1	Impact of the Manaus urban plume on trace gas mixing ratios near the				
2	surface in the Am	azon Basin: Implications for the NO-NO ₂ -O ₃ photo-			
3	stationary state a	nd peroxy radical levels			
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35 Abstract

36 We measured the mixing ratios of NO, NO₂, O₃, and volatile organic carbon (VOC) as well as the aerosol light scattering coefficient on a boat platform cruising on rivers downwind of the 37 38 city of Manaus (Amazonas State, Brazil) in July 2001 (LBA-CLAIRE-2001). The dispersion and 39 impact of the Manaus plume was investigated by a combined analysis of ground based (boat 40 platform) and airborne trace gas and aerosol measurements as well as by meteorological mea-41 surements complemented by dispersion calculations (HYSPLIT model). For the cases with the 42 least anthropogenic influence (including a location in a so far unexplored region ~150 km west of Manaus on the Rio Manacapuru) the aerosol scattering coefficient, σ_s , was below 11 Mm⁻¹, 43 44 NO_x mixing ratios remained below 0.6 ppb, daytime O_3 mixing ratios were mostly below 20 ppb 45 and maximal isoprene mixing ratios were about 3 ppb in the afternoon. The photo-stationary 46 state (PSS) was not established for these cases, as indicated by values of the Leighton ratio, Φ , 47 well above unity. A change of the synoptic wind direction from east-northeast to south-southeast 48 in the afternoon, due to the influence of river breeze systems and other thermally driven mesos-49 cale circulations, often caused a substantial increase of σ_s and trace gas mixing ratios (about 50 threefold for σ_s , fivefold for NO_x and twofold for O₃), which was associated with the arrival of 51 the Manaus pollution plume at the boat location. The ratio Φ reached unity within its uncertainty 52 range at NO_x mixing ratios of about 3 ppb, indicating "steady state" conditions in cases when 53 radiation variations, dry deposition, emissions, and reactions mostly involving peroxy radicals 54 (XO₂) played a minor role. The median midday/afternoon XO₂ mixing ratios estimated using the PSS method range from 90 to 120 ppt for the remote cases ($\sigma_s < 11 \text{ Mm}^{-1}$ and NO_x < 0.6 ppb), 55 56 while for the polluted cases our estimates are 15 to 60 ppt. These values are within the range of 57 XO₂ estimated by the atmospheric chemistry box model CAABA/MECCA-3.0.

58 **1. Introduction**

The chemistry of nitrogen oxides (NO_x, the sum of nitrogen monoxide, NO, and nitrogen dioxide, NO₂) within the planetary boundary layer was studied in detail during the last decades in many regions of the globe. It is well known that NO_x acts as a key catalyst in the formation of tropospheric ozone (O₃) [*Crutzen and Lelieveld*, 2001]. In the absence of competing reactions, a dynamic equilibrium between NO, NO₂ and O₃ (also called photo-stationary state) may be established during daytime in the troposphere:

65
$$\operatorname{NO}_2 + \operatorname{hv} (\lambda < 420 \text{ nm}) \rightarrow \operatorname{NO} + \operatorname{O}({}^3\mathrm{P})$$
 R1

66

$$O(^{3}P) + O_{2} + M \rightarrow O_{3} + M$$
 R2

$$67 \qquad \text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2 \qquad \text{R3}$$

68 R2 is much faster than R1, and therefore R1 and R3 determine the overall conversion rate.

69 The photo-stationary state (PSS) parameter, or Leighton ratio, Φ [*Leighton*, 1961], is defined by:

70
$$\Phi = \frac{j(NO_2) \times [NO_2]}{k_3 \times [O_3] \times [NO]}$$
(1)

where $j(NO_2)$ is the photolysis frequency of NO_2 and k_3 is the rate constant of R3 [*Atkinson et al.*, 2004]. When Φ is unity, no additional O_3 is produced and the reactions R1-R3 represent a null cycle. The PSS approach is only valid under "steady state" conditions, i.e., in the absence of (a) rapid changes of light intensity and fluctuations of mixing ratios, (b) significant influence of dry deposition or fresh emissions of the involved compounds, and (c) other competing reactions. Particularly, the oxidation of NO by peroxy radicals constitutes a major net O₃ production pathway in the troposphere [e.g., *Warneck*, 2000]:

78
$$HO_2 + NO \rightarrow NO_2 + HO$$
 R4

79	$RO_2 + NO \rightarrow NO_2 + RO$ R5	
80	where R is an organic functional group (e.g., CH ₃ or C ₂ H ₅). HO ₂ and RO ₂ radicals (their	sum he-
81	reafter referred to as XO ₂) are formed from the oxidation of carbon monoxide (CO) and	hydro-
82	carbons by OH radicals. Most previous studies reported that PSS is readily established up	nder pol-
83	luted (high-NO _x) conditions [Carpenter et al., 1998; Griffin et al., 2007; Parrish et al., 1	986;
84	Volz-Thomas et al., 2003]. It was found, however, that under cleaner (low-NO _x) condition	ons, XO ₂
85	may substantially perturb PSS, resulting in values of $\Phi >> 1$ and measured NO ₂ mixing	ratios
86	exceeding those predicted by eq. (1) [Bakwin et al., 1994; Hauglustaine et al., 1996; Rid	ley et
87	al., 1992]. Some authors [e.g., Mannschreck et al., 2004; Volz-Thomas et al., 2003] sugg	ested
88	that other unknown oxidation processes may cause an additional conversion of NO to NO	O ₂ lead-
89	ing to values of $\Phi >> 1$. This was recently confirmed by <i>Hosaynali Beygi et al.</i> [2011] for	or back-
90	ground conditions in the remote marine boundary layer.	
91	Only few studies have investigated the NO-NO ₂ -O ₃ triad in the Amazon Basin. Thes	se either
92	focused on the surface-atmosphere exchange fluxes of individual compounds [Bakwin et	al.,
93	1990; Ganzeveld et al., 2002; Gut et al., 2002; Jacob and Wofsy, 1990; Rummel et al., 20)02;
94	Rummel et al., 2007; Trebs et al., 2006], on airborne studies, and modeling activities [An	dreae et
95	al., 1988; Crutzen et al., 1985; Harriss et al., 1988; Kuhn et al., 2010; Lelieveld et al., 20)08;
96	Torres and Buchan, 1988]. In the Amazon, mixing ratios of the NO-NO ₂ -O ₃ triad are sul	ostantial-
97	ly enhanced during the dry (biomass burning) season from June to September. However,	the
98	plume of pollutants originating from the city of Manaus (currently about 1.7 million inha	ubitants)
99	is known to have an additional significant impact on atmospheric chemistry in the region	west of
100	Manaus [Kuhn et al., 2010]. Furthermore, high emissions of hydrocarbons have recently	been
101	shown to sustain high radical concentrations above the above the Brazilian rainforest clo	se to

102	Manaus [Kuhn et al., 2007] as well as above other tropical South American rainforest locations
103	[cf. Lelieveld et al., 2008; Martinez et al., 2010], thereby affecting the cycling of NO _x and the
104	production of O_3 in this region. The influence of anthropogenic activities and mixing conditions
105	on atmospheric chemistry within the boundary layer in the Amazon Basin is not well understood,
106	and no direct measurements of prevailing XO ₂ radical mixing ratios have been reported so far.
107	Likewise, an investigation of the PSS of the NO-NO ₂ -O ₃ triad has not yet been performed in this
108	region. A thorough understanding of the NO-NO ₂ -O ₃ chemistry and the role of peroxy radicals is
109	required for an assessment of future chemistry-climate interactions in tropical regions with in-
110	creasing anthropogenic impact [cf. Ganzeveld et al., 2010].
111	In this paper, we investigate the PSS for the first time at unique (including very remote)
112	measurement locations in the Amazon Basin, based on field measurements made during the
113	Large-Scale Biosphere-Atmosphere Experiment in Amazonia - Cooperative LBA Airborne Re-
114	gional Experiment (LBA-CLAIRE-2001). We used a boat platform cruising on rivers in the
115	downwind region west of Manaus, at a distances ranging from ~20-150 km. We sampled clean
116	air at remote locations as well as air masses affected by the Manaus plume, which had interacted
117	with the rain forest and water surfaces of rivers west of Manaus. We infer XO ₂ mixing ratios and
118	we show that the PSS approach is not valid in the low- NO_x regime of remote tropical air masses,
119	in agreement with results obtained in other remote regions of the globe. Furthermore, our analy-
120	sis provides information about the impact of anthropogenic emissions and mesoscale circulations
121	on the pristine tropical forest photochemistry and radical formation near the surface downwind
122	of Manaus.

123 **2. Experimental**

124 2.1 Description of measurement sites and implementation

125 The measurements were made within the framework of LBA-CLAIRE-2001 from 10 to 27 126 July 2001 (wet-to-dry season transition) on a boat cruising downwind of the city of Manaus 127 (Figure 1). Manaus had a population of about 1.2 million human inhabitants in 2001; major 128 emission sources in and around Manaus comprise thermoelectric power plants, vehicles, saw 129 mills, charcoal kilns, brick factories and small-scale burnings [Kuhn et al., 2010]. 130 The average annual rainfall in Manaus is ~2200 mm with two marked seasons (November-131 May wet; June-October dry). The rainforest ecosystem characteristics of the Manaus area are de-132 scribed in detail by *Kuhn et al.* [2007]. The air flow is typically dominated by easterly trade 133 winds transporting humid oceanic air masses from the Atlantic more than 1000 km over the al-134 most undisturbed rain forest of the northeastern Amazon Basin, before reaching the urban area of 135 Manaus. 136 Three boat trips were performed on Rio Manacapuru (10-11 July) and on Rio Negro (18-20 137 and 25-27 July) (see Figure 1). During weekends the boat was located at the main station in Ma-138 nacapuru (Paraiso D' Angelo: 3.297°S, 60.604°W), where instruments were maintained and cali-139 brated. Two boat platforms were used during our experiment. The first boat was equipped with 140 an automatic weather station, and the trace gas and aerosol measuring instrumentation, while the 141 second boat carried a diesel generator providing a power of 15 kW for operation of the instru-142 ments. A power cable was connected to an electrical manifold onboard of the first boat and was 143 kept on the water surface using floating devices. The second boat was pulled by the first boat us-144 ing a 5 mm steel cable and was kept always downwind of the first boat to minimize the risk of

145 local contamination.

