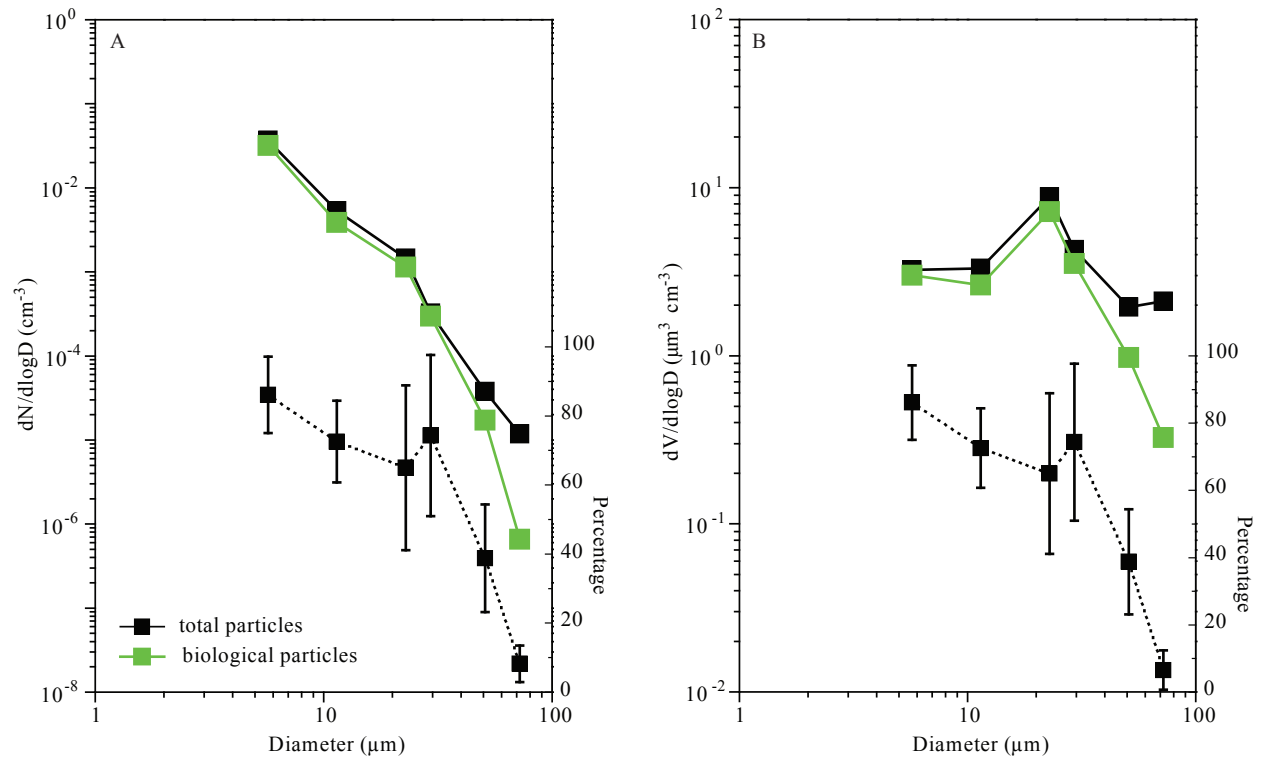


Figure 7

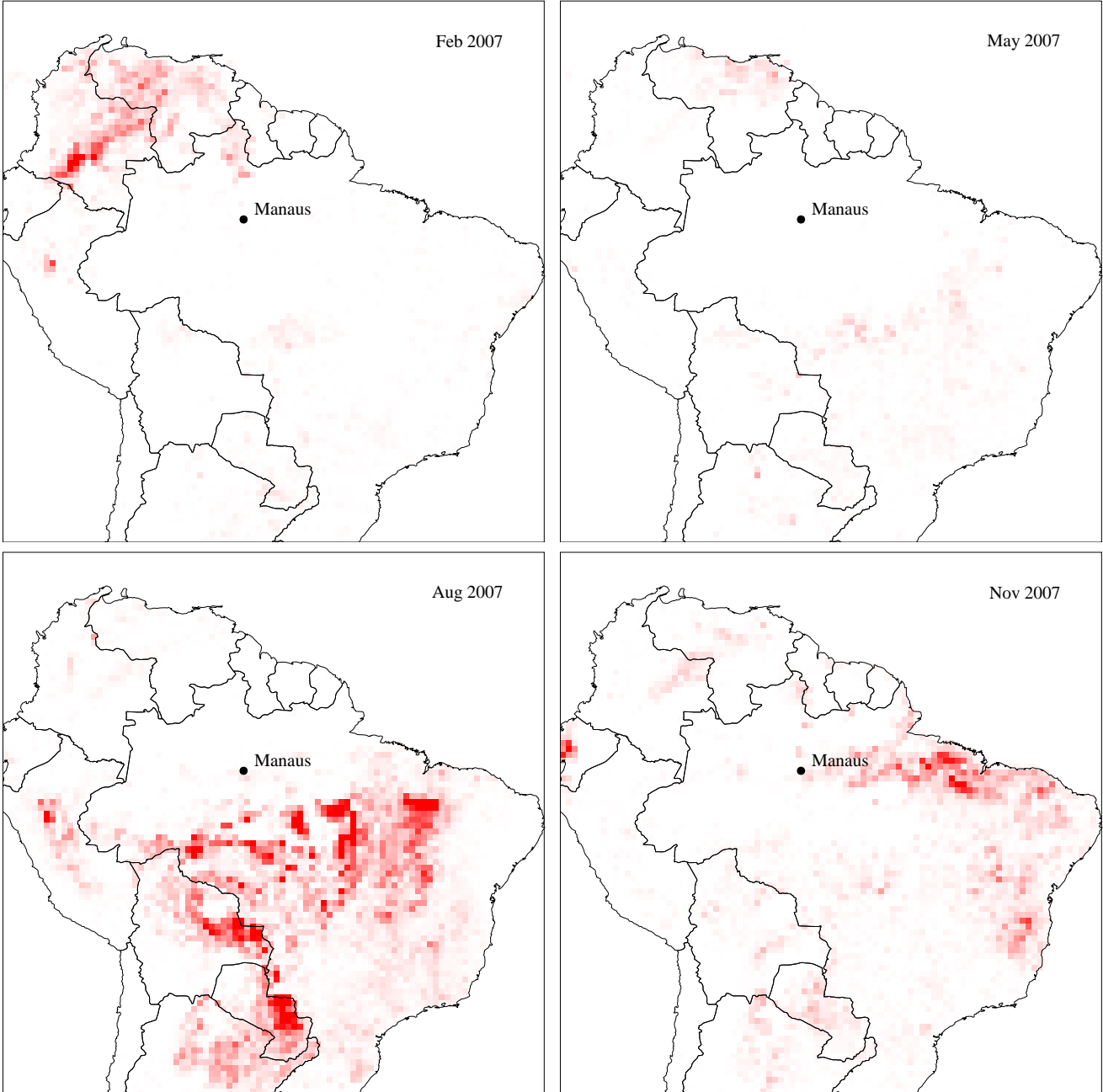


Figure 8

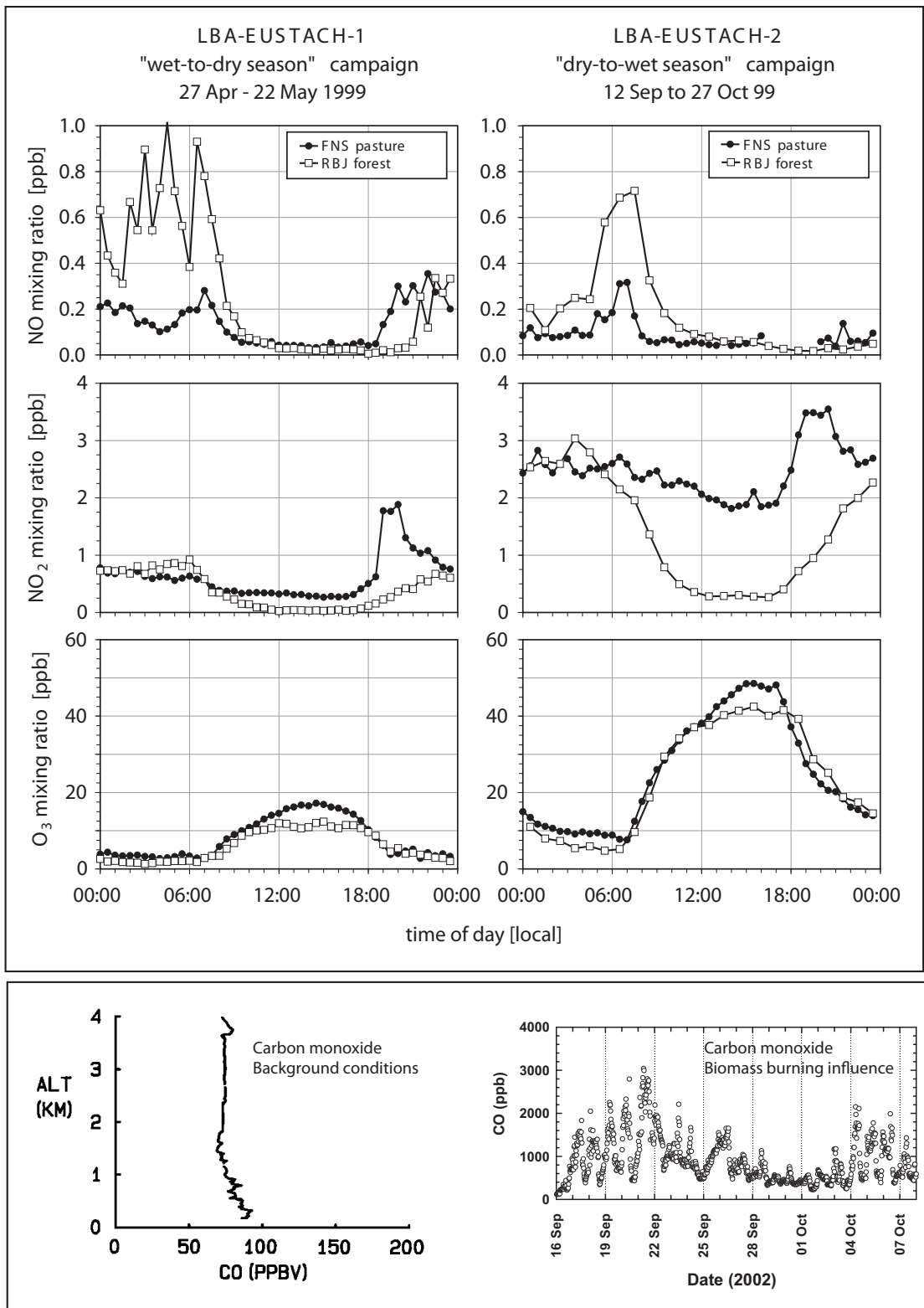