146 2.2 Meteorological measurements

147 Meteorological quantities were monitored with a time resolution of one minute on a mast at 148 a height of 8 m above the water surface. These comprised wind speed (cup anemometer 149 A101ML, Vector Instruments, UK), wind direction (wind vane, self-referencing, SRW1, Vector 150 Instruments, UK), solar global irradiance (pyranometer, LI-1000, LI-COR, USA), air tempera-151 ture and relative humidity (Hygromer® IN-1 & Pt100 sensor in aspirated housing, Rotronic 152 Messgeräte GmbH, Germany) and barometric pressure (pressure analog barometer, PTA 427A, 153 Vaisala, Finland). Rainfall was measured using a tipping rain gauge (ARG 100-EC, Campbell 154 Scientific, UK) on the edge of the boat. The position of the boat was tracked continuously using 155 a Global Positioning System unit (GPSHAND, ESYS, Berlin, Germany).

156 2.3 Trace gas and aerosol measurements

157 The inlets for trace gas and aerosol measurements were located at a level of 3 m above the 158 water surface. The reactive trace gases, nitric oxide (NO), nitrogen dioxide (NO₂) and ozone (O_3) 159 were measured with a time resolution of one minute using sensitive commercially available ana-160 lyzers located in the interior of the boat. NO was measured by red-filtered detection of the che-161 miluminescence produced during the reaction of NO and O₃ (CLD 780 TR, ECO Physics, Swit-162 zerland). Excess O₃ was continuously added in a pre-reaction chamber to account for the cross 163 sensitivity by other gases. NO₂ was photolytically converted to NO by exposure of the sample air 164 to a high-pressure xenon lamp (Tecan PLC 760 photolytic converter) and then detected by the 165 chemiluminescence analyzer. The analyzer was calibrated four times during the experiment (on 166 8, 13, 22 and 25 July 2001) by diluting a 5.32 ppm NO standard gas. The efficiency of the photo-167 lytic conversion of NO₂ to NO was 0.52, determined by a back titration procedure involving the

reaction of O₃ with NO using a gas phase titration system (SYCOS K-GPT, ANSYCO GmbH,
Germany).

170 The climatic conditions in the Amazon Basin and the absence of air conditioning on the boat 171 caused higher detection limits and uncertainties of the CLD 780 TR analyzer than typical for 172 controlled conditions [cf. Mannschreck et al., 2004; Volz-Thomas et al., 2003]. The limit of de-173 tection (LOD) (2σ-definition, 10 min averages) for NO and NO₂ determined during the experi-174 ment by sampling synthetic (zero) air that had passed an active charcoal filter were 75 ppt and 175 135 ppt, respectively. The correction applied for the systematic offset of the CLD 780 TR ana-176 lyzer determined when sampling zero-air was on average 50 ppt. This value was confirmed by 177 sampling clean air with ambient NO levels near zero during nighttime at the remote locations 178 [cf. Kley and McFarland, 1980]. The precision of the chemiluminescence analyzer determined 179 from the signal noise during the calibrations was on average 4% (2σ). The variation of the spans 180 of the individual calibrations made on the boat platform was 20%. The influence of ambient 181 temperature and humidity variations on the sensitivity of the chemiluminescence analyzer are about 1.3% K⁻¹ and -0.27% per hPa H₂O [see Volz-Thomas et al., 2003]. The temperature in the 182 183 boat ranged from 25 to 35°C. Since the calibrations were made at temperatures covering this 184 range, the temperature effect is included in the span variation stated above. Data were not cor-185 rected for the humidity effect since it applies to both NO and NO₂ measurements, and would thus 186 not affect the value of Φ (see eq. 1).

187 O_3 was measured with a UV absorption analyzer (Thermo Instruments, TE49C, USA), 188 which has an accuracy of 5% and a precision of 1% at mixing ratios ≥ 20 ppb. The offset of the 189 analyzer was checked by sampling synthetic air that had passed an active charcoal filter. Grab 190 samples for the determination of VOCs (50 min sampling interval) were collected on fused-

191 silica-lined stainless steel cartridges (89 mm length, 5.33 mm I.D., Silicosteel, Restek, USA) 192 with 2-bed graphitic carbon adsorbents (130 mg Carbograph 1, followed by 130 mg Carbograph 193 5: Lara s.r.l., Rome, Italy) using customized constant-flow sampler devices (*courtesy* Jim Green-194 berg, NCAR). The samples were analyzed off-line in the laboratory using a thermal desorption 195 gas chromatograph with a flame ionization detector (GC-FID) as described in *Kuhn et al.* [2002, 196 2004]. Calibration was accomplished by different gaseous standards containing isoprene, several 197 n-alkanes, methyl vinyl ketone (MVK), and methacrolein (MACR). The detection limit was de-198 termined from the standard deviation of the blank values and was typically 30 ppt for isoprene 199 and 10 ppt for monoterpenes. Overall uncertainties were 10 % for isoprene at 1 ppb and ranged 200 from 5 to 30 % at 100 ppt for monoterpenes [cf. Kuhn et al., 2007]. Additionally, aerosol light-201 scattering measurements were made with a single-wavelength ($\lambda = 545$ nm) nephelometer (Mod-202 el M903, Radiance Research, USA) at ambient RH.

203 2.4 Companion measurements

204 Kuhn et al. [2010] reported on airborne measurements of trace gases and aerosols in the re-205 gion west of Manaus, which were part of LBA-CLAIRE-2001. In this study, we use data (O₃ and 206 aerosol number concentrations) measured during crosswind transects in the urban outflow at a 207 longitude of ~ 60.41°W (~ 43 km downwind of Manaus) on 19 July (Flight #18: 10:00-14:00 LT, 208 Flight #19: 15:45-18:00 LT). O₃ was measured with a time resolution of 10 s using the same type 209 of instrument as on the boat platform and the aerosol number concentrations (CN) were meas-210 ured at 1 Hz time resolution with a condensation particle counter (CPC, Model 3762 and 3010, 211 TSI, USA) [cf. Kuhn et al., 2010].

In addition, meteorological measurements (temperature, relative humidity, wind speed, wind direction) were made on a 52-m high walk-up scaffolding tower (K34, ca. 20 m above the forest

214 canopy) [see *Kuhn et al.*, 2007]. The tower is located in the National Institute of Amazon Re-

215 search (INPA) forest management area (Reserva Biológica do Cuieiras) in undisturbed primary

forest about 60 km NNW of Manaus (2.6091°S, 60.2092°W) (see Figure 1).

217 2.5 Data evaluation

218 The one-minute time series of all measured quantities on the boat platform were combined 219 into arithmetic averages for 10-minute intervals. However, as recommended by Yang et al. 220 [2004], geometric averages of all input quantities were used for the calculation of Φ and XO₂ 221 mixing ratios to minimize the possibility of bias caused by measurement imprecision. Data were 222 excluded from our analysis when the boat was moving or when pollution from the diesel genera-223 tor was present. About 43% of the remaining numbers of data points were below the detection 224 limit for NO (6% for NO₂). These data were not eliminated from the time series, but are treated 225 with caution for the PSS calculations (see below).

The PSS approach should not be applied under "non-steady state" conditions, i.e., when rapid changes of light intensity or mixing ratios occur. In order to exclude these cases, we calculated the time required to establish the photo-stationary state, τ_{PSS} (1/e-definition), according to:

229
$$\tau_{PSS} = \frac{1}{j(NO_2) + k_3 \times [O_3]}$$
(2)

The calculated τ_{PSS} values ranged between 60 and 300 s during daytime. When analyzing the one-minute time series we found that trace gas mixing ratios typically remained fairly constant within these short time periods. Similar to previous studies [e.g., *Yang et al.*, 2004] we ensured adequate stability of *j*(NO₂) by excluding data from the PSS evaluation during periods when the standard deviation of the solar global irradiance of the 10-min averaging interval was higher than 5% of the average value. Since radiation also strongly varies during sunrise and sunset, these periods were automatically removed from the PSS evaluation based on this criterion. Thus, ouranalysis is confined to clear-sky periods with low radiation fluctuations (see below).

Moreover, we estimated the prevailing XO₂ mixing ratios under "steady state" conditions for cases when $\Phi > 1$, expressing their role in the conversion of NO to NO₂, by assuming that NO₂ is produced via R3, R4 and R5 and re-arranging eq. (1) to [*Carpenter et al.*, 1998; *Volz-Thomas et al.*, 2003]:

242
$$[XO_2] = \sum_i [R_iO_2] + [HO_2] = (\Phi - 1)\frac{k_3[O_3]}{k_{eff}}$$
(3)

Since the rate constants k_4 and k_5 are nearly identical [*Atkinson et al.*, 2004; 2006] we used an effective rate constant $k_4 \approx k_5 = k_{eff} = 3.6 \times 10^{-12} \text{ e}^{(270/\text{T})} \text{ cm}^3$ molecule⁻¹ s⁻¹ for our analysis. It should be noted that eq. (3) only holds in case no other processes convert NO to NO₂ (i.e., it ignores the potential role of e.g., NO₃ and halogens).

247 2.6 Measurement uncertainties

248 Photolytic converters which emit radiation in a very narrow spectral band, making the con-249 version of NO₂ to NO more specific (e.g., Blue-Light Converters) were unfortunately not yet 250 available in 2001. However, the high-pressure xenon lamp of the Tecan PLC 760 photolytic con-251 verter was equipped with an optical filter with a maximal transmittance at 360 nm. As discussed 252 in Ryerson et al. [2000] this minimizes interferences from other species (e.g., HONO). Under 253 clean conditions the HONO mixing ratio is about 50 ppt during daytime in the Amazon Basin 254 [Trebs et al., 2004]. The residence time of the air in the converter was 8 seconds (52 % conver-255 sion efficiency) and with $j(HONO) = 0.3 \times j(NO_2)$ the potential maximal HONO interference is 256 thus ~ 8 ppt. The thermal decomposition of PAN at 30°C for a residence time of 8 seconds is <257 1 %. Assuming an upper limit of 500 ppt, the maximal PAN interference would be \sim 3 ppt. In

addition, the zero-air measurement (see section 2.3) revealed no indication for a spurious NO₂

signal induced by irradiation of photolysis cell surfaces as described in *Kley and McFarland*

- 260 [1980]. The light stream in the photolysis cell was focused in a long narrow cylinder thereby mi-
- 261 nimizing irradiation of the chamber walls that may cause secondary reactions.

262 The uncertainty of the Leighton ratio can be estimated using Gaussian error propagation [cf. 263 Hauglustaine et al., 1996; Mannschreck et al., 2004]. An uncertainty of 20% was assigned to the 264 rate constant k_3 [see Atkinson et al., 2004]. Since we did not measure $j(NO_2)$, we applied an em-265 pirical parameterization (derived from in-situ measurements in the Amazon) from Trebs et al. 266 [2009] to estimate $i(NO_2)$ from the measured solar global irradiance. This method has a 2σ un-267 certainty of about 25%. An error of 5 % was assigned to the ozone mixing ratio. The uncertainty 268 of the NO and NO2 measurements was estimated from the statistical error and the systematic er-269 ror [cf. Yang et al., 2004]. The statistical error comprises the detection limit (75 ppt for NO and 270 135 ppt for NO₂) and the analyzer precision, 2σ) of 4 % (see section 2.3). The systematic error 271 from calibrations and potential artifacts and uncertainties that could not be determined directly 272 from our measurements (see above) was estimated to 20 %. The uncertainty increases with de-273 creasing signal to noise ratio and was < 30 % ([NO] > 0.8 ppb, [NO₂] > 1.4 ppb), 30 - 95 % $([NO] = 0.1 - 0.8 \text{ ppb}, [NO_2] = 0.2 - 1.4 \text{ ppb}) \text{ and } > 95 \% ([NO] < 0.1 \text{ ppb}, [NO_2] < 0.2 \text{ ppb}).$ 274 275 Hence, the estimated uncertainty of Φ (see section 3.4.2) is clearly dominated by the errors of the 276 NO_x measurements, in particular when mixing ratios approach the LOD. The uncertainty of es-277 timated XO₂ mixing ratios was also estimated using Gaussian error propagation. Considering 278 that mainly HO₂, CH₃O₂ and C₂H₅O₂ contribute to XO₂, an average uncertainty of 20 % was as-279 signed to the rate constant k_{eff} [see Atkinson et al., 2004; 2006].

280 3. Results and Discussion

281

Our dataset includes five days of remote measurements when relatively clean air masses 282 were sampled with only occasional anthropogenic influence, distinguishing three different cases 283 (Figure 1). These are (i.) 10-11 July (3.0324°S, 61.3445°W, "Lago Manacapuru", distance to 284 Manaus ~147 km, color-code blue), (ii.) 18-19 July (3.0948°S, 60.4102°W, distance to Manaus 285 ~43 km, color-code orange) and (iii.) 25 July (3.0316°S, 60.4736°W, distance to Manaus 286 \sim 50 km, color-code turquoise). In addition, we observed two distinct cases when air masses were 287 substantially influenced by anthropogenic pollution from the city of Manaus. These are (i.) 20 288 July (3.0922°S, 60.1968°W, distance to Manaus ~19 km, color-code violet) and (ii.) 26-27 July 289 (3.0185°S, 60.1725°W, distance to Manaus ~19 km, color-code magenta).

290 3.1 Meteorological conditions

291 The convective boundary layer (CBL) above the Amazonian rain forest typically reaches its 292 maximum height of 1000 to 1500 m at around 17:00 LT [Fisch et al., 2004; Martin et al., 1988]. 293 Radiative cooling starts after 17:00 LT and the thermodynamically stable stratified nocturnal 294 boundary layer develops to a final height of 100 to 300 m. Meteorological conditions at the boat 295 platform were mainly determined by background synoptic winds (northeasterly trade winds) and 296 the formation of thermally driven diurnal mesoscale circulations. Wind roses averaged for all 297 boat locations at the southern bank of the rivers (Figure 1) reveal prevailing winds from east-298 northeast during the night and morning (Figure 2a), while during the afternoon and evening 299 winds from south-southeast were dominating (Figure 2b). Both regimes are characterized by relatively high wind speeds (2 to 5 m s^{-1}). The transition between the two regimes typically oc-300 301 curred around 15:00 LT and 00:00 LT. A similar pattern had been observed previously by de

302 *Oliveira and Fitzjarrald* [1993] from measurements in an area north of the Rio Negro, about 60 303 km north of Manaus. They hypothesized that this circulation pattern can be attributed to a river 304 breeze caused by the thermal contrast between the Rio Negro and the land surface. They found 305 that the river breeze is asymmetric (i.e., not exactly perpendicular to the river) due to the influ-306 ence of the easterly trade winds.

307 This wind pattern was most distinct on 18/19 July. The HYSPLIT model [Draxler and Hess, 308 1998; meteorological REANALYSIS data set] provided supporting information on the dispersion 309 of air masses originating from Manaus. Wind vectors expressed as one-hour forward trajectories 310 starting at 3.0948°S, 60.4102°W on 18 July at 18:00 LT at different heights within the CBL 311 (Figure 3) reveal a southeasterly wind direction in the lower part of the boundary layer (< 300312 m), which agrees with our observations (Figure 4a). This was also experimentally confirmed by 313 results of radio soundings at the K34 tower (not shown). The wind direction turned towards 314 northeast with increasing height, indicating that the southeastern flow close to the ground was 315 generated by processes within the CBL (e.g., decoupling, local pressure gradients). On 18/19 Ju-316 ly the wind direction measured on the boat platform also compares quite well with the wind di-317 rection measured at the K34 tower about 58 km north-northeast of the boat position (see Figure 318 4a). This confirms the existence of a mesoscale circulation system. *Betts et al.* [2008] pointed out 319 that the generation of such mesoscale systems is not only caused by river breezes (depending on 320 the shape of the rivers and their position towards the background synoptic flow), but may also be 321 facilitated by other surface heterogeneities (topography, differences in roughness length between 322 water and land, different vegetation types, soil water availability). Results from the HYSPLIT 323 model for days outside our measurement period indicate that this wind system is not always pre-

vailing around the Manaus region, but was obviously the predominant pattern during the particu-lar days when our measurements took place.

We observed one significant exception from the typical wind pattern, which occured on 26 and 27 July, the only instance when the boat was located at the northern bank of the Rio Negro close to Manaus (Figure 1). A diel oscillation of the local wind perpendicular to the river was observed. The wind blew from the rain forest (north) to the Rio Negro river during nighttime and from the river (south) to the rainforest during daytime (Figure 4a), which we attribute to a symmetrical river breeze.

Typical daytime wind speeds measured on the boat were 2 to 5 m s⁻¹, and during nighttime the wind speed dropped to values between 0.5 and 2 m s⁻¹ (Figure 4a). The relative humidity (RH) typically ranged from 60% during daytime up to 90-100% during nighttime (Figure 4b). Daytime and nighttime temperatures ranged between 29-32°C and 23-26°C, respectively. The global irradiance featured a typical diel course with maximal values of 900-1000 W m⁻² during clear sky periods (Figure 4c). On most days, cloudy skies prevailed during the afternoon hours when rain showers and, occasionally, thunderstorms occurred.

339 3.2 Trace gas mixing ratios and aerosol concentrations

340 *3.2.1 Ozone (O₃)*

The O_3 mixing ratios reveal a typical diel cycle with significantly higher values during daytime (Figure 4c). For the remote cases on 10, 18 and 19 July a near twofold increase of the O_3 mixing ratios during the late afternoon indicates the influence of anthropogenic pollution (see below). Excluding these events, the measured daytime O_3 mixing ratios for the remote cases (10-11, 18-19 and 25 July) were on average 21 ± 2 ppb. For comparison, *Kuhn et al.* [2010] measured background O_3 mixing ratios of 18 to 24 ppb downwind of Manaus during the CLAIRE-

2001 airborne measurements. Wet season median daytime O₃ values measured by *Rummel et al.*[2007] over a remote rainforest canopy were around 13 ppb, which is substantially lower than
our values.

350 Observed daytime O₃ mixing ratios for the polluted cases (20, 26-27 July) were on average 351 31 ± 14 ppb with maximal values of 60 ppb on 26 July at a distance of 19 km downwind of Ma-352 naus. Kuhn et al. [2010] observed maximal O₃ mixing ratios of 65 ppb at 100 km distance from 353 Manaus in photochemically aged air masses at altitudes of several hundred meters above ground. 354 Furthermore, we observed major differences in nighttime O₃ mixing ratios depending on 355 whether or not the air had a riverine or a tropical forest footprint (Figure 1 and Figure 4a,c). 356 Nighttime O_3 mixing ratios were often considerably higher (~10 ppb) than measured above a 357 tropical rain forest (< 5 ppb, [see e.g., *Rummel et al.*, 2007]). This can be explained by (i.) the 358 less stable nocturnal thermal stratification over (still warm) water surfaces than over land, facili-359 tating intermittent downward transport of O₃ from aloft and (ii.) depending on the residence time 360 of the air masses over the river also by a smaller efficiency of O₃ uptake by a water surface than 361 by vegetation [Ganzeveld et al., 2009; Wesely et al., 1981]. One exception was found during 362 nighttime on 26/27 July (symmetric river breeze, see section 3.1), when air masses with O₃ well 363 below 5 ppb were advected directly from the rain forest. Ozone dry deposition and removal 364 through reaction with soil-biogenic NO emissions accumulated in the forest trunk space 365 [*Rummel*, 2005] resulted in much lower nighttime O₃ levels than observed at the other locations. 366 3.2.2 Nitric oxide (NO) and nitrogen dioxide (NO₂)

The mixing ratios of NO for the remote cases on 10-11, 18-19 and 25 July were mostly lower than 500 ppt and sometimes fell below the detection limit of 75 ppt (Figure 4d). These values compare well with NO mixing ratios measured by *Bakwin et al.* [1990] and *Rummel* [2005] dur-

370 ing the Amazonian wet season within and above the rain forest canopy. The major NO source 371 under these remote conditions is biogenic emission from forest soils, potentially enhancing the 372 NO_x mixing ratios of sub-canopy air masses being sampled at our measurement location on the 373 river as a result of advection. Since the O_3 mixing ratios within and below the forest canopy are 374 typically much lower than above the forest [see *Rummel et al.*, 2007], such events are mostly ac-375 companied by a drop in the O₃ mixing ratio. Fresh pollution from biomass burning was some-376 times transported to the boat (e.g., 10 July after 6:00 LT, see Figure 4d). On 19 and 20 July the 377 wind direction changed abruptly from south-southeast to east-northeast around midnight (see 378 above), resulting in a NO mixing ratio increase to 0.6-1.0 ppb accompanied by a drop of the O_3 379 mixing ratio. The measured NO mixing ratios for the polluted cases (20, 26-27 July), were large-380 ly influenced by fresh pollution originating from anthropogenic activities close to Manaus. Dur-381 ing these events NO mixing ratios reached values of up to 4-10 ppb before midday (Figure 4d). 382 For the remote cases, the NO₂ mixing ratios were typically close to the detection limit of 383 135 ppt, e.g., on 10 July around midday or on 25 July during nighttime (Figure 4d). Very low 384 NO₂ mixing ratios (< 300 ppt) were also observed on 18 July from 00:00 to 15:00 LT at a dis-385 tance of ~43 km from Manaus on the Rio Negro. After that, the typical change of the wind direc-386 tion from east-northeast to south-southeast (Figure 2, 5a) resulted in an NO₂ increase to about 3 387 ppb while NO mixing ratios remained low, suggesting the arrival of an aged pollution plume (in-388 dicated also by an increase of O₃, see above) that lasted until the early morning of the next day. 389 A similar event occurred on 19 July with even higher NO₂ mixing ratios. 390 However, close to Manaus, on 26-27 July, NO₂ mixing ratios were typically highest before 391 midday reaching a maximum of 18 ppb in the morning of 26 July. Comparably high NO₂ mixing

ratios of about 16 ppb were also observed during the airborne measurements (410 m a.g.l.) on 15

July during the same time at nearly the same distance west of Manaus, and were attributed to the plume originating from power plants in the southern part of the city [*Kuhn et al.*, 2010].