Figure 9

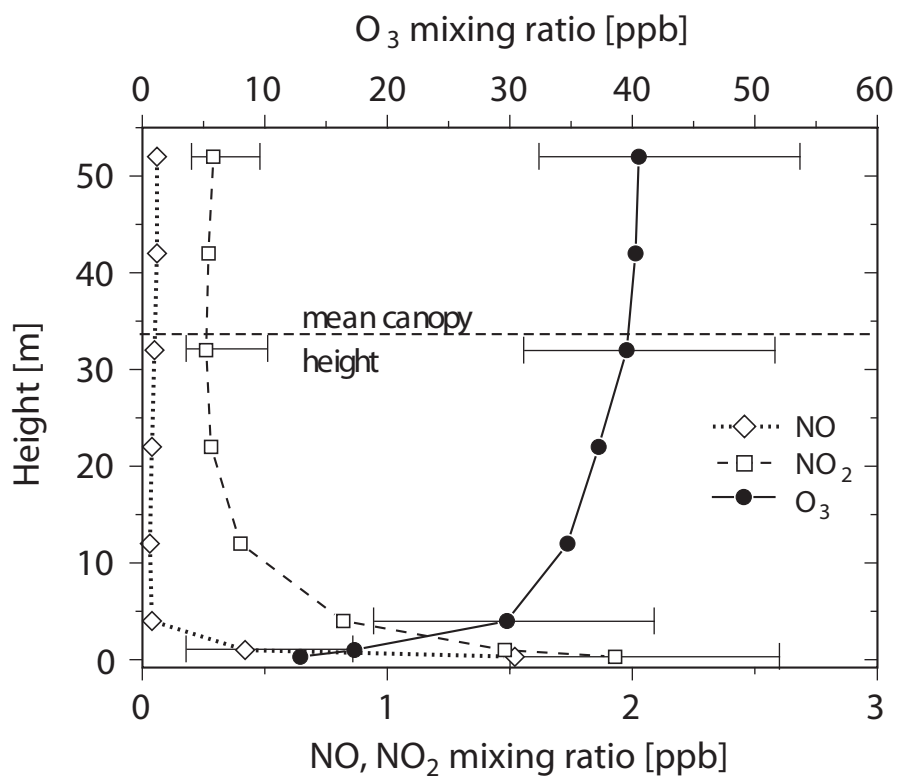


Figure 10

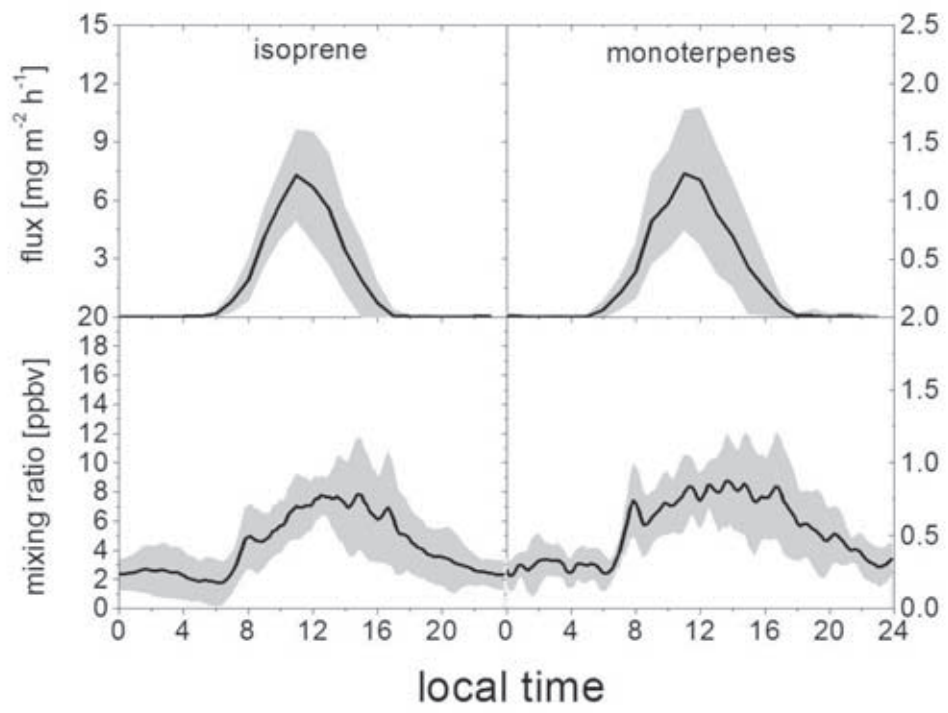


Figure 11