395 *3.2.3 VOCs*

396 VOC mixing ratios and MVK/MACR ratios are presented for a remote (background) case 397 (10 July, distance to Manaus \sim 147 km) and for a polluted case close to Manaus (26 July, distance 398 to Manaus ~19 km) (see Figure 5a,b). As described in detail by Kuhn et al. [2007, 2010] for 399 tower-based and airborne measurements during LBA-CLAIRE-2001, isoprene was the dominant 400 compound observed, followed by the monoterpene α -pinene, both being emitted in a light-401 dependent manner [see Kuhn et al., 2004]. While for the remote case a typical diel pattern for 402 biogenic isoprene with daytime maxima of about 3 ppb was observed (Figure 5a), the polluted 403 case revealed somewhat lower mixing ratios during daytime (< 2 ppb, Figure 5b), which was at 404 least partly caused by lower radiation (Figure 4c). The contribution of anthropogenic isoprene 405 emissions related to industrial activities and in particular to vehicle exhaust [Borbon et al., 2001; 406 Duane et al., 2002; Reimann et al., 2000] might be a reasonable explanation for the higher varia-407 bility for the polluted case, particularly during nighttime. The diel cycle of α -pinene resembles 408 that of isoprene with maximal daytime mixing ratios of 0.2 ppb (not shown). 409 The isoprene degradation reactions yield the unsaturated C4 carbonyls MACR and MVK

Plus formaldehyde (CH₂O). High concentrations of these carbonyls as well as formaldehyde
have been found previously in the tropical atmosphere [*Kesselmeier et al.*, 2002]. They constituted a significant fraction of VOC in the remote background case as well as in the polluted case
close to Manaus during LBA-CLAIRE-2001 [*Kuhn et al.*, 2010]. The observed MVK/MACR
values close to (and exceeding) a value of two resemble those measured throughout the CBL
during LBA-CLAIRE-2001 [*Kuhn et al.*, 2010], and are indicative of OH-dominated isoprene

416 degradation [see Kuhn et al., 2007, 2010], where OH adds to one of the double bonds of iso-

417 prene, forming alkyl radicals that in turn add O₂ to form peroxy radicals (RO₂).

418 3.2.4 Aerosol scattering coefficient

419 For the remote cases, the lowest values of the aerosol scattering coefficient (σ_s) ranged from 3 to 10 Mm⁻¹ during daytime (Figure 4e). These values are comparable to the background air-420 421 borne measurements in the boundary layer made during LBA-CLAIRE-2001 by Kuhn et al. 422 [2010], as well as to those measured by Guyon et al. [2003] in southern Amazonia during the wet 423 season. Nighttime σ_s values were typically several times higher, which was most likely also re-424 lated to the change of the wind direction in the late afternoon (particularly evident on 18/19 July) 425 and the influence of a pollution plume, as already found for O_3 and NO_2 . Maximal σ_s values were 50-60 Mm⁻¹ for the polluted cases. 426

427 As shown in the previous sections, the remote cases were occasionally influenced by anthro-428 pogenic pollution during the afternoon. For further analysis, we classify our data in remote and 429 polluted cases according to the prevailing σ_s , which constitutes a quantity that is relatively inde-430 pendent of the trace gas mixing ratios. Given et al. [2003] found that σ_s values representative for 431 pristine background conditions are typically below 11 Mm⁻¹. In our case, this corresponds to NO_x 432 mixing ratios below ~ 0.6 ppb. Henceforth, we refer to remote cases (mainly clean air masses with very minor influence of anthropogenic pollution) with $\sigma_s < 11 \text{ Mm}^{-1}$ and NO_x < 0.6 ppb and 433 to polluted cases with $\sigma_s > 11 \text{ Mm}^{-1}$ and $NO_x > 0.6 \text{ ppb}$. 434

435 3.3 Impact of the Manaus plume

The impact of the Manaus plume on Amazonian remote background conditions can be stu-died very well on 10, 18 and 19 July. During all three days we measured a near twofold increase

438 of the O_3 mixing ratio in the late afternoon (see section 3.2.1), which was due to the typical 439 change of the wind direction from east-northeast to south-southeast. On 10 July (distance to Ma-440 naus \sim 147 km) after 15:40 LT the O₃ mixing ratio increased sharply from 19 ppb to 32 ppb with-441 in a few minutes (Figure 4c). Besides the potential impact of anthropogenic pollution, a sudden 442 drop in global irradiance indicates an increase in cloud cover associated with a thunderstorm, 443 which occurred near the boat location. According to *Betts et al.* [2002], this O₃ increase may 444 suggest the occurrence of a strong convective downdraft, which brought O₃ rich air from the 445 lower troposphere down to the surface. However, results from the HYSPLIT model indicate that 446 the boat was located north of the Manaus plume in the morning, but that due to a change of the 447 wind direction to south-southeast near the surface the plume swung northwards in the afternoon 448 and reached the boat location at around 16:00 LT (not shown).

449 We demonstrate this pattern in further detail for 18 and 19 July (distance to Manaus 450 \sim 43 km), when comparable increases of the O₃ mixing ratio were observed. Figure 6 shows re-451 sults of air mass dispersion calculations with the HYSPLIT model for 18 July. The main flow 452 was from northeast before midday and the plume did not reach the boat (Figure 6a). Measured 453 trace gas mixing ratios and aerosol concentrations around midday were close to Amazonian 454 background conditions ([NO] ~ 50 ppt (< LOD), [NO₂] = 200 ppt, [O₃] = 16 ppb, $\sigma_s = 6 \text{ Mm}^{-1}$) 455 (Figure 4). The change of the wind direction to south-southeast near the surface in the late after-456 noon (measured on the boat and calculated from HYPSLIT, see Figure 2 and 4) is evident in 457 Figure 6b, causing a northward swing of the plume that now reached the boat. Accordingly, trace 458 gas mixing ratios and aerosol concentrations increased substantially ($[NO] = 300 \text{ ppt}, [NO_2] = 3$ ppb, $[O_3] = 35$ ppb, $\sigma_s = 30$ Mm⁻¹). 459

460 Since no airborne measurements west of Manaus were made on 18 July, we further validate 461 these findings using results from Flight #18: 10:00-14:00 LT and Flight #19: 15:45-18:00 LT on 462 19 July [see *Kuhn et al.*, 2010]. This can be justified by the fact that (i.) the diel cycle of the wind 463 direction is nearly identical for 18 and 19 July (Figure 4a), (ii.) the diel variation of O₃ mixing 464 ratios shows a comparable pattern for 18 and 19 July (Figure 4c) and (iii.) in agreement with our 465 wind measurements at the boat and at the K34 tower site (see above, Figure 4a) the HYSPLIT 466 dispersion calculations reveal a comparable result for 18 and 19 July. After takeoff from Manaus 467 airport, Flights #18 and #19 were set up as a Lagrangian experiment with a series of stacked ho-468 rizontal profiles [Kuhn et al., 2010]. We use results measured at various altitudes for a longitude 469 of $\sim 60.41^{\circ}$ W directly over the boat. On 19 July between 11:08 LT and 11:40 LT, the O₃ mixing 470 ratios were 26 ppb, 28 ppb, 29 ppb, 23 ppb and 33 ppb at heights of 3 m (boat), 30 m, 220 m, 471 544 m and 860 m a.g.l., respectively. These vertical mixing ratio differences suggest that convec-472 tive transport away from the surface enables precursor compounds to achieve their full ozone 473 generating potential [cf. Kuhn et al., 2010].

474 Figure 7 presents crosswind transects in the urban outflow for O₃ and aerosol number (CN) 475 concentrations at 220 m a.g.l.. Just before midday (11:25-11:34 LT, Flight#18) on 19 July the 476 boat was located at the northern edge of the plume (3.0948°S, 60.4102°W) (Figure 7). Compared 477 to 18 July, when no anthropogenic influence was present just before midday, O₃ mixing ratios 478 measured on the boat were slightly elevated (26 ppb) at 11:31 LT (the time the airplane flew 479 over) on 19 July. In the afternoon the O_3 mixing ratio on the boat was 30 ppb when the airplane 480 flew over again at 16:45, while a mixing ratio of 34 ppb was measured at a height of 220 m a.g.l. 481 (Figure 7). The center of the plume before midday ($[O_3] = 44$ ppb) was at around 3.18°S (south 482 of the Rio Negro); while during the afternoon (16:40-16:48 LT, Flight#19) the center ($[O_3] = 50$

483 ppb) was at around 3.0° S (northern bank of the Rio Negro). This is also visible in the corres-484 ponding CN concentrations. Hence, Figure 6 strongly suggests that the plume indeed swung 485 northwards as already indicated by results from the HYSPLIT model due to the southeasterly 486 flow component (Figure 3 and 6). Apparently, the center of the plume swept over the boat plat-487 form around 14:00 LT when O₃ mixing ratios on the boat exceeded 40 ppb. Hence, we sampled 488 air masses that were significantly affected by anthropogenic pollution from Manaus during the 489 afternoon. The arrival of the plume at the boat platform on 18/19 July is also evident from NO₂ 490 mixing ratios and σ_s values that increased substantially during the late afternoon and typically 491 remained high during the whole night. On 20 July, elevated mixing ratios measured on the boat 492 in the proximity of Manaus (~19 km distance) suggest that the air masses sampled were influ-493 enced by the Manaus plume, which is supported by the results from the airborne measurements 494 in the afternoon.