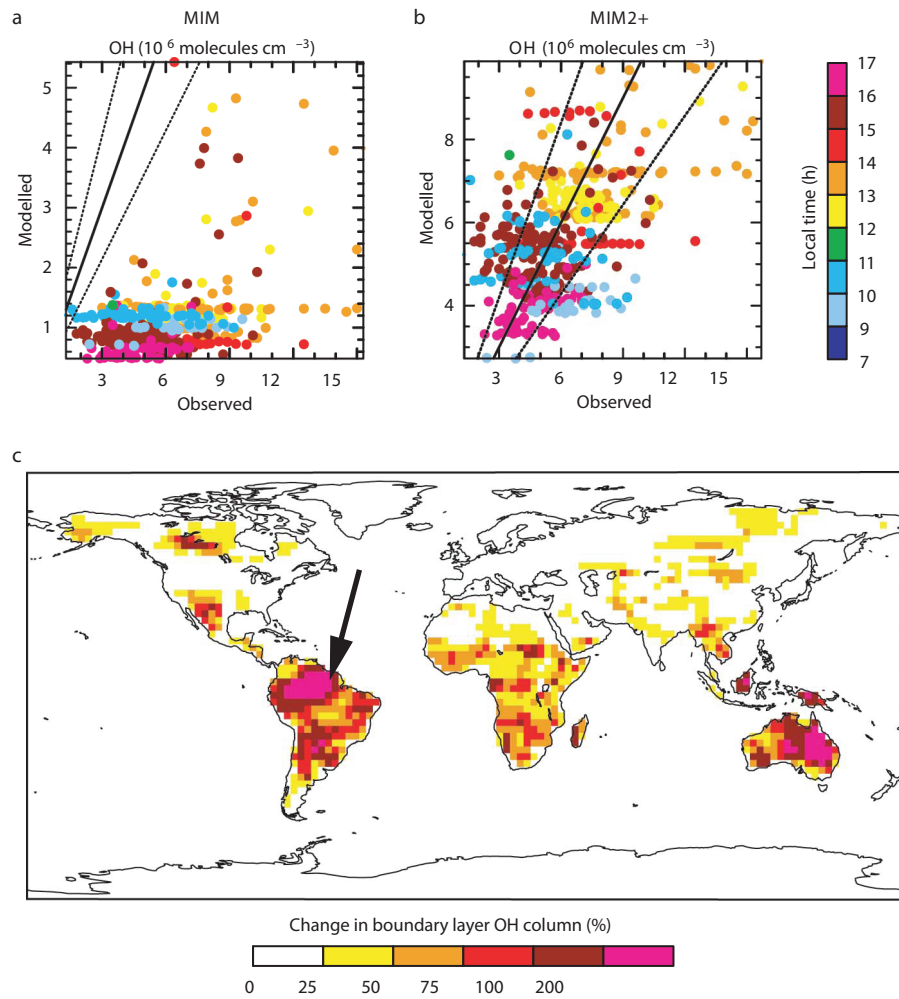


Figure 12

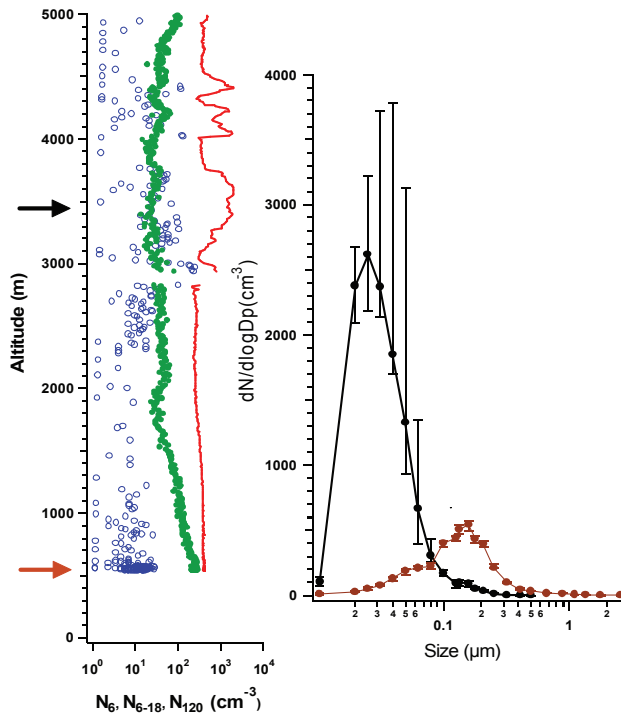


Figure 13

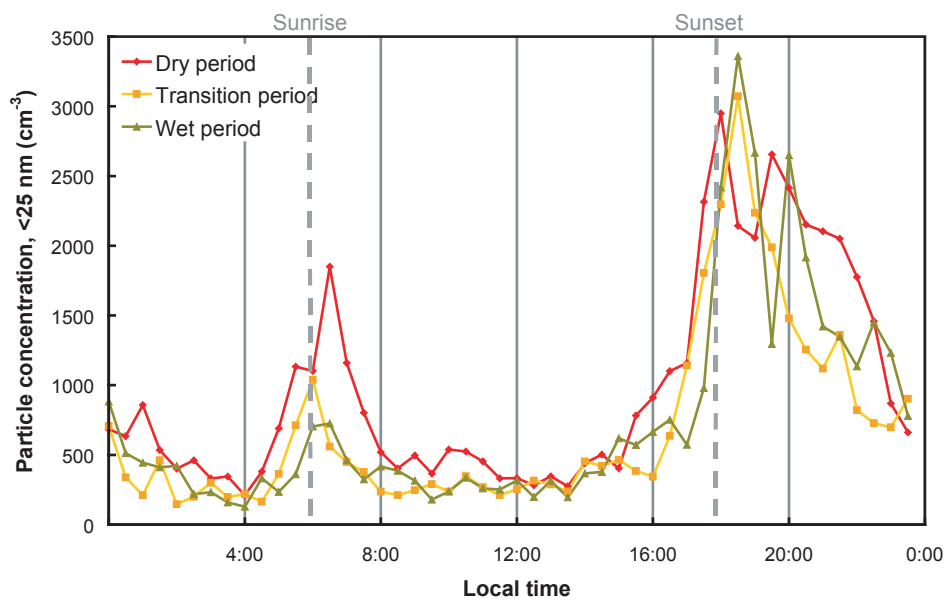