495

496 3.4 Photo-stationary state (PSS)

497 Since data measured during the afternoon often had to be rejected due to cloudy conditions 498 (section 2.5, Figure 4c), only a limited number of data points was available for the PSS analysis. 499 The application of the rejection criteria generally resulted in the rejection of measured NO₂ mix-500 ing ratios lower than NO₂ mixing ratios predicted by the PSS approach ($\Phi < 1$). Such cases 501 might be attributed to local fresh NO emissions [cf. Rohrer et al., 1998], e.g., from the generator, 502 but also to sunrise and sunset periods. Only 42 % of the original daytime data could be used for 503 the calculation of Φ . As shown in Figure 4e, the calculated Φ values (eq. 1) exceed unity in most 504 cases. Just after sunrise (high solar zenith angle) Φ frequently reveals a tendency towards unity, 505 which most likely is caused by the absence of peroxy radicals during this period. This is ex-

506 pected because during these times the emission of hydrocarbons from vegetation [e.g.,

Kesselmeier and Staudt, 1999] (see Figure 5) as well as photochemistry forming OH and peroxy
radicals is still low [see *Griffin et al.*, 2007], and no significant additional oxidation pathway for

509 NO is present. The deviation of Φ from unity sometimes follows the diel course of $j(NO_2)$ (Fig-

510 ure 4f), indicating the relation to photochemically derived radicals, which is consistent with find-

511 ings by, e.g., *Griffin et al.* [2007] and *Rohrer et al.* [1998].

512 We found the largest deviations from PSS ($\Phi > 2.5$) for the clean remote cases ($\sigma_s < 11$

513 Mm⁻¹ and NO_x < 0.6 ppb). This was mainly caused by the coincidence of NO mixing ratios be-

514 low 200 ppt and low O₃ mixing ratios (13-17 ppb). For comparison, above a very remote Taiga

515 woodland, *Bakwin et al.* [1994] found Φ values > 5, with prevailing NO_x levels < 100 ppt and O₃

516 < 30 ppb. *Hauglustaine et al.* [1996] and *Ridley et al.* [1992] investigated PSS at the Mauna Loa

517 Observatory in Hawaii and despite remarkably low NO_x levels (< 100 ppt), Φ values were less

518 than 3. The significantly lower O₃ mixing ratios observed for our remote cases compared to the

519 O₃ mixing ratios measured at Mauna Loa, coinciding with low NO mixing ratios, explain that our

520 Φ values sometimes exceeded a value of 3. *Hosaynali Beygi et al.* [2011] found Φ values >> 10

under pristine background conditions ($NO_x = 5-25$ ppt) in the Southern Atlantic boundary layer

522 at O_3 levels of about 20 ppb.

523 Previously reported near-ground Φ values at locations influenced by urban or suburban air

524 masses were 1 – 3 [e.g., *Carpenter et al.*, 1998; *Griffin et al.*, 2007; *Matsumoto et al.*, 2006;

525 *Parrish et al.*, 1986; *Rohrer et al.*, 1998; *Volz-Thomas et al.*, 2003]. These Φ values compare

526 well with cases that were influenced by anthropogenic pollution in our study.

527 3.4.1 Influence of environmental factors and anthropogenic pollution on PSS

528

529 when the timescales of turbulent transport are comparable or shorter than τ_{PSS} , e.g., when there 530 are nearby soil biogenic NO emissions and/or dry deposition of NO₂ and O₃. Turbulent time-531 scales above a flat surface are often faster than the establishment of PSS (1-10 seconds, [cf. 532 Trebs et al., 2006]), indicating that the photochemical equilibrium ("steady state") may not be 533 reached. Moreover, advection of O₃-depleted and NO_x-enriched air masses from the rain forest, 534 reflecting an efficient dry deposition and reaction of O₃ with biogenic soil NO emissions in the 535 trunk space, may have a significant influence on Φ values. However, these influences can be 536 largely neglected if the footprint of the measurements consists of a fetch of open water, lacking 537 substantial sources or sinks (deposition) of NO_x and O_3 (see section 3.2.1). Maximal Φ values were calculated for the clean remote cases ($\sigma_s < 11 \text{ Mm}^{-1}$, NO_x < 0.6 ppb, 538 539 mostly on 10, 18/19, 24/25 July), comprising about 45 % of all calculated Φ values. They ranged 540 from 2.5 to 16, and the upper values were often derived from either NO or NO₂ mixing ratios 541 below the LOD (white circles, Figure 4e). About 55% of the Φ values are assigned to polluted 542 cases and were mostly below 2.

Generally, the photochemical equilibrium is not reached and Φ values deviate from unity

Periods in the afternoon when a change of the wind direction occurred (see section 3.1) often coincided with periods of increased cloudiness due to convection (Figure 4a,c). Since the rejection criteria had to be applied during these periods (see above), an analysis of the influence of the Manaus plume arrival during 18/19 July (see section 3.3) on the diel evolution of Φ values was not possible. These periods also do not meet the "steady state" criterion due to strong mixing ratio fluctuations (see e.g., 18 July, Figure 4c). 549 However, we found that the Φ values were dependent on the position of the boat relative to 550 the Manaus plume, and consequently also on the distance of the measurement location to Ma-551 naus. From 18 to 20 July we observed a gradual decline of calculated midday Φ values (Figure 552 4e). As previously shown, on 18 July no anthropogenic influence was present during midday, 553 while on 19 July the boat was located at the edge of the plume and on 20 July the boat moved 554 closer to Manaus and the sampled air masses were strongly affected by the plume. The Φ values 555 were closest to unity when polluted air was sampled, particularly on 20 and 26/27 July, when the 556 distance of the boat was only 19 km from Manaus.

557

3.4.2 Discussion of uncertainties and dependence on NO_x

558 Figure 8 combines the results from all locations. The 2σ uncertainty of Φ (black line, see 559 section 2.6) increases with decreasing NO_x. All Φ values above the 2 σ uncertainty present signif-560 icant deviations from PSS. The values below the black line may be the result of measurement 561 imprecision or represent true deviations from PSS. More than half of the estimated Φ values de-562 viate significantly from PSS. This deviation is most pronounced at NO_x mixing ratios below 1 563 ppb, whereas Φ approaches unity with increasing NO_x mixing ratios, which is the behavior also 564 found in previous studies [e.g., Griffin et al., 2007; Volz-Thomas et al., 2003]. This trend is cor-565 roborated by the increase of Φ during midday (higher $i(NO_2)$ values), which is related to radical 566 formation, as previously discussed. The Φ values close to unity at low NO_x mixing ratios are 567 mainly confined to periods with lower $i(NO_2)$ and low VOC mixing ratios (see Figure 5) just af-568 ter sunrise and before sunset. On the other hand, higher NO_x levels favor the rapid establishment 569 of the PSS, as the sink for XO₂ increases when higher NO mixing ratios are present [cf. Volz-570 *Thomas et al.*, 2003]. Furthermore, under high-NO_x conditions the reaction of NO₂ with OH,

571 forming HNO₃, leads to a competition with the oxidation of CO and hydrocarbons by OH, limit-572 ing the production of significant levels of peroxy radicals [e.g., Mannschreck et al., 2004; Volz-573 Thomas et al., 2003]. In our study, the ratio Φ reaches unity within its uncertainty range at NO_x 574 mixing ratios of about 3 ppb. This value is comparable to studies at other locations and implies 575 that no significant net production of ozone takes place at NO_x mixing ratios >> 3 ppb (corres-576 ponding to NO mixing ratios >> 1.4 ppb). For example, Griffin et al. [2007] also found $\Phi \sim 1$ at NO_x mixing ratios > 3 ppb and *Volz-Thomas et al.* [2003] found $\Phi \sim 1$ at NO mixing ratios > 1 577 578 ppb.

579 3.5 XO₂ mixing ratios

580 The formation and cycling of radicals in tropical environments like the Amazon Basin is 581 still subject of ongoing discussion. The high isoprene mixing ratios over tropical forest clearly 582 have a considerably smaller effect on OH (depletion) than previously thought. Recent estimates 583 and measurements within the CBL indicate that prevailing OH concentrations are an order of magnitude higher (about 6×10^6 molecules cm⁻³ or 0.25 ppt, respectively) than currently pre-584 585 dicted by state-of-the-art atmospheric chemistry models [see Kuhn et al., 2007; Ganzeveld et al., 586 2008; Lelieveld et al., 2008; Martinez et al., 2010; Kubistin et al., 2010]. The presence of OH 587 radicals controls the formation of XO₂ in the atmosphere through the oxidation of CO and hy-588 drocarbons.

589 3.5.1 Estimation of XO₂ from PSS

In this study, we provide estimates of XO₂ in the region west of Manaus according to eq. (3) (see Figure 4f). It should be noted that some XO₂ mixing ratios were derived from either NO or NO₂ mixing ratios below the LOD (white circles, Figure 4f). Median midday/afternoon (PSS de-

rived) XO₂ mixing ratios (eq. 3) for the remote cases ($\sigma_s < 11 \text{ Mm}^{-1}$ and NO_x < 0.6 ppb) range 593 from 90 to 120 ppt, while for the polluted cases ($\sigma_s > 11 \text{ Mm}^{-1}$ and NO_x > 0.6 ppb) our estimates 594 595 range from 15 to 60 ppt. The uncertainty of the peroxy radical mixing ratios is a function of 596 Φ (see eq. 3) and significantly exceeds 100 %, which is mainly caused by the imprecision of the 597 NO_x measurements at low mixing ratios (see section 2.6). This constitutes a major disadvantage 598 of the PSS method to estimate XO₂. Furthermore, PSS is known to generally overestimate XO₂ 599 mixing ratios by a factor of two or higher, which is most likely attributable to yet unidentified or 600 unaccounted-for NO₂ formation pathways [Cantrell et al., 1997; Volz-Thomas et al., 2003; Ho-601 saynali Beygi et al., 2011].