Figure 14

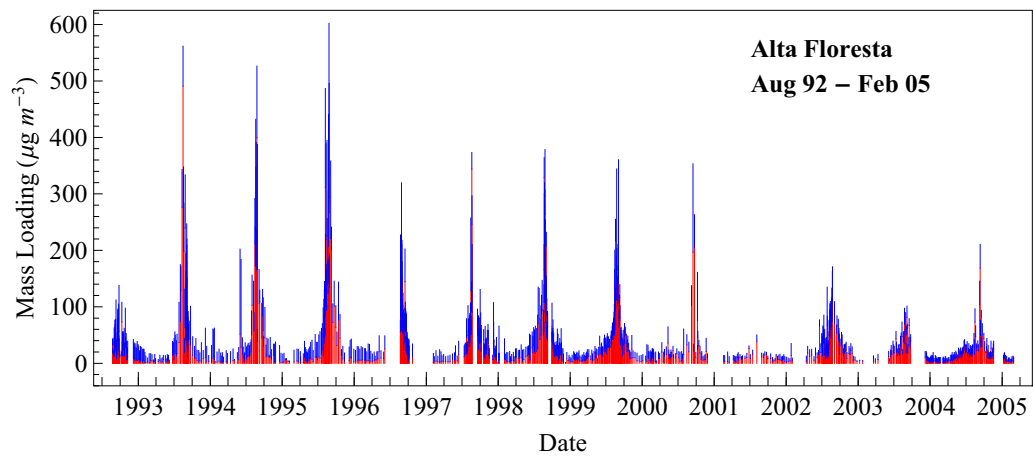
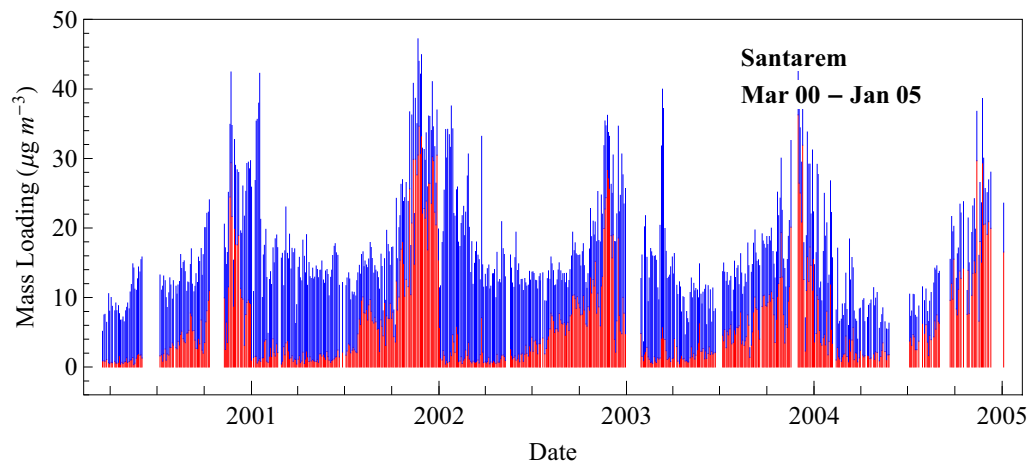
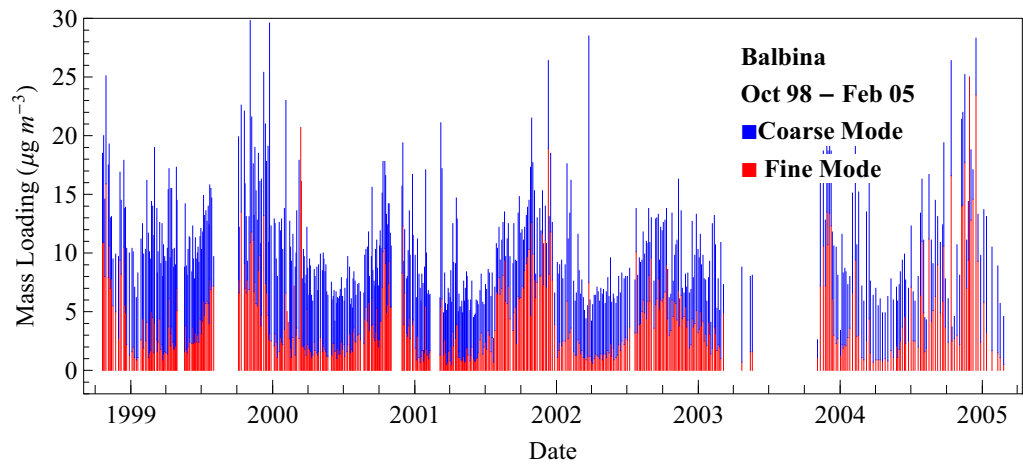


Figure 15

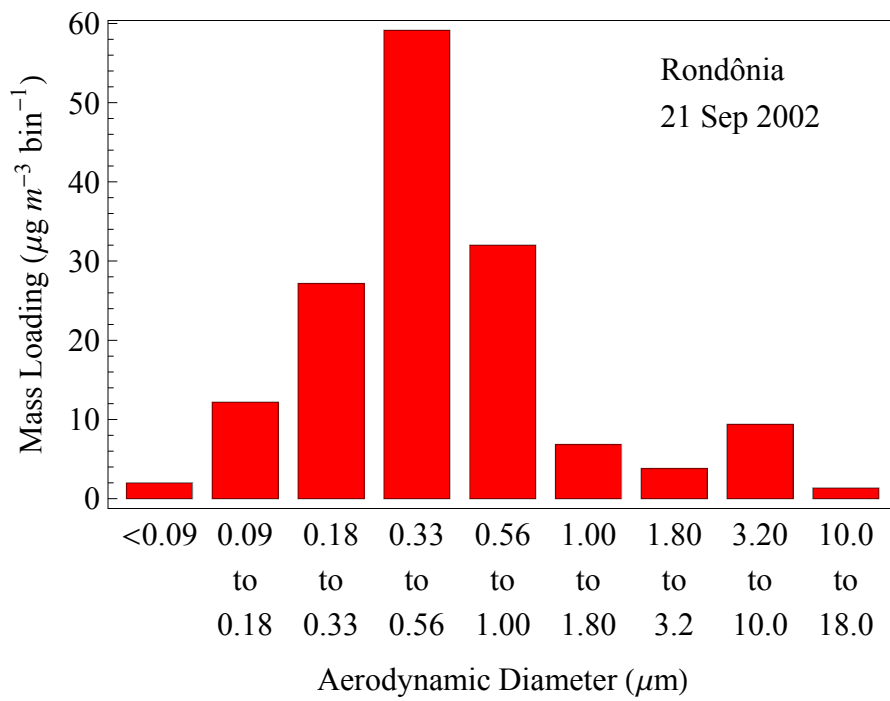
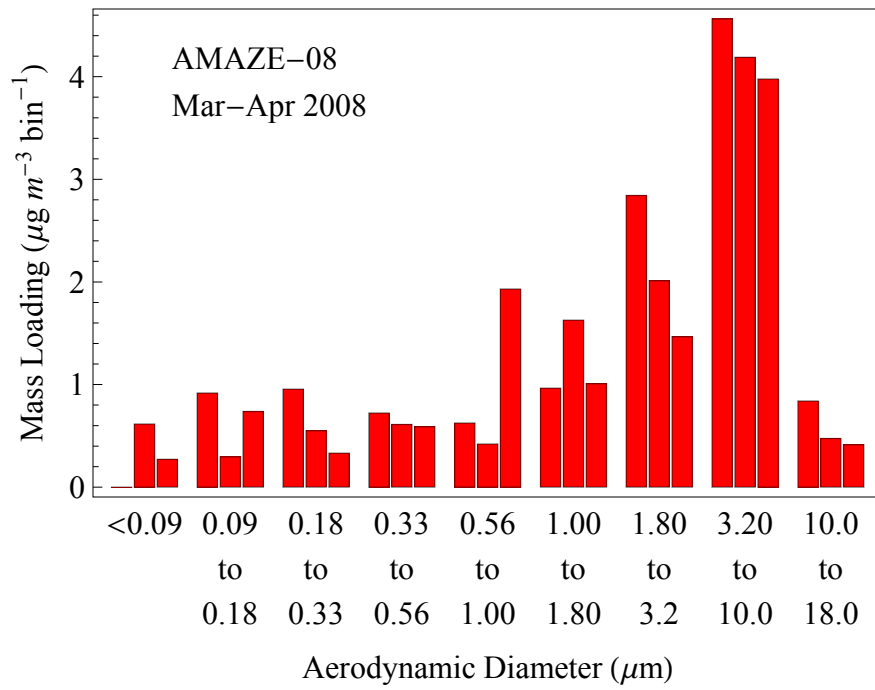


Figure 16

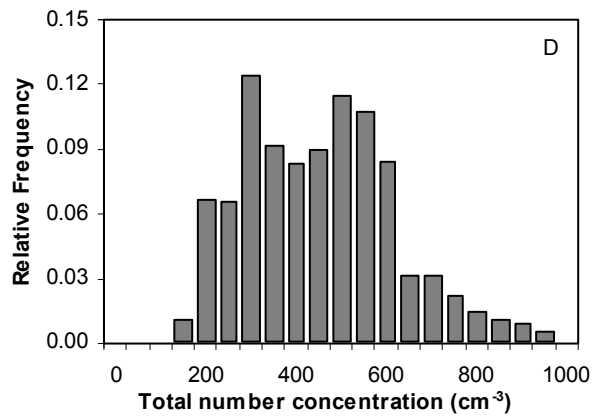
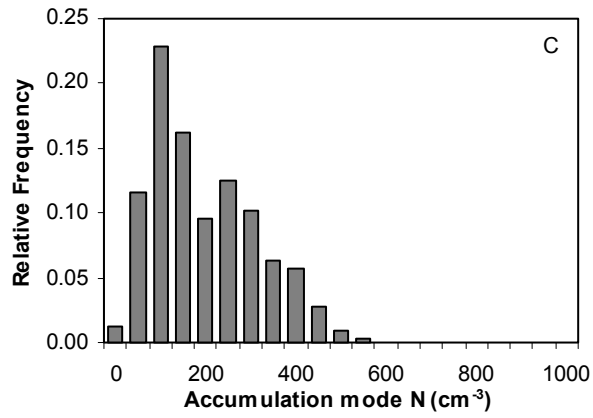
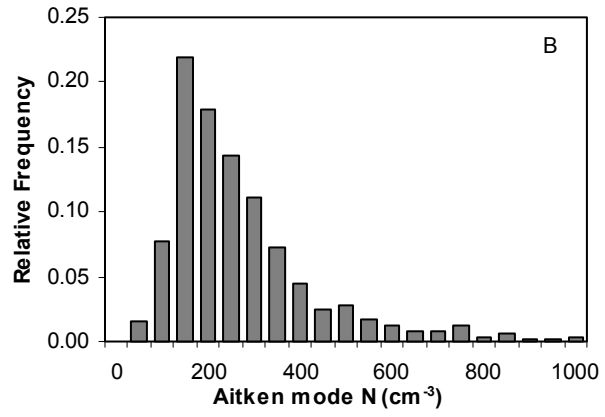
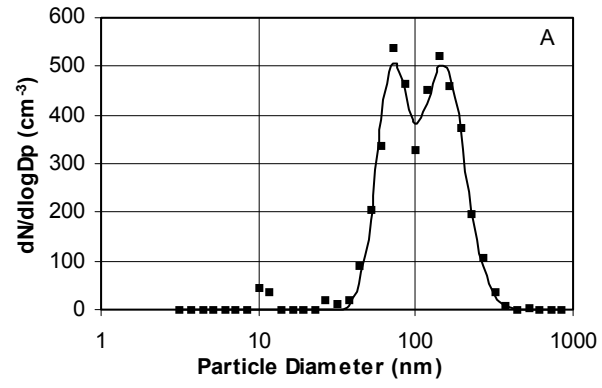