3.5.2 Estimation of XO_2 using a chemistry box model

603 We have additionally used the atmospheric chemistry box model CAABA/MECCA-3.0 604 (Chemistry As A Boxmodel Application-Module Efficiently Calculating the Chemistry of the 605 Atmosphere) [Sander et al., 2011], which contains the Mainz Isoprene Mechanism 2 (MIM2) 606 [Taraborrelli et al., 2009] to predict XO₂ mixing ratios. The model runs were constrained by 607 fixing the measured species (NO, NO₂, O₃, H₂O, C₅H₈, MVK, MACR) to their observed values 608 (scenario 1). Peeters and Müller [2010] and Peeters et al. [2009] proposed that unimolecular 609 reactions of certain RO₂ radicals lead to additional OH formation. Since MECCA-3.0 does not 610 include these new reactions yet, it currently significantly underpredicts OH and consequently 611 also HO₂ within the CBL in the Amazon [see Kubistin et al., 2010]. As a consequence, we 612 performed additional model runs based on the first scenario but also fixing OH [Kuhn et al., 613 2007; Martinez et al., 2010], HCHO [Kesselmeier et al., 2002] and H₂O₂ [Ouesada et al. 2001] 614 to measured values (scenario 2). For all cases the model was run until a steady state value of HO₂ 615 was reached. The results presented in Table 1 indicate that the predicted XO₂ mixing ratios are

616 mostly within the range of those estimated by the PSS method. For scenario 2, the XO₂ predicted

by the model would be two times lower if the OH mixing ratio fixed in the model input is

618 reduced by a factor of two. The modeled XO₂ is sometimes higher than the PSS estimates in the

afternoon, which might be caused by the missing RO_2 sink attributed to the not yet included OH

620 formation mechanism proposed by *Peeters and Müller* [2010] and *Peeters et al.* [2009] in

621 MIM2. In the early afternoon of 26 July (polluted case), the XO₂ mixing ratios estimated from

622 PSS significantly exceed the model results.

623 3.5.3 Comparison to previous results from tropical environments

624 Mixing ratios of HO₂ only (providing a lower limit on the XO₂ mixing ratios), measured in 625 the CBL over pristine rainforest in Suriname during the GABRIEL campaign, showed maxima 626 of about 70 ppt around noontime [Martinez et al., 2010]. Within the surface layer west of Ma-627 naus the HO₂ mixing ratio calculated with a single column chemistry-climate model (SCM) 628 [Ganzeveld et al., 2002], also being applied in another study of the Manaus city plume [Kuhn et 629 al_{2} , 2010], reached 50 ppt, with the CH₃O₂ mixing ratio as small as 3 ppt during midday. The 630 XO₂ mixing ratios measured in a comparably remote tropical environment in the rainforest of 631 Borneo ranged from 20 to 40 ppt around midday [Pugh et al., 2010].

632

633 4. Conclusions

We have measured mixing ratios of the trace gases NO, NO₂,O₃, and VOCs as well as the aerosol scattering coefficient, σ_s , in a previously unexplored environment using a boat platform cruising on rivers west of Manaus (Amazon Basin) on 10-11, 18-20, 25-27 July 2001. To our knowledge, these are the first trace gas measurements made on rivers in the Amazon Basin, which are a unique location for studying atmospheric processes within the tropical boundary

639 layer. The temporal and spatial dispersion of the Manaus plume had a significant influence on 640 the measured mixing ratios at the different boat locations. The measured trace gas mixing ratios 641 and calculated Φ values reflect the interaction between river breezes and/or mesoscale circula-642 tions, and the impact of synoptic winds transporting pollution from Manaus to pristine areas of 643 the Amazonian rain forest. In particular, during our measurement period the wind mainly blew 644 from east-northeast during the night and in the morning (influence of easterly trade winds), while 645 wind directions from south-southeast dominated during the afternoon and evening. The remote 646 cases with very minor anthropogenic influence were characterized by σ_s values below 11 Mm⁻¹ 647 and NO_x below 0.6 ppb, coinciding with midday O₃ mixing ratios often lower than 20 ppb. For 648 these cases, significant deviations from the photo-stationary state in the NO-NO₂-O₃ triad were 649 observed. The combination of low NO (< 200 ppt) with low O_3 (< 20 ppb) in the absence of 650 anthropogenic pollution caused the Leighton ratio, Φ , sometimes to be higher than 3. The change 651 of the wind direction to south-southeast in the afternoon facilitated a northward swing of the 652 Manaus plume, which then often reached the boat. These polluted air masses were characterized by σ_s values of about 30 Mm⁻¹, NO_x mixing ratios between 2 and 4 ppb and by O₃ mixing ratios 653 654 of up to 35 ppb. For these polluted cases the calculated Φ values were close to unity. Results 655 from the PSS method and the box model CAABA/MECCA-3.0 suggest that the XO₂ mixing ra-656 tios are at least a factor of two higher in remote areas west of Manaus compared to locations in-657 fluenced by the Manaus urban plume. This implies that anthropogenic pollution from Manaus 658 significantly alters the chemical composition of the otherwise pristine Amazonian background 659 air near the surface. This may in turn have a significant influence on surface-atmosphere ex-660 change processes of reactive species. The limited knowledge of HO₂ and RO₂ mixing ratios in 661 tropical regions like the Amazon Basin emphasizes the need for in-situ measurements in the near

future to improve our understanding of O_3 production pathways and the cycling of reactive nitrogen as well as of OH and XO_2 in clean and polluted air masses.

We anticipate that due to the rapidly growing population of Manaus to meanwhile more than 1.7 million human inhabitants, and consequently rising pollution levels, the impact on atmospheric chemistry in the region west of Manaus is growing. It is recommended to establish longterm monitoring stations downwind of Manaus and to apply mesoscale models to investigate the influence of the Rio Negro river breeze on the dispersion of the Manaus urban plume.

669

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- 686
- 687 **References**
- Andreae, M. O., E. V. Browell, M. Garstang, G. L. Gregory, R. C. Harriss, G. F. Hill, D. J.
 Jacob, M. C. Pereira, G. W. Sachse, A. W. Setzer, P. L. S. Dias, R. W. Talbot, A. L. Torres,
 and S. C. Wofsy (1988), Biomass burning emissions and associated haze layers over
 Amazonia, *Journal of Geophysical Research-Atmospheres*, *93*(D2), 1509-1527.
- Atkinson, R., D. L. Baulch, R. A. Cox, J. N. Crowley, R. F. Hampson, R. G. Hynes, M. E.
 Jenkin, M. J. Rossi, and J. Troe (2004), Evaluated kinetic and photochemical data for
- 694 atmospheric chemistry: Volume I gas phase reactions of O_x , HO_x , NO_x and SO_x species, 695 *Atmospheric Chemistry and Physics*, 4, 1461-1738.
- Atkinson, R., D. L. Baulch, R. A. Cox, J. N. Crowley, R. F. Hampson, R. G. Hynes, M. E.
 Jenkin, M. J. Rossi, and J. Troe (2006), Evaluated kinetic and photochemical data for
 atmospheric chemistry: Volume II gas phase reactions of organic species, *Atmospheric Chemistry and Physics*, 6, 3625-4055.
- Bakwin, P. S., S. C. Wofsy, and S. M. Fan (1990), Measurements of Reactive Nitrogen-Oxides
 (NOy) within and above a Tropical Forest Canopy in the Wet Season, *Journal of Geophysical Research-Atmospheres*, 95(D10), 16765-16772.
- Bakwin, P. S., D. J. Jacob, S. C. Wofsy, J. W. Munger, B. C. Daube, J. D. Bradshaw, S. T.
 Sandholm, R. W. Talbot, H. B. Singh, G. L. Gregory, and D. R. Blake (1994), Reactive
 Nitrogen-Oxides and Ozone above a Taiga Woodland, *Journal of Geophysical Research- Atmospheres*, 99(D1), 1927-1936.
- Betts, A. K., L. V. Gatti, A. M. Cordova, M. Dias, and J. D. Fuentes (2002), Transport of ozone
 to the surface by convective downdrafts at night, *Journal of Geophysical Research- Atmospheres*, 107(D20).
- Betts, A. K., G. Fisch, C. von Randow, M. A. F. Silva Dias, J. C. P. Cohen, R. da Silva, and D.
 R. Fitzjarrald (2008), The Amazonian Boundary Layer and Mesoscale Circulations, *Amazonia and Global Change, Geophysical Monograph Series*, *186*, 163-181.
- Borbon, A., H. Fontaine, M. Veillerot, N. Locoge, J. C. Galloo, and R. Guillermo (2001), An
 investigation into the traffic-related fraction of isoprene at an urban location, *Atmospheric Environment*, 35(22), 3749-3760.
- Cantrell, C. A., R. E. Shetter, J. G. Calvert, F. L. Eisele, E. Williams, K. Baumann, W. H. Brune,
 P. S. Stevens, and J. H. Mather (1997), Peroxy radicals from photostationary state deviations
- and steady state calculations during the Tropospheric OH Photochemistry Experiment at
 Idaho Hill, Colorado, 1993, *Journal of Geophysical Research-Atmospheres*, *102*(D5), 6369 6378.
- 721 Carpenter, L. J., K. C. Clemitshaw, R. A. Burgess, S. A. Penkett, J. N. Cape, and G. C.
- 722 McFadyen (1998), Investigation and evaluation of the NO_x/O_3 photochemical steady state, 723 *Atmospheric Environment*, 32(19), 3353-3365.
- Crutzen, P. J., and J. Lelieveld (2001), Human impacts on atmospheric chemisty, *Annual Review of Earth and Planetary Sciences*, *29*, 17-45.

- Crutzen, P. J., A. C. Delany, J. Greenberg, P. Haagenson, L. Heidt, R. Lueb, W. Pollock, W.
 Seiler, A. Wartburg, and P. Zimmerman (1985), Tropospheric chemical-composition
 measurements in Brazil during the dry season, *Journal of Atmospheric Chemistry*, 2(3), 233256.
- de Oliveira, A. P., and D. R. Fitzjarrald (1993), The Amazon River Breeze and the Local
 Boundary-Layer: 1. Observations, *Boundary-Layer Meteorology*, 63(1-2), 141-162.
- Draxler, R. R., and G. D. Hess (1998), An overview of the HYSPLIT 4 modelling system for
 trajectories, dispersion and deposition, *Australian Meteorological Magazine*, 47(4), 295-
- 734 308.
- Duane, M., B. Poma, D. Rembges, C. Astorga, and B. R. Larsen (2002), Isoprene and its
 degradation products as strong ozone precursors in Insubria, Northern Italy, *Atmospheric Environment*, 36(24), 3867-3879.
- Fisch, G., J. Tota, L. A. T. Machado, M. Dias, R. F. D. Lyra, C. A. Nobre, A. J. Dolman, and J.
 H. C. Gash (2004), The convective boundary layer over pasture and forest in Amazonia, *Theoretical and Applied Climatology*, 78(1-3), 47-59.
- Ganzeveld, L., D. Helmig, C. W. Fairall, J. Hare, and A. Pozzer (2009), Atmosphere-ocean
 ozone exchange: A global modeling study of biogeochemical, atmospheric, and waterside
 turbulence dependencies, *Global Biogeochemical Cycles*, 23.
- Ganzeveld, L., L. Bouwman, E. Stehfest, D. P. van Vuuren, B. Eickhout, and J. Lelieveld (2010),
 Impact of future land use and land cover changes on atmospheric chemistry-climate
 interactions, *Journal of Geophysical Research-Atmospheres*, *115*.
- Ganzeveld, L., G. Eerdekens, G. Feig, H. Fischer, H. Harder, R. Konigstedt, D. Kubistin, M.
 Martinez, F. X. Meixner, H. A. Scheeren, V. Sinha, D. Taraborrelli, J. Williams, J. V. G. de
 Arellano, and J. Lelieveld (2008), Surface and boundary layer exchanges of volatile organic
 compounds, nitrogen oxides and ozone during the GABRIEL campaign, *Atmospheric Chemistry and Physics*, 8(20), 6223-6243.
- Ganzeveld, L. N., J. Lelieveld, F. J. Dentener, A. C. Krol, and G. Roelofs (2002), Atmospherebiosphere trace gas exchanges simulated with a single-column model, *Journal of Geophysical Research*, *107*(D16), ACH8-1-21.
- Griffin, R. J., P. J. Beckman, R. W. Talbot, B. C. Sive, and R. K. Varner (2007), Deviations from
 ozone photostationary state during the the International Consortium for Atmospheric
 Research on Transport and Transformation 2004 campaign: use of measurements and
 photochemical modeling to assess potential causes, *Journal of Geophysical Research-Part D-Atmospheres*, 1-16.
- Gut, A., M. Scheibe, S. Rottenberger, U. Rummel, M. Welling, C. Ammann, G. A. Kirkman, U.
 Kuhn, F. X. Meixner, J. Kesselmeier, B. E. Lehmann, W. Schmidt, E. Muller, and M. T. F.
 Piedade (2002), Exchange fluxes of NO₂ and O₃ at soil and leaf surfaces in an Amazonian
 rain forest, *Journal of Geophysical Research-Atmospheres*, *107*(D20).
- Guyon, P., B. Graham, J. Beck, O. Boucher, E. Gerasopoulos, O. L. Mayol-Bracero, G. C.
 Roberts, P. Artaxo, and M. O. Andreae (2003), Physical properties and concentration of
 aerosol particles over the Amazon tropical forest during background and biomass burning
 conditions, *Atmospheric Chemistry and Physics*, *3*, 951-967.
- 768 Harriss, R. C., S. C. Wofsy, M. Garstang, E. V. Browell, L. C. B. Molion, R. J. McNeal, J. M.
- Hoell, R. J. Bendura, S. M. Beck, R. L. Navarro, J. T. Riley, and R. L. Snell (1988), The
- Amazon boundary layer experiment (ABLE-2A) Dry season 1985, *Journal of Geophysical Research-Atmospheres*, 93(D2), 1351-1360.