Figure 17

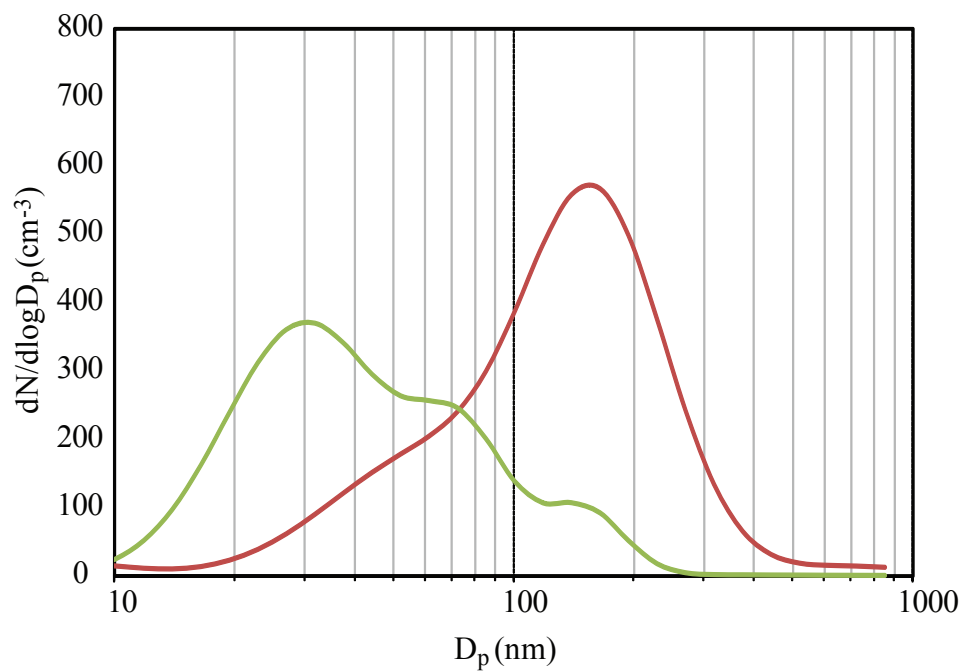
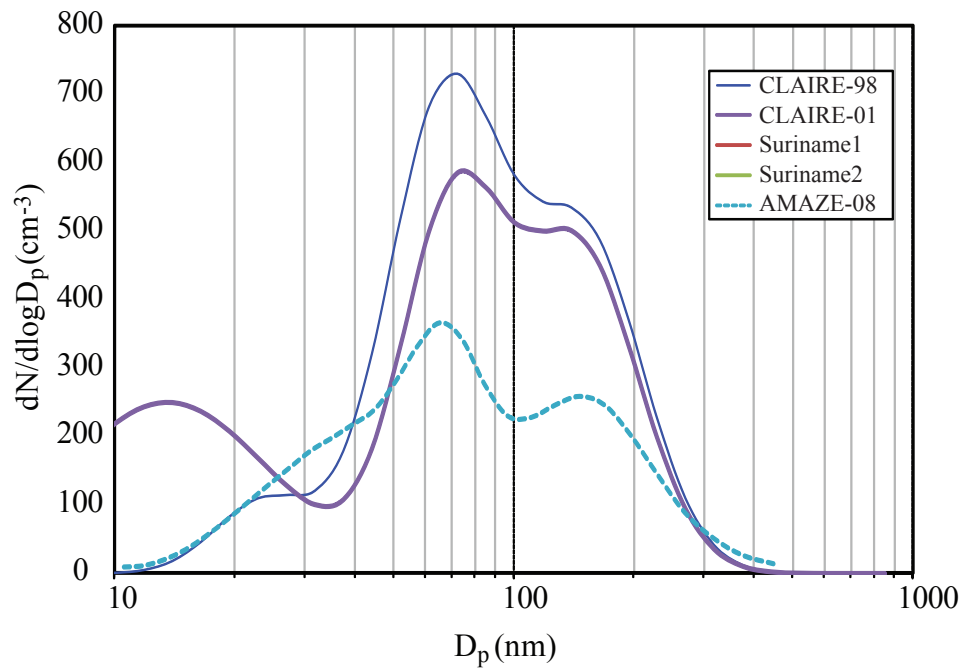


Figure 18

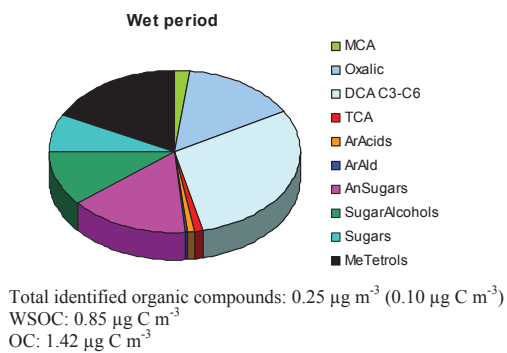
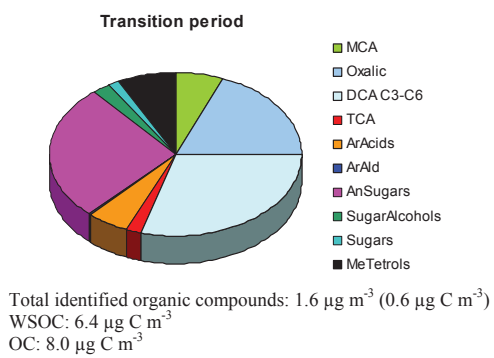
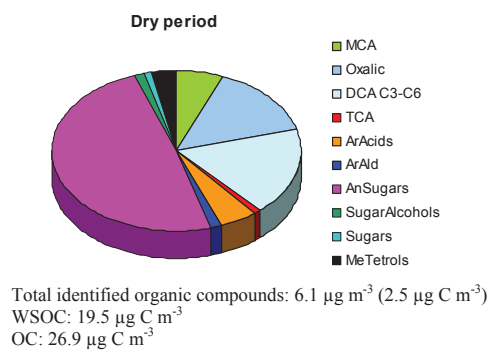


Figure 19

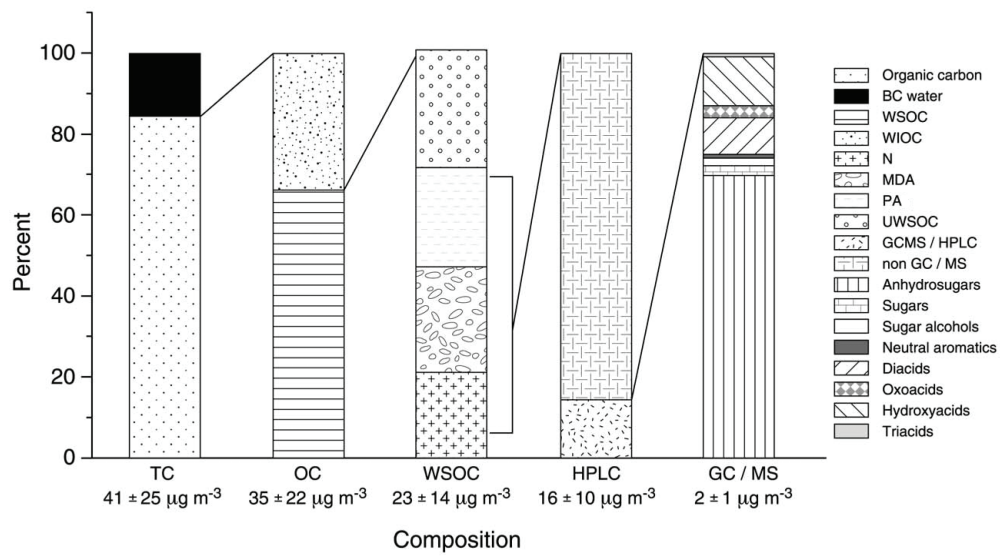


Figure 20

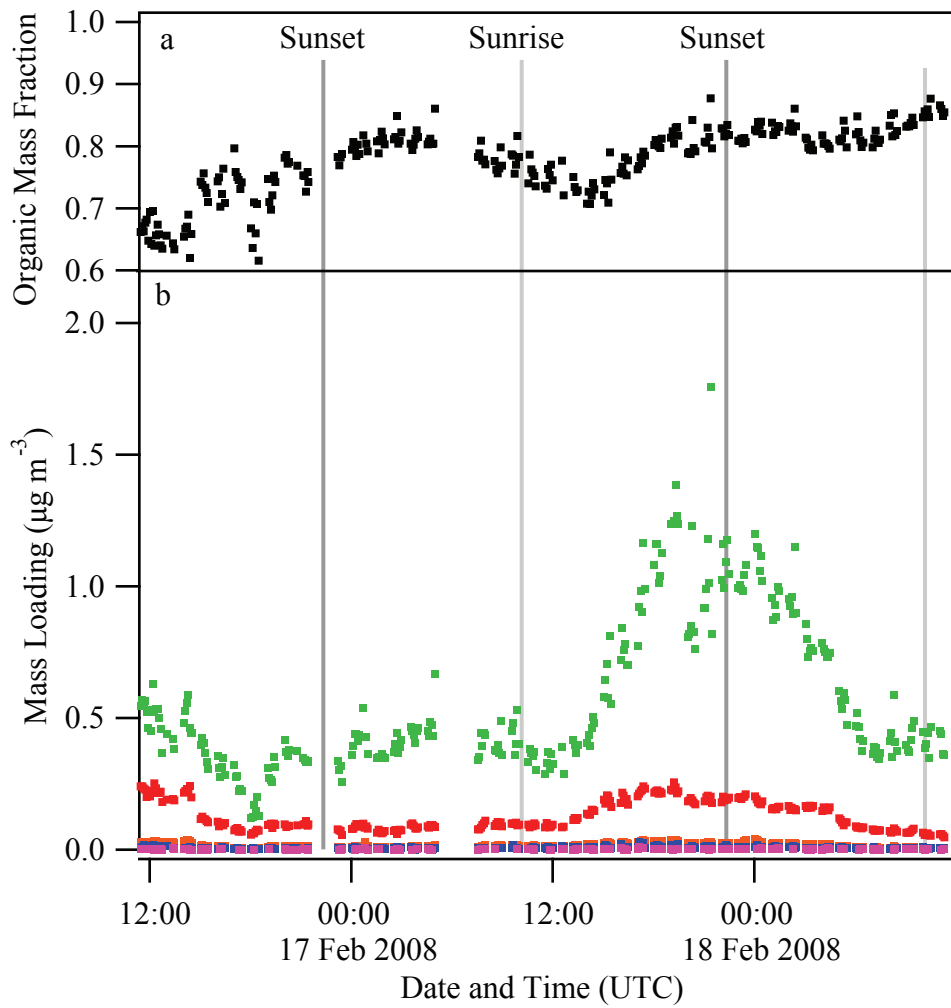


Figure 21

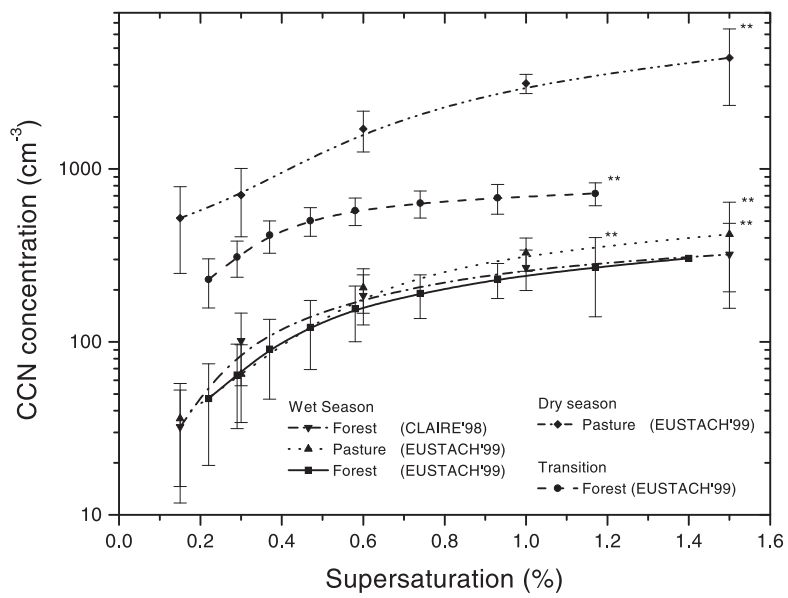