- Hauglustaine, D. A., S. Madronich, B. A. Ridley, J. G. Walega, C. A. Cantrell, R. E. Shetter, and
 G. Hubler (1996), Observed and model-calculated photostationary state at Mauna Loa
 observatory during MLOPEX 2, *Journal of Geophysical Research-Atmospheres*, *101*(D9),
 14681-14696.
- Hosaynali Beygi, Z., H. Fischer, H.D. Harder, M. Martinez, R. Sander, J. Williams, D.M.
 Brookes, P.S. Monks, and J. Lelieveld (2011), Oxidation photochemistry in the Southern Atlantic boundary layer: Unexpected deviations of photochemical steady state, *Atmospheric Chemistry and Physics*, 11, 8497-8513.
- Jacob, D. J., and S. C. Wofsy (1990), Budgets of reactive nitrogen, hydrocarbons, and ozone
 over the Amazon forest during the wet season, *Journal of Geophysical Research- Atmospheres*, 95(D10), 16737-16754.
- Kesselmeier, J., U. Kuhn, S. Rottenberger, T. Biesenthal, A. Wolf, G. Schebeske, M. O.
 Andreae, P. Ciccioli, E. Brancaleoni, M. Frattoni, S. T. Oliva, M. L. Botelho, C. M. A.
 Silva, and T. M. Tavares (2002), Concentrations and species composition of atmospheric
 volatile organic compounds (VOCs) as observed during the wet and dry season in Rondonia

787 (Amazonia), Journal of Geophysical Research-Atmospheres, 107(D20).

- Kesselmeier, J., and M. Staudt (1999), Biogenic volatile organic compounds (VOC): An
 overview on emission, physiology and ecology, *Journal of Atmospheric Chemistry*, 33(1),
 23-88.
- Kley, D., and M. McFarland (1980), Chemiluminescence Detector for NO and NO₂, *Atmospheric Technology*, *12*, 62-69. Kubistin, D., H. Harder, M. Martinez, M. Rudolf, R. Sander, H. Bozem, G. Eerdekens, H. Fischer, C. Gurk, T. Klüpfel, R. Königstedt, U. Parchatka, C. L.
 Schiller, A. Stickler, D. Taraborrelli, J. Williams, and J. Lelieveld (2010), Hydroxyl radicals
 in the tropical troposphere over the Suriname rainforest: comparison of measurements with
 the box model MECCA, *Atmospheric Chemistry and Physics*, *10*(19), 9705-9728.
- Kuhn, U., L. Ganzeveld, A. Thielmann, T. Dindorf, G. Schebeske, M. Welling, J. Sciare, G.
 Roberts, F. X. Meixner, J. Kesselmeier, J. Lelieveld, O. Kolle, P. Ciccioli, J. Lloyd, J.
 Trentmann, P. Artaxo, and M. O. Andreae (2010), Impact of Manaus City on the Amazon
 Green Ocean atmosphere: ozone production, precursor sensitivity and aerosol load,

801 *Atmospheric Chemistry and Physics*, 10(19), 9251-9282.

- Kuhn, U., M. O. Andreae, C. Ammann, A. C. Araùjo, E. Brancaleoni, P. Ciccioli, T. Dindorf, M.
 Frattoni, L. V. Gatti, L. Ganzeveld, B. Kruijt, J. Lelieveld, J. Lloyd, F. X. Meixner, A. D.
 Nobre, U. Pöschl, U. Spirig, P. Stefani, A. Thielmann, R. Valentini, and J. Kesselmeier
- (2007), Isoprene and monoterpene fluxes from Central Amazonian rainforest inferred from
 tower-based and airborne measurements, and implications on the atmospheric chemistry and
 the local carbon budget, *Atmospheric Chemistry and Physics*, 7, 2855–2879.
- Kuhn, U., S. Rottenberger, T. Biesenthal, A. Wolf, G. Schebeske, P. Ciccioli, and J. Kesselmeier
- 809 (2004), Strong correlation between isoprene emission and gross photosynthetic capacity
- during leaf phenology of the tropical tree species Hymenaea courbaril with fundamental
 changes in volatile organic compounds emission composition during early leaf development, *Plant Cell and Environment*, 27(12), 1469-1485.
- 813 Kuhn, U., S. Rottenberger, T. Biesenthal, A. Wolf, G. Schebeske, P. Ciccioli, E. Brancaleoni, M.
- 814 Frattoni, T. M. Tavares, and J. Kesselmeier (2002), Isoprene and monoterpene emissions of
- 815 Amazonian tree species during the wet season: Direct and indirect investigations on
- 816 controlling environmental functions, *Journal of Geophysical Research-Atmospheres*,
- 817 *107*(D20).

- 818 Leighton, P. A. (1961), *Photochemistry of Air Pollution*, Academic Press, New York and
- 819 London.
- Lelieveld, J., T. M. Butler, J. N. Crowley, T. J. Dillon, H. Fischer, L. Ganzeveld, H. Harder, M.
 G. Lawrence, M. Martinez, D. Taraborrelli, and J. Williams (2008), Atmospheric oxidation
 capacity sustained by a tropical forest, *Nature*, 452(7188), 737-740.
- Mannschreck, K., S. Gilge, C. Plass-Duelmer, W. Fricke, and H. Berresheim (2004), Assessment
 of the applicability of NO-NO₂-O₃ photostationary state to long-term measurements at the
 Hohenpeissenberg GAW Station, Germany, *Atmospheric Chemistry and Physics*, *4*, 12651277.
- Martin, C. L., D. Fitzjarrald, M. Garstang, A. P. Oliveira, S. Greco, and E. Browell (1988),
 Structure and growth of the mixing layer over the Amazonian rain forest, *Journal of Geophysical Research-Atmospheres*, 93(D2), 1361-1375.
- Martinez, M., H. Harder, D. Kubistin, M. Rudolf, H. Bozem, G. Eerdekens, H. Fischer, T.
 Klupfel, C. Gurk, R. Konigstedt, U. Parchatka, C. L. Schiller, A. Stickler, J. Williams, and J.
 Lelieveld (2010), Hydroxyl radicals in the tropical troposphere over the Suriname rainforest:
- airborne measurements, *Atmospheric Chemistry and Physics*, *10*(8), 3759-3773.
 Matsumoto, J., N. Kosugi, A. Nishiyama, R. Isozaki, Y. Sadanaga, S. Kato, H. Bandow, and Y.
- Kajii (2006), Examination on photostationary state of NOx in the urban atmosphere in
 Japan, *Atmospheric Environment*, 40(18), 3230-3239.
- Parrish, D. D., M. Trainer, E. J. Williams, D. W. Fahey, G. Hubler, C. S. Eubank, S. C. Liu, P. C.
 Murphy, D. L. Albritton, and F. C. Fehsenfeld (1986), Measurements of the NO_x-O₃
 Photostationary State at Niwot Ridge, Colorado, *Journal of Geophysical Research- Atmospheres*, *91*(D5), 5361-5370.
- Peeters, J., and J.-F. Muller (2010), HO(x) radical regeneration in isoprene oxidation via peroxy
 radical isomerisations. II: experimental evidence and global impact, *Physical Chemistry Chemical Physics*, 12(42), 14227-14235.
- Peeters, J., T. L. Nguyen, and L. Vereecken (2009), HO_x radical regeneration in the oxidation of
 isoprene, *Physical Chemistry Chemical Physics*, 11(28).
- Pugh, T. A. M., A. R. MacKenzie, C. N. Hewitt, B. Langford, P. M. Edwards, K. L. Furneaux,
 D. E. Heard, J. R. Hopkins, C. E. Jones, A. Karunaharan, J. Lee, G. Mills, P. Misztal, S.
 Moller, P. S. Monks, and L. K. Whalley (2010), Simulating atmospheric composition over a
 South-East Asian tropical rainforest: performance of a chemistry box model, *Atmospheric Chemistry and Physics*, 10(1), 279-298.
- Quesada, J., D. Grossmann, E. Fernandez, J. Romero, E. Sanhueza, G. Moortgat, and P. J.
 Crutzen (2001), Ground based gas phase measurements in Surinam during the LBACLAIRE 98 experiment, *Journal of Atmospheric Chemistry*, *39*(1), 15-36.
- Reimann, S., P. Calanca, and P. Hofer (2000), The anthropogenic contribution to isoprene
 concentrations in a rural atmosphere, *Atmospheric Environment*, *34*(1), 109-115.
- Ridley, B. A., S. Madronich, R. B. Chatfield, J. G. Walega, R. E. Shetter, M. A. Carroll, and D.
 D. Montzka (1992), Measurements and model simulations of the photo stationary state
 during the Mauna-Loa-Observatory Photochemistry Experiment Implications for radical
 concentrations and ozone production and loss rates, *Journal of Geophysical Research- Atmospheres*, 97(D10), 10375-10388.
- 861 Rohrer, F., D. Bruning, E. S. Grobler, M. Weber, D. H. Ehhalt, R. Neubert, W. Schussler, and I.
- 862 Levin (1998), Mixing ratios and photostationary state of NO and NO₂ observed during the