Figure 22

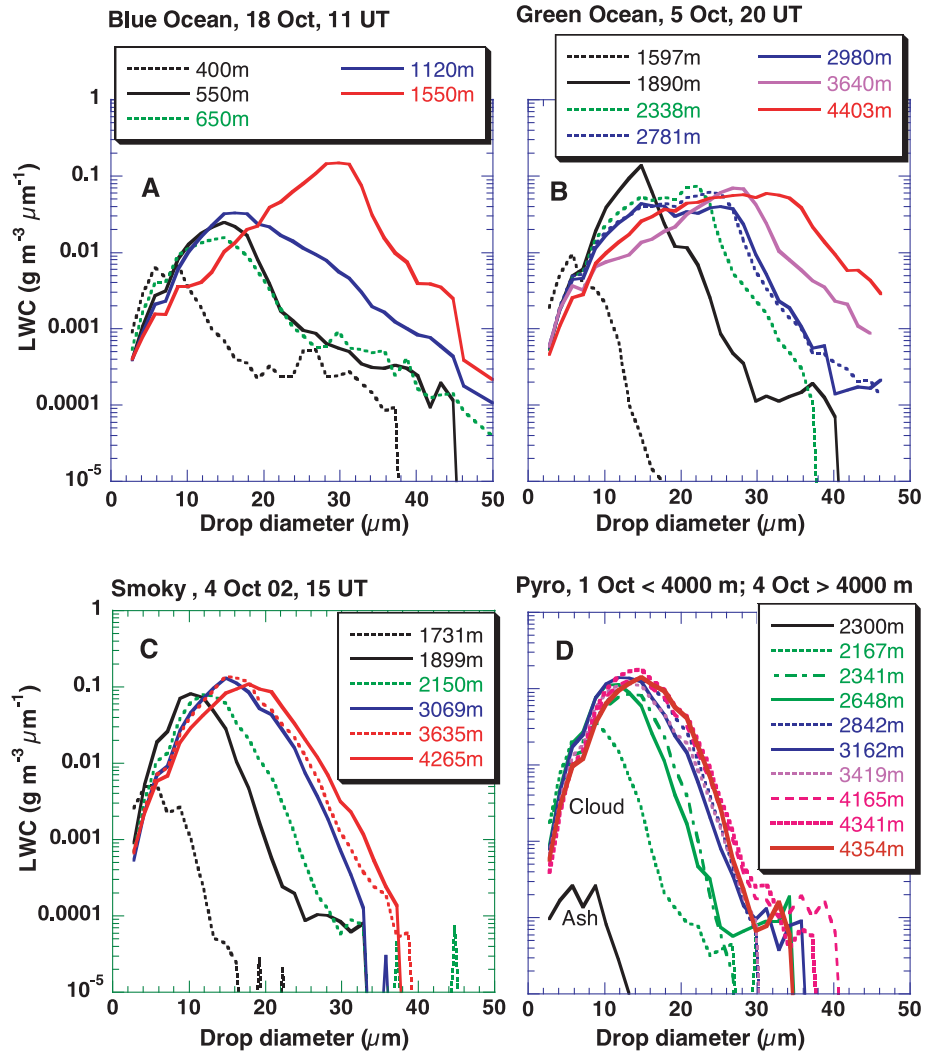


Figure 23

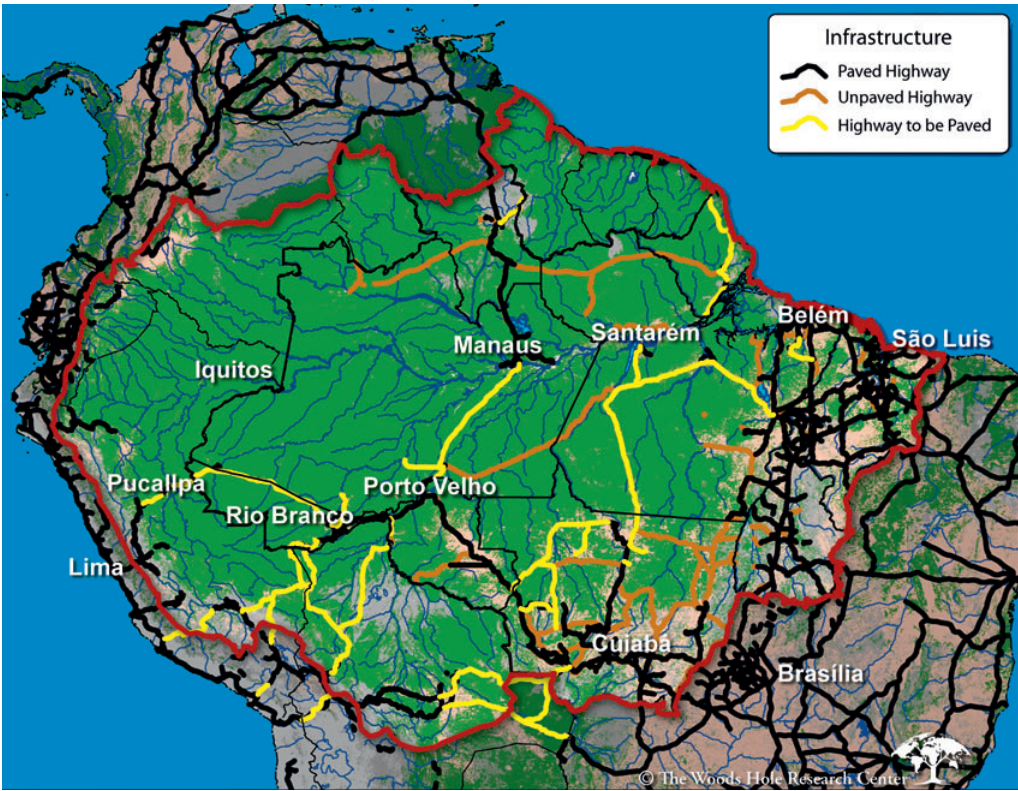


Figure 24

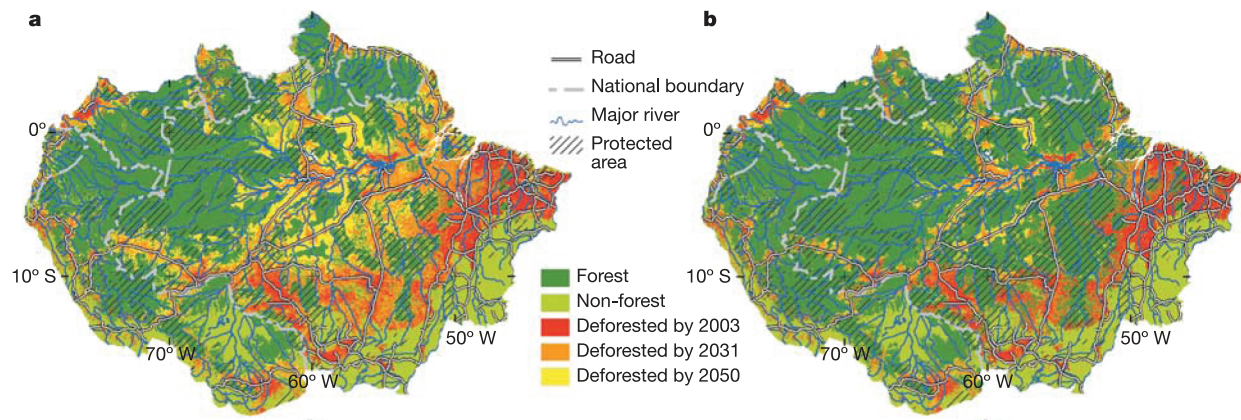


Figure 25