- POPCORN field campaign at a rural site in Germany, *Journal of Atmospheric Chemistry*,
 31(1-2), 119-137.
- Rummel, U. (2005), Turbulent exchange of ozone and nitrogen oxides between an Amazonian
 rain forest and the atmosphere, University of Bayreuth, Bayreuth.
- Rummel, U., C. Ammann, A. Gut, F. X. Meixner, and M. O. Andreae (2002), Eddy covariance
 measurements of nitric oxide flux within an Amazonian rain forest, *Journal of Geophysical Research-Atmospheres*, *107*, D20, 8050, doi:8010.1029/2001JD000520.
- Rummel, U., C. Ammann, G. A. Kirkman, M. A. L. Moura, T. Foken, M. O. Andreae, and F. X.
 Meixner (2007), Seasonal variation of ozone deposition to a tropical rain forest in southwest
 Amazonia, *Atmospheric Chemistry and Physics*, 7, 5415-5435.
- Ryerson, T. B., E. J. Williams, and F. C. Fehsenfeld (2000), An efficient photolysis system for
 fast-response NO₂ measurements, *Journal of Geophysical Research-Atmospheres*, *105*(D21), 26447-26461.
- Sander, R., A. Baumgaertner, S. Gromov, H. Harder, P. Jöckel, A. Kerkweg, D. Kubistin, E.
 Regelin, H. Riede, A. Sandu, D. Taraborrelli, H. Tost, and Z. Q. Xie (2011), The
 atmospheric chemistry box model CAABA/MECCA-3.0, *Geoscientific Model Development*,
 4(2), 373-380.
- Taraborrelli, D., M. G. Lawrence, T. M. Butler, R. Sander, and J. Lelieveld (2009), Mainz
 Isoprene Mechanism 2 (MIM2): an isoprene oxidation mechanism for regional and global
 atmospheric modelling, *Atmospheric Chemistry and Physics*, 9(8), 2751-2777.
- Torres, A. L., and H. Buchan (1988), Tropospheric nitric oxide mesurements over the Amazon
 Basin, *Journal of Geophysical Research-Atmospheres*, 93(D2), 1396-1406.
- Trebs, I., F. X. Meixner, J. Slanina, R. P. Oties, P. Jongejan, and M. O. Andreae (2004), Realtime measurements of ammonia, acidic trace gases and water-soluble inorganic aerosol
 species at a rural site in the Amazon Basin, *Atmospheric Chemistry and Physics*, *4*, 967-987.
- 888 Trebs, I., L. S. Lara, L. M. Zeri, L. V. Gatti, P. Artaxo, R. Dlugi, J. Slanina, M. O. Andreae, and
- F. X. Meixner (2006), Dry and wet deposition of atmospheric inorganic nitrogen in a
 tropical environment (Rondônia, Brazil), *Atmos. Chem. Phys.*, *6*, 447–469.
- Trebs, I., B. Bohn, C. Ammann, U. Rummel, M. Blumthaler, R. Königstedt, F. X. Meixner, S.
 Fan, and M. O. Andreae (2009), Relationship between the NO₂ photolysis frequency and the
 solar global irradiance, *Atmospheric Measurement Techniques*, *2*, 725–739.
- Volz-Thomas, A., H. W. Patz, N. Houben, S. Konrad, D. Mihelcic, T. Klupfel, and D. Perner
 (2003), Inorganic trace gases and peroxy radicals during BERLIOZ at Pabstthum: An
 investigation of the photostationary state of NO_x and O₃, *Journal of Geophysical Research- Atmospheres*, 108(D4).
- 898 Warneck, P. (2000), *Chemistry of the Natural Atmosphere* 2. ed., Academic Press Inc.
- 899 Wesely, M. L., D. R. Cook, and R. M. Williams (1981), Field measurement of small ozone
- 900 fluxes to snow, wet bare soil, and lake water, *Boundary-Layer Meteorology*, 20(4), 459-471.
- 901 Yang, J., R. E. Honrath, M. C. Peterson, D. D. Parrish, and M. Warshawsky (2004),
- Photostationary state deviation-estimated peroxy radicals and their implications for HOx and
 ozone photochemistry at a remote northern Atlantic coastal site, *Journal of Geophysical Research-Atmospheres*, *109*, D02312, doi:02310.01029/02003JD003983.
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908 Figure Captions

909 Figure 1. GIS map showing the study area and boat positions on 10-11, 18-20, 25-27 July (LBA-

910 CLAIRE-2001 boat experiment). The position of the boat on the different days is represented by

911 colored circles.

912 Figure 2. Wind roses for (a) 00:00 to 14:50 LT and (b) 15:00 to 23:50 LT including 10-min av-

913 eraged wind data from all locations shown in Figure 1 except 26/27 July (for explanation see

914 text) during the LBA-CLAIRE-2001 boat experiment. The time periods chosen for the wind ros-

915 es are in accordance with *de Oliveira and Fitzjarrald* [1993].

916 Figure 3. Wind vectors (expressed as one-hour forward trajectories) calculated for five different

917 heights within the CBL using the HYSPLIT model based on the meteorological REANALYSIS

dataset. The start time at the boat position on 18 July (3.0948°S, 60.4102°W) was 18:00 LT.

919 Figure 4. Measured and calculated time series (10-min averages) of (a) wind speed (ff) and 920 wind direction (DIR, for 18/19 July also shown for the K34 tower site about 58 km NNE of the 921 boat), (b) relative humidity (RH) and air temperature (T), (c) global irradiance (G) and O₃ mix-922 ing ratios, (d) NO and NO₂ mixing ratio, (e) aerosol scattering coefficient (σ_s) and the Leighton 923 ratio, (Φ), and (f) *j*(NO₂) and the inferred XO₂ mixing ratio during 10-11, 18-20 and 25-27 July 924 2001 for the LBA-CLAIRE-2001 boat experiment. Error bars for trace gas measurements indi-925 cate standard deviations. Measured data points below the LOD and Φ values and XO₂ mixing 926 ratios calculated from NO or NO₂ mixing ratios below the LOD are represented by light or white 927 circles. Gaps are caused by data rejection due to local contamination by moving boats or the di-928 esel generator and the application of the "steady state" criterion. The color code at the top refers 929 to the locations of the boat platform shown in Figure 1.

Figure 5. Measured isoprene mixing ratios and MVK/MACR ratios on (a) 10 July (remote case)
and (b) 26 July (polluted case) during the LBA-CLAIRE-2001 boat experiment.

932 Figure 6. Simplified GIS map showing the study area during the LBA-CLAIRE-2001 boat expe-933 riment (orange circle shows the boat location on 18/19 July) including the results of a HYSPLIT 934 dispersion model run (0-1000 m level) based on the meteorological REANALYSIS dataset. The 935 start time in Manaus City (3.11°S, 60.03°W) was 17 July, 08:00 LT. The duration of the run was 936 36 hours. Shown are average air mass dispersions for (a) 18 July, 08:00-11:00 LT indicated by 937 the grey contours and (b) for 18 July, 17:00 to 20:00 LT indicated by the red contours. The flight 938 tracks of (a) Flight #18: 10:00-14:00 LT and (b) Flight #19: 15:45-18:00 LT on 19 July are also 939 shown.

Figure 7. Measured O_3 mixing ratios and aerosol number (CN) concentrations (σ_s data were not

always available) from crosswind transects in the urban outflow during Flight #18 and Flight #19

942 (10:00-14:00 LT and 15:45-18:00 LT on 19 July). Transects are shown for a longitude of \sim

943 60.41°W (~ 43 km downwind of Manaus) directly over the boat platform at 220 m (a.g.l.). CN

data were cut off at 10,000 cm⁻³ [cf. *Kuhn et al.*, 2010]. Corresponding O₃ mixing ratios meas-

ured at the boat platform when the airplane flew over (at 11:31 and 16:45 LT) are also shown.

946 Figure 8. Leighton ratio, Φ , versus measured NO_x mixing ratios (geometric averages) from 10-

947 11, 18-20 and 25-27 July 2001 (N = 242) color-coded with $j(NO_2)$ during the LBA-CLAIRE-

948 2001 boat experiment. The black line indicates the calculated 2σ uncertainty of Φ . Values above

949 the black line significantly deviate from PSS. The Φ values calculated from NO or NO₂ mixing

950 ratios below the LOD are indicated by white circles.

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954 Tables

955

956 **Table 1.** Comparison of XO₂ mixing ratios modeled with CAABA/MECCA-3.0 to those derived

957 from PSS (eq. 3) for 10 July (remote case) and 26 July (polluted case). The input values used for

958 the model (NO, NO₂, O₃, H₂O, photolysis frequencies^{*}) were averaged for the respective VOC

sampling period. Mixing ratios of long-lived trace gases were kept constant (10 July: [CO] = 80

960 ppb, [CH₄] = 1820 ppb, 26 July: [CO] = 175 ppb, [CH₄] = 1875 ppb).

961

VOC sample	XO_2 (scenario 1 ^{**})	XO_2 (scenario 2 ^{**})	mean XO ₂ (PSS)
(local time)	(ppt)	(ppt)	(ppt)
10 July 8:41-9:31	30	22	30
10 July 11:31-12:27	83	115	97
10 July 14:21-15:11	115	247	144
26 July 13:25-14:15	20	11	34
26 July 14:25-15:15	48	30	107
26 July 16:25-17:15	38	112	81

962

963 *Photolysis frequencies, such as $j(O(^{1}D))$, were calculated using the radiative transfer model TUV v4.1. The ozone

964 column was determined from GOME satellite data. The calculated photolysis frequencies were corrected for cloud 965 and aerosol effects by scaling to $j(NO_2)$ [cf. *Kubistin et al.*, 2010].

966 ** For explanation of scenarios see text.





Figure 1.

(a)



(b)









Figure 5.

(a)





(b)



Figure 6.

998 (a)







Figure 7.







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Figure 8.



NO_x, ppb



00:00 - 14:50 LT



15:00 - 23:50 LT













HYSPLIT plume dispersion on 18 July 8:00-11:00 LTtrack of Flight#18 on 19 July, 10:00-14:00 LT









